

# **Amine, Amido, and Imido Complexes of Tantalum Supported by a Pyridine-Linked Bis(phenolate) Pincer Ligand. Ta-N $\pi$ Bonding Influences Pincer Ligand Geometry.**

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## **Supporting Information**

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## Crystal Structure Analysis of:

### Complex 1 (IAT03)

#### Contents

Table 1. Crystal data

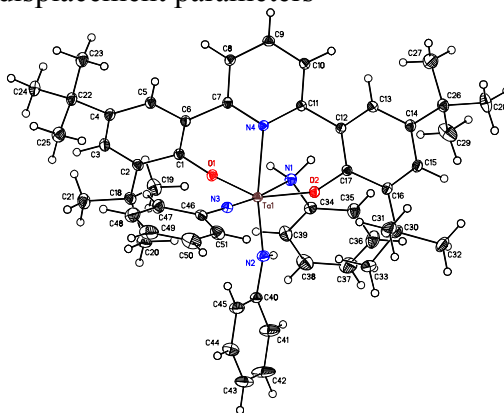
Figures Minimum overlap

Table 2. Atomic Coordinates

Table 3. Selected bond distances and angles

Table 4. Full bond distances and angles

Table 5. Anisotropic displacement parameters



IAT03

**Note:** The crystallographic data have been deposited in the Cambridge Database (CCDC) and has been placed on hold pending further instructions from me. The deposition number is 680573. Ideally the CCDC would like the publication to contain a footnote of the type: "Crystallographic data have been deposited at the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 680573."

**Table 1. Crystal data and structure refinement for IAT03 (CCDC 680573).**

Empirical formula	C <sub>51</sub> H <sub>61</sub> N <sub>4</sub> O <sub>2</sub> Ta • 1½(C <sub>6</sub> H <sub>6</sub> )
Formula weight	1060.15
Crystallization Solvent	Benzene
Crystal Habit	Fragment
Crystal size	0.22 x 0.18 x 0.15 mm <sup>3</sup>
Crystal color	Colorless

### Data Collection

Type of diffractometer	Bruker KAPPA APEX II	
Wavelength	0.71073 Å MoKα	
Data Collection Temperature	100(2) K	
θ range for 9678 reflections used in lattice determination	2.45 to 43.03°	
Unit cell dimensions	a = 25.7250(8) Å b = 17.8546(5) Å c = 23.6141(7) Å	β = 105.9790(10)°
Volume	10427.1(5) Å <sup>3</sup>	
Z	8	
Crystal system	Monoclinic	
Space group	C2/c	
Density (calculated)	1.351 Mg/m <sup>3</sup>	
F(000)	4376	
Data collection program	Bruker APEX2 v2.1-0	
θ range for data collection	2.07 to 43.26°	
Completeness to θ = 43.26°	97.4 %	
Index ranges	-49 ≤ h ≤ 48, -32 ≤ k ≤ 33, -44 ≤ l ≤ 45	
Data collection scan type	ω scans; 16 settings	
Data reduction program	Bruker SAINT-Plus v7.34A	
Reflections collected	233565	
Independent reflections	38134 [R <sub>int</sub> = 0.0821]	
Absorption coefficient	2.154 mm <sup>-1</sup>	
Absorption correction	None	
Max. and min. transmission	0.7382 and 0.6486	

**Table 1 (cont.)****Structure solution and Refinement**

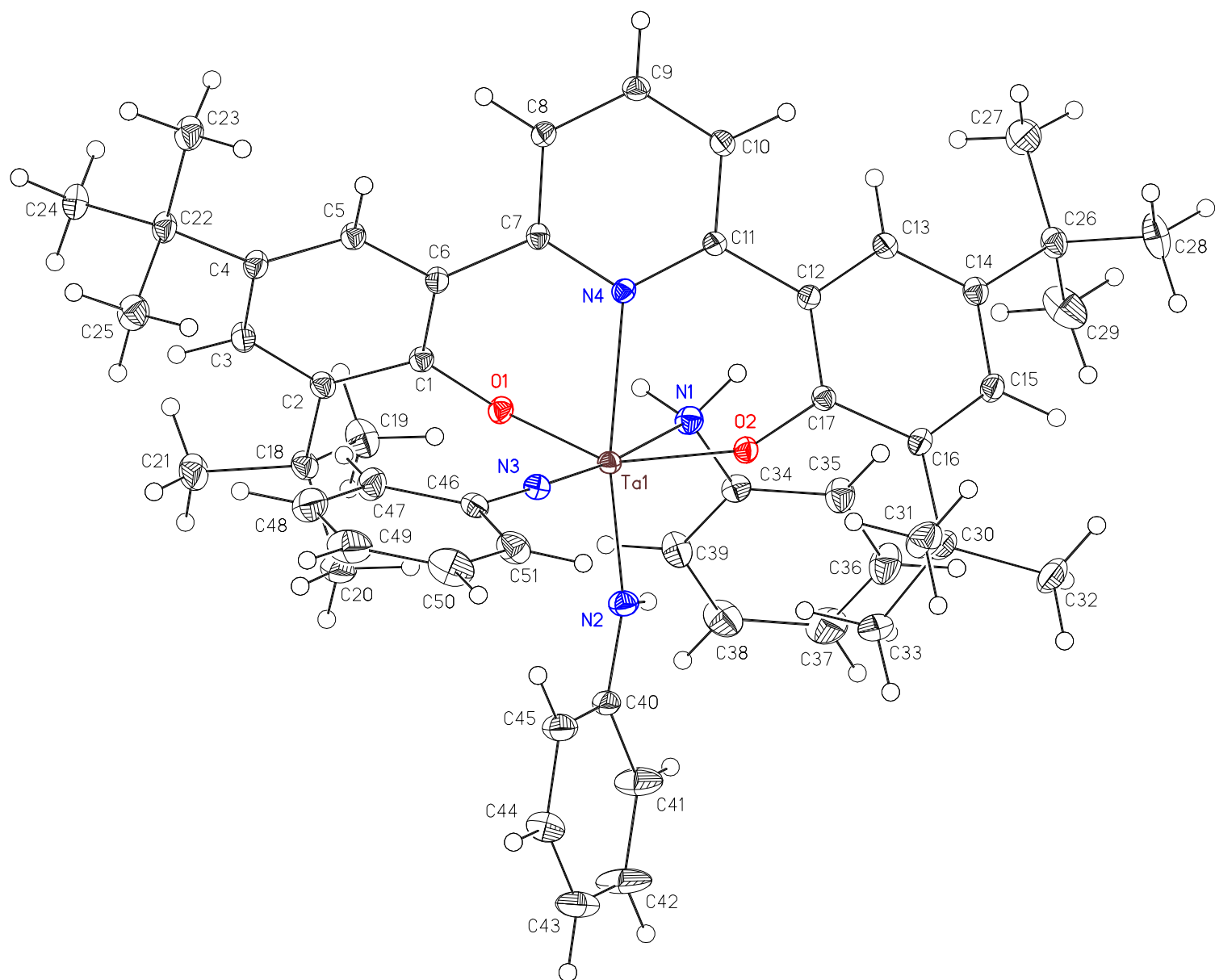
Structure solution program	SHELXS-97 (Sheldrick, 2008)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 2008)
Refinement method	Full matrix least-squares on $F^2$
Data / restraints / parameters	38134 / 0 / 617
Treatment of hydrogen atoms	Riding
Goodness-of-fit on $F^2$	1.117
Final R indices [ $I > 2\sigma(I)$ , 25419 reflections]	$R1 = 0.0299$ , $wR2 = 0.0520$
R indices (all data)	$R1 = 0.0611$ , $wR2 = 0.0557$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.009
Average shift/error	0.000
Largest diff. peak and hole	3.085 and -2.632 e.Å <sup>-3</sup>

**Special Refinement Details**

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

Refinement of  $F^2$  against ALL reflections. The weighted R-factor ( $wR$ ) and goodness of fit ( $S$ ) are based on  $F^2$ , conventional R-factors ( $R$ ) are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.



**Table 2. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for IAT03 (CCDC 680573).  $U(\text{eq})$  is defined as the trace of the orthogonalized  $U^{\text{ij}}$  tensor.**

	x	y	z	$U_{\text{eq}}$
Ta(1)	1305(1)	10037(1)	1393(1)	10(1)
O(1)	1020(1)	10606(1)	1975(1)	13(1)
O(2)	1236(1)	9210(1)	806(1)	12(1)
N(1)	1076(1)	9102(1)	2062(1)	14(1)
N(2)	2063(1)	9811(1)	1897(1)	14(1)
N(3)	1436(1)	10802(1)	963(1)	13(1)
N(4)	377(1)	9973(1)	1003(1)	10(1)
C(1)	747(1)	11249(1)	1834(1)	11(1)
C(2)	886(1)	11880(1)	2210(1)	13(1)
C(3)	628(1)	12553(1)	2013(1)	14(1)
C(4)	236(1)	12634(1)	1467(1)	14(1)
C(5)	83(1)	11995(1)	1132(1)	13(1)
C(6)	320(1)	11294(1)	1312(1)	12(1)
C(7)	83(1)	10615(1)	982(1)	11(1)
C(8)	-467(1)	10632(1)	669(1)	13(1)
C(9)	-714(1)	10003(1)	372(1)	13(1)
C(10)	-420(1)	9350(1)	412(1)	13(1)
C(11)	123(1)	9340(1)	740(1)	11(1)
C(12)	412(1)	8616(1)	798(1)	11(1)
C(13)	134(1)	7958(1)	861(1)	13(1)
C(14)	380(1)	7262(1)	909(1)	13(1)
C(15)	901(1)	7235(1)	822(1)	14(1)
C(16)	1192(1)	7866(1)	750(1)	13(1)
C(17)	956(1)	8574(1)	786(1)	11(1)
C(18)	1319(1)	11824(1)	2807(1)	15(1)
C(19)	1173(1)	11197(1)	3181(1)	22(1)
C(20)	1876(1)	11671(1)	2709(1)	23(1)
C(21)	1356(1)	12546(1)	3164(1)	22(1)
C(22)	35(1)	13418(1)	1243(1)	15(1)
C(24)	-102(1)	13893(1)	1721(1)	19(1)
C(23)	-471(1)	13383(1)	716(1)	19(1)
C(25)	495(1)	13800(1)	1052(1)	21(1)
C(26)	118(1)	6550(1)	1067(1)	16(1)
C(27)	-452(1)	6692(1)	1123(1)	30(1)
C(28)	82(1)	5943(1)	602(1)	26(1)
C(29)	468(1)	6256(1)	1663(1)	30(1)
C(30)	1749(1)	7810(1)	630(1)	18(1)
C(32)	1914(1)	6996(1)	586(1)	28(1)
C(31)	1721(1)	8201(1)	44(1)	24(1)
C(33)	2193(1)	8182(1)	1126(1)	24(1)
C(34)	1443(1)	8610(1)	2466(1)	14(1)
C(35)	1480(1)	7864(1)	2323(1)	19(1)
C(36)	1849(1)	7396(1)	2703(1)	25(1)
C(37)	2184(1)	7677(1)	3227(1)	27(1)
C(38)	2144(1)	8421(1)	3369(1)	25(1)
C(39)	1773(1)	8894(1)	2991(1)	19(1)
C(40)	2600(1)	9999(1)	1928(1)	15(1)

C(41)	3024(1)	9629(1)	2336(1)	27(1)
C(42)	3560(1)	9799(1)	2368(1)	35(1)
C(43)	3686(1)	10331(1)	2002(1)	27(1)
C(44)	3269(1)	10698(1)	1599(1)	23(1)
C(45)	2732(1)	10533(1)	1562(1)	20(1)
C(46)	1535(1)	11428(1)	661(1)	13(1)
C(47)	1420(1)	12149(1)	825(1)	19(1)
C(48)	1533(1)	12772(1)	527(1)	26(1)
C(49)	1756(1)	12688(1)	58(1)	28(1)
C(50)	1868(1)	11982(1)	-111(1)	28(1)
C(51)	1763(1)	11352(1)	190(1)	20(1)
C(61)	2518(1)	210(1)	4402(1)	42(1)
C(62)	2770(1)	144(1)	3966(1)	39(1)
C(63)	3323(1)	-13(1)	4102(1)	34(1)
C(64)	3609(1)	-102(1)	4685(1)	35(1)
C(65)	3354(1)	-32(1)	5126(1)	38(1)
C(66)	2807(1)	123(1)	4986(1)	42(1)
C(71)	10000	9883(1)	2500	21(1)
C(72)	9679(1)	9489(1)	2022(1)	22(1)
C(73)	9675(1)	8714(1)	2025(1)	25(1)
C(74)	10000	8323(1)	2500	26(1)

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**Table 3. Selected bond lengths [Å] and angles [°] for IAT03 (CCDC 680573).**

Ta(1)-N(4)	2.3112(8)	N(3)-Ta(1)-O(2)	99.52(4)
Ta(1)-O(2)	1.9999(9)	N(3)-Ta(1)-O(1)	99.46(4)
Ta(1)-O(1)	2.0013(9)	O(2)-Ta(1)-O(1)	150.56(3)
		N(3)-Ta(1)-N(2)	100.59(4)
Ta(1)-N(1)	2.4785(11)	O(2)-Ta(1)-N(2)	98.40(4)
Ta(1)-N(2)	2.0272(9)	O(1)-Ta(1)-N(2)	99.98(4)
Ta(1)-N(3)	1.7872(10)	N(3)-Ta(1)-N(4)	98.47(4)
		O(2)-Ta(1)-N(4)	78.60(3)
		O(1)-Ta(1)-N(4)	76.51(3)
		N(2)-Ta(1)-N(4)	160.94(4)
		N(3)-Ta(1)-N(1)	172.49(4)
		O(2)-Ta(1)-N(1)	87.29(4)
		O(1)-Ta(1)-N(1)	73.04(4)
		N(2)-Ta(1)-N(1)	81.39(4)
		N(4)-Ta(1)-N(1)	79.68(3)
N(1)-C(34)	1.4401(15)	C(34)-N(1)-Ta(1)	127.22(7)
N(2)-C(40)	1.4018(14)	C(40)-N(2)-Ta(1)	139.23(8)
N(3)-C(46)	1.3880(16)	C(46)-N(3)-Ta(1)	176.04(9)



**Table 4. Bond lengths [Å] and angles [°] for IAT03 (CCDC 680573).**

Ta(1)-N(3)	1.7872(10)	C(37)-C(38)	1.381(2)
Ta(1)-O(2)	1.9999(9)	C(38)-C(39)	1.3961(19)
Ta(1)-O(1)	2.0013(9)	C(40)-C(45)	1.3899(19)
Ta(1)-N(2)	2.0272(9)	C(40)-C(41)	1.4061(18)
Ta(1)-N(4)	2.3112(8)	C(41)-C(42)	1.394(2)
Ta(1)-N(1)	2.4785(11)	C(42)-C(43)	1.380(2)
O(1)-C(1)	1.3396(14)	C(43)-C(44)	1.388(2)
O(2)-C(17)	1.3387(14)	C(44)-C(45)	1.3913(18)
N(1)-C(34)	1.4401(15)	C(46)-C(51)	1.3991(18)
N(2)-C(40)	1.4018(14)	C(46)-C(47)	1.3994(18)
N(3)-C(46)	1.3880(16)	C(47)-C(48)	1.389(2)
N(4)-C(11)	1.3662(15)	C(48)-C(49)	1.388(2)
N(4)-C(7)	1.3672(15)	C(49)-C(50)	1.376(3)
C(1)-C(6)	1.4103(15)	C(50)-C(51)	1.395(2)
C(1)-C(2)	1.4166(16)	C(61)-C(62)	1.365(3)
C(2)-C(3)	1.3898(17)	C(61)-C(66)	1.383(3)
C(2)-C(18)	1.5403(16)	C(62)-C(63)	1.398(3)
C(3)-C(4)	1.4068(16)	C(63)-C(64)	1.380(3)
C(4)-C(5)	1.3843(17)	C(64)-C(65)	1.381(3)
C(4)-C(22)	1.5344(17)	C(65)-C(66)	1.383(3)
C(5)-C(6)	1.4056(16)	C(71)-C(72)#1	1.3903(18)
C(6)-C(7)	1.4788(16)	C(71)-C(72)	1.3904(18)
C(7)-C(8)	1.4046(16)	C(72)-C(73)	1.385(2)
C(8)-C(9)	1.3838(17)	C(73)-C(74)	1.387(2)
C(9)-C(10)	1.3784(17)	C(74)-C(73)#1	1.387(2)
C(10)-C(11)	1.3979(16)		
C(11)-C(12)	1.4793(16)	N(3)-Ta(1)-O(2)	99.52(4)
C(12)-C(13)	1.4049(16)	N(3)-Ta(1)-O(1)	99.46(4)
C(12)-C(17)	1.4072(15)	O(2)-Ta(1)-O(1)	150.56(3)
C(13)-C(14)	1.3836(17)	N(3)-Ta(1)-N(2)	100.59(4)
C(14)-C(15)	1.4117(17)	O(2)-Ta(1)-N(2)	98.40(4)
C(14)-C(26)	1.5330(17)	O(1)-Ta(1)-N(2)	99.98(4)
C(15)-C(16)	1.3902(17)	N(3)-Ta(1)-N(4)	98.47(4)
C(16)-C(17)	1.4148(16)	O(2)-Ta(1)-N(4)	78.60(3)
C(16)-C(30)	1.5379(17)	O(1)-Ta(1)-N(4)	76.51(3)
C(18)-C(21)	1.5297(18)	N(2)-Ta(1)-N(4)	160.94(4)
C(18)-C(19)	1.5347(19)	N(3)-Ta(1)-N(1)	172.49(4)
C(18)-C(20)	1.5391(18)	O(2)-Ta(1)-N(1)	87.29(4)
C(22)-C(24)	1.5300(18)	O(1)-Ta(1)-N(1)	73.04(4)
C(22)-C(23)	1.5352(17)	N(2)-Ta(1)-N(1)	81.39(4)
C(22)-C(25)	1.5360(19)	N(4)-Ta(1)-N(1)	79.68(3)
C(26)-C(28)	1.527(2)	C(1)-O(1)-Ta(1)	122.09(7)
C(26)-C(27)	1.5304(19)	C(17)-O(2)-Ta(1)	126.53(7)
C(26)-C(29)	1.5381(19)	C(34)-N(1)-Ta(1)	127.22(7)
C(30)-C(32)	1.526(2)	C(40)-N(2)-Ta(1)	139.23(8)
C(30)-C(31)	1.534(2)	C(46)-N(3)-Ta(1)	176.04(9)
C(30)-C(33)	1.544(2)	C(11)-N(4)-C(7)	119.09(9)
C(34)-C(35)	1.3839(19)	C(11)-N(4)-Ta(1)	122.35(7)
C(34)-C(39)	1.3919(17)	C(7)-N(4)-Ta(1)	118.20(7)
C(35)-C(36)	1.3921(19)	O(1)-C(1)-C(6)	119.63(10)
C(36)-C(37)	1.391(2)	O(1)-C(1)-C(2)	120.14(10)

C(6)-C(1)-C(2)	120.22(10)	C(28)-C(26)-C(27)	108.13(12)
C(3)-C(2)-C(1)	117.56(10)	C(28)-C(26)-C(14)	110.87(11)
C(3)-C(2)-C(18)	121.41(10)	C(27)-C(26)-C(14)	112.29(11)
C(1)-C(2)-C(18)	121.01(10)	C(28)-C(26)-C(29)	108.74(12)
C(2)-C(3)-C(4)	123.50(11)	C(27)-C(26)-C(29)	108.09(13)
C(5)-C(4)-C(3)	117.24(11)	C(14)-C(26)-C(29)	108.63(10)
C(5)-C(4)-C(22)	122.55(10)	C(32)-C(30)-C(31)	108.41(12)
C(3)-C(4)-C(22)	119.97(11)	C(32)-C(30)-C(16)	111.41(11)
C(4)-C(5)-C(6)	121.96(10)	C(31)-C(30)-C(16)	108.74(10)
C(5)-C(6)-C(1)	118.97(10)	C(32)-C(30)-C(33)	107.54(11)
C(5)-C(6)-C(7)	119.28(10)	C(31)-C(30)-C(33)	109.27(12)
C(1)-C(6)-C(7)	121.52(10)	C(16)-C(30)-C(33)	111.40(11)
N(4)-C(7)-C(8)	120.33(10)	C(35)-C(34)-C(39)	120.08(11)
N(4)-C(7)-C(6)	121.72(9)	C(35)-C(34)-N(1)	120.14(11)
C(8)-C(7)-C(6)	117.84(10)	C(39)-C(34)-N(1)	119.75(11)
C(9)-C(8)-C(7)	120.27(11)	C(34)-C(35)-C(36)	120.08(12)
C(10)-C(9)-C(8)	119.03(10)	C(37)-C(36)-C(35)	120.12(14)
C(9)-C(10)-C(11)	119.65(11)	C(38)-C(37)-C(36)	119.63(13)
N(4)-C(11)-C(10)	121.41(10)	C(37)-C(38)-C(39)	120.61(13)
N(4)-C(11)-C(12)	120.99(9)	C(34)-C(39)-C(38)	119.47(13)
C(10)-C(11)-C(12)	117.59(10)	C(45)-C(40)-N(2)	122.52(10)
C(13)-C(12)-C(17)	119.67(10)	C(45)-C(40)-C(41)	118.10(11)
C(13)-C(12)-C(11)	118.89(10)	N(2)-C(40)-C(41)	119.37(12)
C(17)-C(12)-C(11)	121.44(10)	C(42)-C(41)-C(40)	120.27(14)
C(14)-C(13)-C(12)	121.68(10)	C(43)-C(42)-C(41)	121.04(14)
C(13)-C(14)-C(15)	116.72(11)	C(42)-C(43)-C(44)	118.94(12)
C(13)-C(14)-C(26)	122.91(11)	C(43)-C(44)-C(45)	120.59(14)
C(15)-C(14)-C(26)	120.34(11)	C(40)-C(45)-C(44)	121.06(12)
C(16)-C(15)-C(14)	123.70(11)	N(3)-C(46)-C(51)	120.53(12)
C(15)-C(16)-C(17)	117.51(10)	N(3)-C(46)-C(47)	121.02(11)
C(15)-C(16)-C(30)	122.01(11)	C(51)-C(46)-C(47)	118.44(12)
C(17)-C(16)-C(30)	120.48(11)	C(48)-C(47)-C(46)	120.40(13)
O(2)-C(17)-C(12)	118.86(10)	C(49)-C(48)-C(47)	120.54(15)
O(2)-C(17)-C(16)	121.51(10)	C(50)-C(49)-C(48)	119.67(14)
C(12)-C(17)-C(16)	119.63(10)	C(49)-C(50)-C(51)	120.37(14)
C(21)-C(18)-C(19)	106.53(11)	C(50)-C(51)-C(46)	120.57(14)
C(21)-C(18)-C(20)	108.26(11)	C(62)-C(61)-C(66)	120.41(17)
C(19)-C(18)-C(20)	110.07(11)	C(61)-C(62)-C(63)	120.55(18)
C(21)-C(18)-C(2)	111.71(10)	C(64)-C(63)-C(62)	118.84(18)
C(19)-C(18)-C(2)	110.17(10)	C(63)-C(64)-C(65)	120.53(16)
C(20)-C(18)-C(2)	110.04(10)	C(64)-C(65)-C(66)	120.13(18)
C(24)-C(22)-C(4)	111.96(10)	C(61)-C(66)-C(65)	119.5(2)
C(24)-C(22)-C(23)	107.60(10)	C(72)#1-C(71)-C(72)	119.32(19)
C(4)-C(22)-C(23)	111.94(10)	C(73)-C(72)-C(71)	120.32(14)
C(24)-C(22)-C(25)	108.92(11)	C(72)-C(73)-C(74)	120.18(15)
C(4)-C(22)-C(25)	106.99(10)	C(73)-C(74)-C(73)#1	119.7(2)
C(23)-C(22)-C(25)	109.39(11)		

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Symmetry transformations used to generate equivalent atoms:

#1 -x+2,y,-z+1/2

**Table 5. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^4$ ) for IAT03 (CCDC 680573). The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$**

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Ta(1)	88(1)	96(1)	108(1)	6(1)	16(1)	0(1)
O(1)	133(4)	111(4)	121(3)	1(3)	11(3)	19(3)
O(2)	127(4)	101(4)	148(4)	-1(3)	49(3)	-2(3)
N(1)	145(4)	133(4)	142(4)	19(3)	35(3)	9(3)
N(2)	115(4)	149(5)	159(4)	24(3)	24(3)	6(3)
N(3)	115(4)	127(4)	135(4)	-3(3)	27(3)	-2(3)
N(4)	105(3)	103(4)	102(3)	4(3)	24(2)	10(3)
C(1)	121(4)	109(5)	112(4)	6(3)	27(3)	4(4)
C(2)	134(5)	125(5)	116(4)	-9(4)	18(4)	-7(4)
C(3)	172(5)	111(5)	140(4)	-10(4)	21(4)	0(4)
C(4)	152(5)	109(5)	140(4)	6(4)	34(4)	15(4)
C(5)	140(5)	113(5)	121(4)	3(4)	17(4)	9(4)
C(6)	129(4)	99(4)	112(4)	-4(3)	21(3)	5(4)
C(7)	129(4)	99(4)	101(4)	3(3)	27(3)	4(3)
C(8)	120(5)	114(5)	137(4)	0(4)	18(4)	15(4)
C(9)	110(4)	135(5)	134(3)	12(4)	15(3)	1(4)
C(10)	124(5)	117(5)	140(5)	-9(4)	20(4)	-11(4)
C(11)	106(4)	101(4)	122(4)	1(3)	30(3)	-4(3)
C(12)	115(4)	106(5)	119(4)	-5(3)	30(3)	7(3)
C(13)	123(5)	116(5)	140(5)	-4(4)	30(4)	-9(4)
C(14)	150(5)	114(5)	134(4)	4(4)	33(4)	-4(4)
C(15)	152(5)	107(5)	155(5)	-2(4)	33(4)	23(4)
C(16)	139(5)	111(5)	149(5)	1(4)	41(4)	18(4)
C(17)	117(4)	103(4)	111(4)	4(3)	27(3)	-3(3)
C(18)	156(5)	141(5)	130(4)	-15(4)	-1(4)	-3(4)
C(19)	288(7)	206(6)	123(5)	12(4)	9(4)	-26(5)
C(20)	149(5)	271(7)	233(6)	-46(5)	1(5)	-1(5)
C(21)	257(6)	189(6)	170(5)	-52(5)	-33(5)	18(5)
C(22)	181(5)	102(5)	149(5)	8(4)	26(4)	18(4)
C(24)	234(6)	130(5)	197(5)	-17(4)	64(5)	15(4)
C(23)	224(6)	142(5)	175(5)	19(4)	1(4)	36(4)
C(25)	237(6)	158(6)	229(6)	42(5)	80(5)	4(5)
C(26)	181(5)	115(5)	166(5)	17(4)	37(4)	-11(4)
C(27)	238(7)	173(6)	524(10)	45(6)	186(7)	-30(5)
C(28)	394(8)	144(6)	253(6)	-31(5)	113(6)	-80(5)
C(29)	338(8)	279(8)	225(6)	108(6)	-9(6)	-102(6)
C(30)	155(5)	137(5)	257(6)	6(4)	92(4)	32(4)
C(32)	225(6)	161(6)	515(10)	-13(6)	185(6)	52(5)
C(31)	243(6)	245(7)	273(7)	10(5)	161(5)	32(5)
C(33)	135(5)	231(7)	337(7)	-9(6)	29(5)	30(5)
C(34)	144(5)	143(5)	127(4)	32(4)	37(4)	12(4)
C(35)	234(6)	154(6)	164(5)	12(4)	27(4)	17(5)
C(36)	324(7)	165(6)	251(6)	51(5)	51(6)	67(5)
C(37)	250(7)	265(7)	255(7)	116(6)	-10(5)	52(6)
C(38)	256(7)	253(7)	176(5)	54(5)	-29(5)	-27(5)
C(39)	240(6)	163(6)	146(5)	9(4)	22(4)	-21(5)
C(40)	110(4)	167(5)	148(4)	-27(5)	9(3)	9(4)

C(41)	146(5)	379(9)	266(7)	127(6)	11(5)	27(6)
C(42)	114(5)	521(11)	363(8)	143(7)	-4(5)	37(6)
C(43)	121(5)	380(8)	281(7)	-18(6)	30(5)	-16(5)
C(44)	171(6)	268(7)	241(6)	6(5)	56(5)	-34(5)
C(45)	133(5)	211(6)	230(6)	33(5)	27(4)	5(4)
C(46)	115(4)	132(5)	134(4)	17(4)	17(4)	-19(4)
C(47)	203(6)	130(5)	257(6)	23(4)	99(5)	2(4)
C(48)	214(6)	158(6)	419(8)	94(6)	109(6)	13(5)
C(49)	207(6)	277(8)	357(8)	162(6)	76(6)	-25(5)
C(50)	273(7)	357(9)	221(6)	64(6)	105(5)	-78(6)
C(51)	199(6)	222(6)	196(5)	-17(5)	83(5)	-48(5)
C(61)	264(7)	241(8)	731(14)	28(8)	84(8)	-50(6)
C(62)	387(8)	194(8)	483(10)	-10(7)	-49(8)	-54(6)
C(63)	401(8)	197(7)	433(8)	-61(7)	104(7)	-24(7)
C(64)	296(7)	221(8)	495(9)	22(7)	66(7)	26(6)
C(65)	400(8)	280(8)	428(8)	75(8)	76(7)	-49(7)
C(66)	428(9)	293(9)	605(12)	40(8)	241(9)	-66(7)
C(71)	226(8)	187(9)	260(8)	0	137(6)	0
C(72)	223(6)	271(7)	201(6)	13(5)	116(5)	18(5)
C(73)	277(7)	270(7)	249(6)	-76(5)	154(5)	-35(5)
C(74)	340(11)	184(9)	345(11)	0	214(9)	0

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**Crystal Structure Analysis of:**  
**Complex 2 (IAT09)**

Contents

Table 1. Crystal data

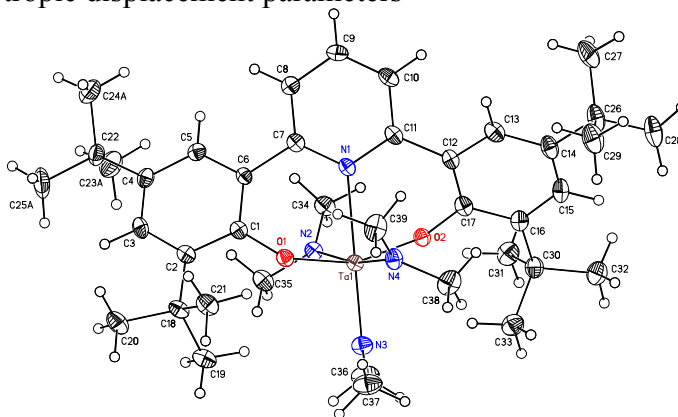
Figures Minimum overlap

Table 2. Atomic Coordinates

Table 3. Selected bond distances and angles

Table 4. Full bond distances and angles

Table 5. Anisotropic displacement parameters



IAT09

**Note:** The crystallographic data have been deposited in the Cambridge Database (CCDC) and has been placed on hold pending further instructions from me. The deposition number is 693510. Ideally the CCDC would like the publication to contain a footnote of the type: "Crystallographic data have been deposited at the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 963510."

**Table 1. Crystal data and structure refinement for IAT09 (CCDC 693510).**

Empirical formula	C <sub>39</sub> H <sub>61</sub> N <sub>4</sub> O <sub>2</sub> Ta
Formula weight	798.87
Crystallization Solvent	Diethylether
Crystal Habit	Blade
Crystal size	0.22 x 0.11 x 0.03 mm <sup>3</sup>
Crystal color	Colorless

### Data Collection

Type of diffractometer	Bruker KAPPA APEX II	
Wavelength	0.71073 Å MoK $\alpha$	
Data Collection Temperature	100(2) K	
$\theta$ range for 9936 reflections used in lattice determination	2.39 to 33.10°	
Unit cell dimensions	a = 9.8149(5) Å b = 29.0840(14) Å c = 14.0826(6) Å	$\beta$ = 104.357(3)°
Volume	3894.4(3) Å <sup>3</sup>	
Z	4	
Crystal system	Monoclinic	
Space group	P2 <sub>1</sub> /c	
Density (calculated)	1.363 Mg/m <sup>3</sup>	
F(000)	1648	
$\theta$ range for data collection	1.65 to 33.20°	
Completeness to $\theta$ = 33.20°	93.8 %	
Index ranges	-13 $\leq$ h $\leq$ 14, -39 $\leq$ k $\leq$ 43, -21 $\leq$ l $\leq$ 20	
Data collection scan type	$\omega$ scans; 7 settings	
Reflections collected	53883	
Independent reflections	13983 [R <sub>int</sub> = 0.0450]	
Absorption coefficient	2.858 mm <sup>-1</sup>	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.4353 and 0.3323	

**Table 1 (cont.)****Structure solution and Refinement**

Structure solution program	SHELXS-97 (Sheldrick, 2008)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 2008)
Refinement method	Full matrix least-squares on $F^2$
Data / restraints / parameters	13983 / 0 / 449
Treatment of hydrogen atoms	Riding
Goodness-of-fit on $F^2$	1.426
Final R indices [ $I > 2\sigma(I)$ , 10288 reflections]	$R1 = 0.0350$ , $wR2 = 0.0448$
R indices (all data)	$R1 = 0.0631$ , $wR2 = 0.0471$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.002
Average shift/error	0.000
Largest diff. peak and hole	1.666 and -1.582 e.Å <sup>-3</sup>

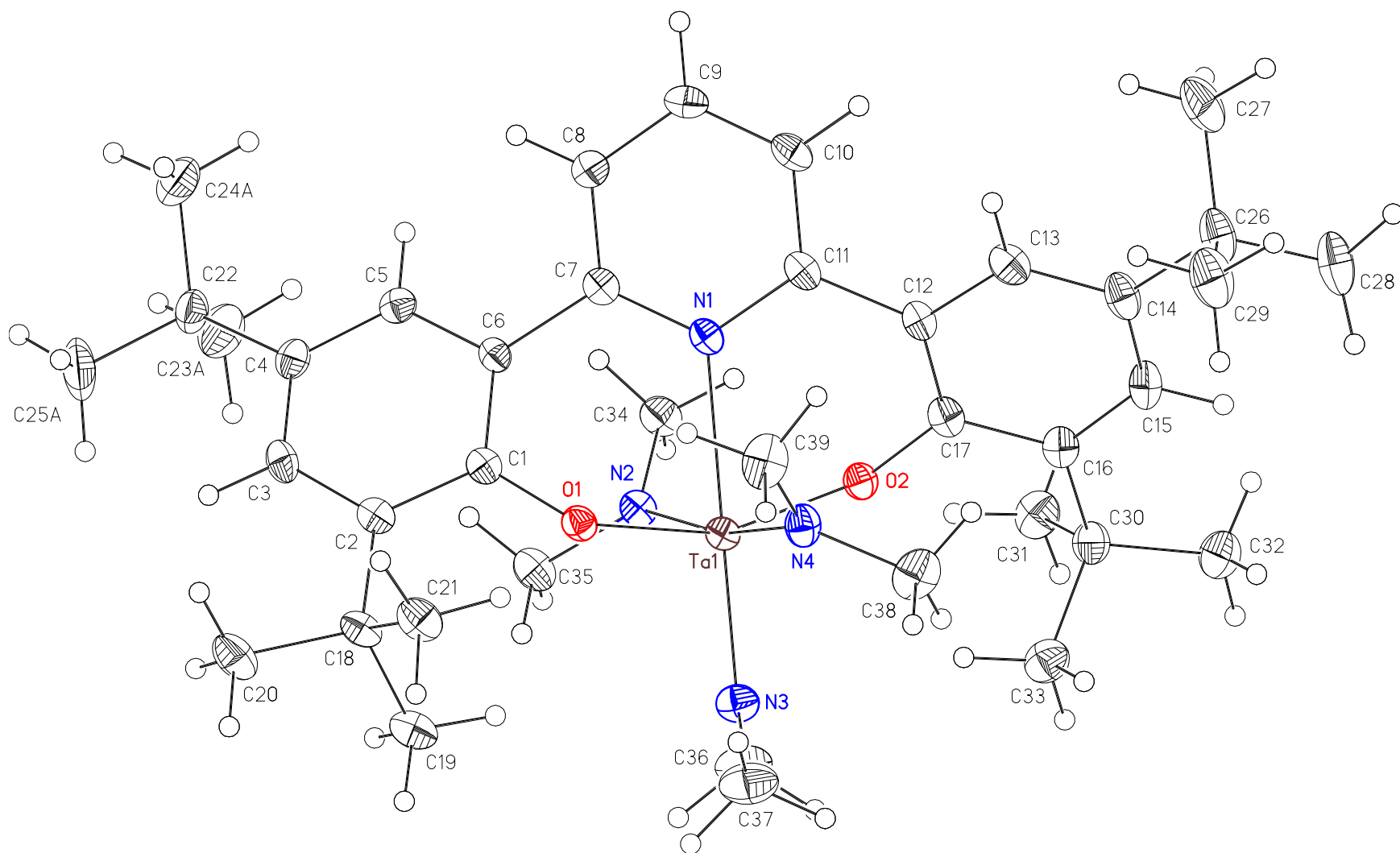
**Special Refinement Details**

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

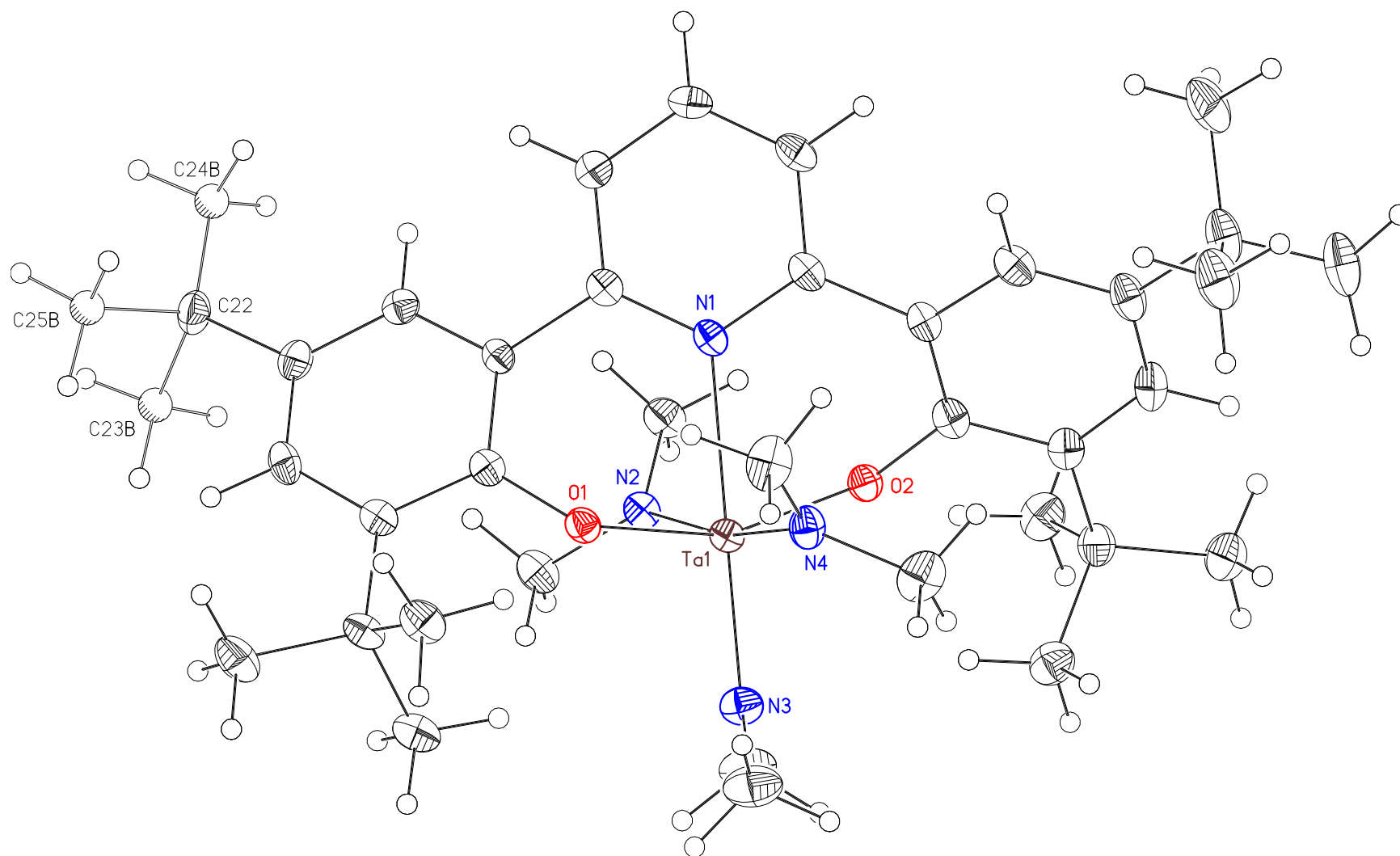
One t-Bu group is disordered as shown in the third figure. The disorder is included in the model.

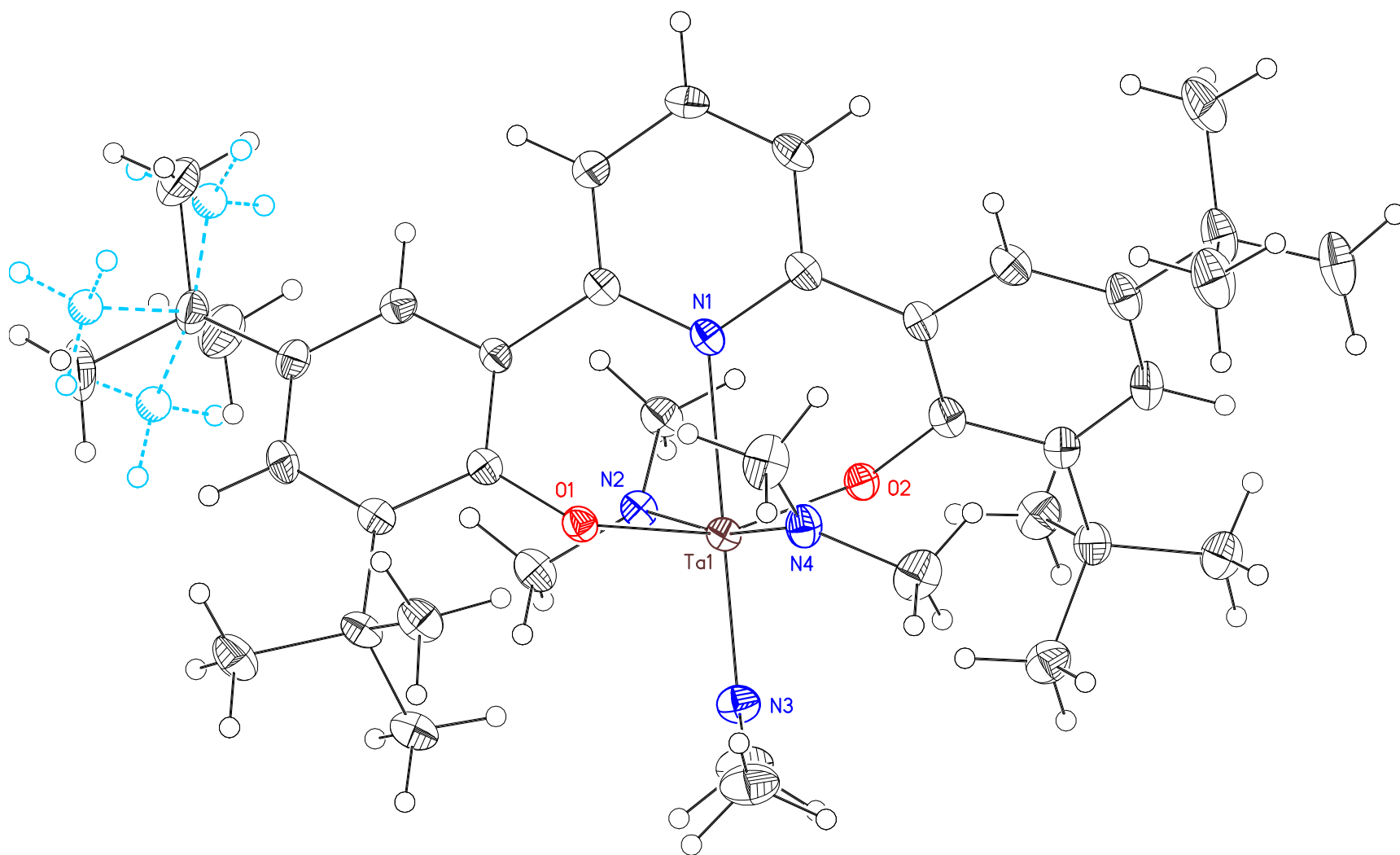
Refinement of  $F^2$  against ALL reflections. The weighted R-factor ( $wR$ ) and goodness of fit ( $S$ ) are based on  $F^2$ , conventional R-factors ( $R$ ) are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.









**Table 2. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for IAT09 (CCDC 693510).  $U(\text{eq})$  is defined as the trace of the orthogonalized  $U^i$  tensor.**

	x	y	z	$U_{\text{eq}}$	Occ
Ta(1)	1345(1)	1155(1)	2702(1)	15(1)	1
O(1)	1189(2)	477(1)	2796(1)	17(1)	1
O(2)	1346(2)	1764(1)	2088(1)	17(1)	1
N(1)	894(2)	951(1)	1064(1)	16(1)	1
N(2)	3396(2)	1117(1)	2672(1)	18(1)	1
N(3)	1705(2)	1328(1)	4111(1)	21(1)	1
N(4)	-786(2)	1151(1)	2459(1)	21(1)	1
C(1)	1969(2)	164(1)	2467(2)	17(1)	1
C(2)	2507(2)	-219(1)	3039(2)	18(1)	1
C(3)	3510(2)	-485(1)	2743(2)	20(1)	1
C(4)	3959(2)	-388(1)	1890(2)	19(1)	1
C(5)	3260(2)	-48(1)	1286(2)	18(1)	1
C(6)	2239(2)	230(1)	1537(2)	15(1)	1
C(7)	1399(2)	548(1)	800(2)	16(1)	1
C(8)	1072(2)	411(1)	-175(2)	19(1)	1
C(9)	273(2)	691(1)	-886(2)	22(1)	1
C(10)	-197(2)	1101(1)	-621(2)	22(1)	1
C(11)	97(2)	1232(1)	365(2)	18(1)	1
C(12)	-434(2)	1682(1)	612(2)	18(1)	1
C(13)	-1656(2)	1866(1)	-14(2)	22(1)	1
C(14)	-2108(2)	2307(1)	87(2)	23(1)	1
C(15)	-1277(2)	2577(1)	830(2)	24(1)	1
C(16)	-75(2)	2417(1)	1498(2)	20(1)	1
C(17)	297(2)	1954(1)	1405(2)	19(1)	1
C(18)	2055(2)	-332(1)	3988(2)	21(1)	1
C(19)	2628(3)	29(1)	4782(2)	27(1)	1
C(20)	2596(3)	-803(1)	4398(2)	29(1)	1
C(21)	440(2)	-347(1)	3776(2)	25(1)	1
C(22)	5215(2)	-642(1)	1666(2)	23(1)	1
C(23A)	6447(3)	-294(1)	1834(2)	34(1)	0.888(5)
C(24A)	4870(4)	-799(1)	589(2)	36(1)	0.888(5)
C(25A)	5690(4)	-1055(1)	2310(3)	46(1)	0.888(5)
C(25B)	4880(30)	-1182(10)	1751(19)	36(6)	0.112(5)
C(23B)	6550(20)	-563(9)	2531(16)	29(6)	0.112(5)
C(24B)	5430(30)	-556(11)	726(18)	32(6)	0.112(5)
C(26)	-3486(2)	2486(1)	-584(2)	28(1)	1
C(27)	-3582(3)	2366(1)	-1661(2)	36(1)	1
C(28)	-3617(3)	3004(1)	-511(2)	38(1)	1
C(29)	-4722(2)	2258(1)	-267(2)	36(1)	1
C(30)	810(2)	2731(1)	2287(2)	22(1)	1
C(31)	2337(2)	2742(1)	2174(2)	28(1)	1
C(32)	271(3)	3224(1)	2202(2)	32(1)	1
C(33)	815(3)	2563(1)	3329(2)	27(1)	1
C(34)	4013(2)	1300(1)	1896(2)	23(1)	1
C(35)	4436(2)	842(1)	3352(2)	23(1)	1
C(36)	2921(3)	1581(1)	4660(2)	34(1)	1
C(37)	746(3)	1237(1)	4731(2)	36(1)	1
C(38)	-1601(2)	1538(1)	2691(2)	30(1)	1

C(39)	-1734(2)	800(1)	1941(2)	31(1)	1
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**Table 3. Selected bond lengths [Å] and angles [°] for IAT09 (CCDC 693510).**

Ta(1)-O(2)	1.9721(16)	O(2)-Ta(1)-O(1)	158.51(6)
Ta(1)-O(1)	1.9848(16)	O(2)-Ta(1)-N(3)	100.93(7)
Ta(1)-N(3)	1.9914(18)	O(1)-Ta(1)-N(3)	100.55(7)
Ta(1)-N(2)	2.0270(16)	O(2)-Ta(1)-N(2)	86.07(7)
Ta(1)-N(4)	2.0336(17)	O(1)-Ta(1)-N(2)	92.31(7)
Ta(1)-N(1)	2.3156(17)	N(3)-Ta(1)-N(2)	95.91(7)
		O(2)-Ta(1)-N(4)	92.49(7)
		O(1)-Ta(1)-N(4)	84.98(8)
		N(3)-Ta(1)-N(4)	95.21(7)
		N(2)-Ta(1)-N(4)	168.86(7)
		O(2)-Ta(1)-N(1)	79.26(6)
		O(1)-Ta(1)-N(1)	79.25(6)
		N(3)-Ta(1)-N(1)	179.22(7)
		N(2)-Ta(1)-N(1)	84.85(6)
		N(4)-Ta(1)-N(1)	84.02(6)

**Table 4. Bond lengths [Å] and angles [°] for IAT09 (CCDC 693510).**

Ta(1)-O(2)	1.9721(16)	C(30)-C(33)	1.546(3)
Ta(1)-O(1)	1.9848(16)	C(30)-C(31)	1.547(3)
Ta(1)-N(3)	1.9914(18)		
Ta(1)-N(2)	2.0270(16)	O(2)-Ta(1)-O(1)	158.51(6)
Ta(1)-N(4)	2.0336(17)	O(2)-Ta(1)-N(3)	100.93(7)
Ta(1)-N(1)	2.3156(17)	O(1)-Ta(1)-N(3)	100.55(7)
O(1)-C(1)	1.343(3)	O(2)-Ta(1)-N(2)	86.07(7)
O(2)-C(17)	1.342(3)	O(1)-Ta(1)-N(2)	92.31(7)
N(1)-C(7)	1.361(3)	N(3)-Ta(1)-N(2)	95.91(7)
N(1)-C(11)	1.366(3)	O(2)-Ta(1)-N(4)	92.49(7)
N(2)-C(35)	1.454(3)	O(1)-Ta(1)-N(4)	84.98(8)
N(2)-C(34)	1.475(3)	N(3)-Ta(1)-N(4)	95.21(7)
N(3)-C(36)	1.450(3)	N(2)-Ta(1)-N(4)	168.86(7)
N(3)-C(37)	1.458(3)	O(2)-Ta(1)-N(1)	79.26(6)
N(4)-C(39)	1.451(3)	O(1)-Ta(1)-N(1)	79.25(6)
N(4)-C(38)	1.463(3)	N(3)-Ta(1)-N(1)	179.22(7)
C(1)-C(2)	1.398(3)	N(2)-Ta(1)-N(1)	84.85(6)
C(1)-C(6)	1.413(3)	N(4)-Ta(1)-N(1)	84.02(6)
C(2)-C(3)	1.396(3)	C(1)-O(1)-Ta(1)	126.16(13)
C(2)-C(18)	1.545(3)	C(17)-O(2)-Ta(1)	126.91(14)
C(3)-C(4)	1.407(3)	C(7)-N(1)-C(11)	119.99(18)
C(4)-C(5)	1.373(3)	C(7)-N(1)-Ta(1)	120.01(14)
C(4)-C(22)	1.536(3)	C(11)-N(1)-Ta(1)	119.98(15)
C(5)-C(6)	1.399(3)	C(35)-N(2)-C(34)	109.66(16)
C(6)-C(7)	1.478(3)	C(35)-N(2)-Ta(1)	123.71(13)
C(7)-C(8)	1.388(3)	C(34)-N(2)-Ta(1)	126.07(14)
C(8)-C(9)	1.375(3)	C(36)-N(3)-C(37)	109.94(18)
C(9)-C(10)	1.365(3)	C(36)-N(3)-Ta(1)	125.38(14)
C(10)-C(11)	1.398(3)	C(37)-N(3)-Ta(1)	124.63(15)
C(11)-C(12)	1.483(3)	C(39)-N(4)-C(38)	109.57(18)
C(12)-C(17)	1.410(3)	C(39)-N(4)-Ta(1)	126.43(16)
C(12)-C(13)	1.407(3)	C(38)-N(4)-Ta(1)	123.65(17)
C(13)-C(14)	1.376(3)	O(1)-C(1)-C(2)	120.42(18)
C(14)-C(15)	1.397(3)	O(1)-C(1)-C(6)	118.8(2)
C(14)-C(26)	1.536(3)	C(2)-C(1)-C(6)	120.8(2)
C(15)-C(16)	1.394(3)	C(3)-C(2)-C(1)	117.63(19)
C(16)-C(17)	1.409(3)	C(3)-C(2)-C(18)	121.0(2)
C(16)-C(30)	1.532(3)	C(1)-C(2)-C(18)	121.3(2)
C(18)-C(20)	1.529(4)	C(4)-C(3)-C(2)	122.5(2)
C(18)-C(19)	1.536(3)	C(5)-C(4)-C(3)	117.4(2)
C(18)-C(21)	1.539(3)	C(5)-C(4)-C(22)	121.5(2)
C(22)-C(24B)	1.41(2)	C(3)-C(4)-C(22)	121.1(2)
C(22)-C(25A)	1.506(4)	C(4)-C(5)-C(6)	122.6(2)
C(22)-C(24A)	1.538(4)	C(5)-C(6)-C(1)	117.8(2)
C(22)-C(23A)	1.551(4)	C(5)-C(6)-C(7)	119.58(18)
C(22)-C(23B)	1.57(2)	C(1)-C(6)-C(7)	122.37(19)
C(22)-C(25B)	1.61(3)	N(1)-C(7)-C(8)	120.4(2)
C(26)-C(28)	1.517(4)	N(1)-C(7)-C(6)	121.62(19)
C(26)-C(27)	1.538(3)	C(8)-C(7)-C(6)	117.9(2)
C(26)-C(29)	1.543(3)	C(9)-C(8)-C(7)	120.1(2)
C(30)-C(32)	1.521(4)	C(10)-C(9)-C(8)	119.2(2)

C(9)-C(10)-C(11)	120.6(2)	C(24A)-C(22)-C(4)	110.7(2)
N(1)-C(11)-C(10)	119.6(2)	C(24B)-C(22)-C(23A)	74.6(12)
N(1)-C(11)-C(12)	121.99(19)	C(25A)-C(22)-C(23A)	108.4(2)
C(10)-C(11)-C(12)	118.3(2)	C(24A)-C(22)-C(23A)	108.5(2)
C(17)-C(12)-C(13)	118.0(2)	C(4)-C(22)-C(23A)	106.7(2)
C(17)-C(12)-C(11)	122.8(2)	C(24B)-C(22)-C(23B)	114.2(13)
C(13)-C(12)-C(11)	118.9(2)	C(25A)-C(22)-C(23B)	64.9(10)
C(14)-C(13)-C(12)	122.4(2)	C(24A)-C(22)-C(23B)	138.2(8)
C(13)-C(14)-C(15)	117.2(2)	C(4)-C(22)-C(23B)	109.2(8)
C(13)-C(14)-C(26)	120.6(2)	C(23A)-C(22)-C(23B)	46.8(9)
C(15)-C(14)-C(26)	122.1(2)	C(24B)-C(22)-C(25B)	108.9(15)
C(16)-C(15)-C(14)	123.8(2)	C(25A)-C(22)-C(25B)	38.6(9)
C(15)-C(16)-C(17)	116.9(2)	C(24A)-C(22)-C(25B)	77.7(9)
C(15)-C(16)-C(30)	121.5(2)	C(4)-C(22)-C(25B)	105.4(9)
C(17)-C(16)-C(30)	121.6(2)	C(23A)-C(22)-C(25B)	142.2(10)
O(2)-C(17)-C(16)	119.8(2)	C(23B)-C(22)-C(25B)	103.3(13)
O(2)-C(17)-C(12)	119.1(2)	C(28)-C(26)-C(27)	107.9(2)
C(16)-C(17)-C(12)	121.1(2)	C(28)-C(26)-C(14)	111.8(2)
C(20)-C(18)-C(2)	111.8(2)	C(27)-C(26)-C(14)	111.0(2)
C(20)-C(18)-C(19)	107.98(19)	C(28)-C(26)-C(29)	108.8(2)
C(2)-C(18)-C(19)	110.55(19)	C(27)-C(26)-C(29)	109.1(2)
C(20)-C(18)-C(21)	107.0(2)	C(14)-C(26)-C(29)	108.2(2)
C(2)-C(18)-C(21)	109.93(18)	C(32)-C(30)-C(16)	112.4(2)
C(19)-C(18)-C(21)	109.5(2)	C(32)-C(30)-C(33)	107.0(2)
C(24B)-C(22)-C(25A)	127.6(12)	C(16)-C(30)-C(33)	111.6(2)
C(24B)-C(22)-C(24A)	34.7(12)	C(32)-C(30)-C(31)	107.4(2)
C(25A)-C(22)-C(24A)	108.6(3)	C(16)-C(30)-C(31)	109.10(19)
C(24B)-C(22)-C(4)	114.9(11)	C(33)-C(30)-C(31)	109.22(19)
C(25A)-C(22)-C(4)	113.8(2)		

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**Table 5. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^4$ ) for IAT09 (CCDC 693510). The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$**

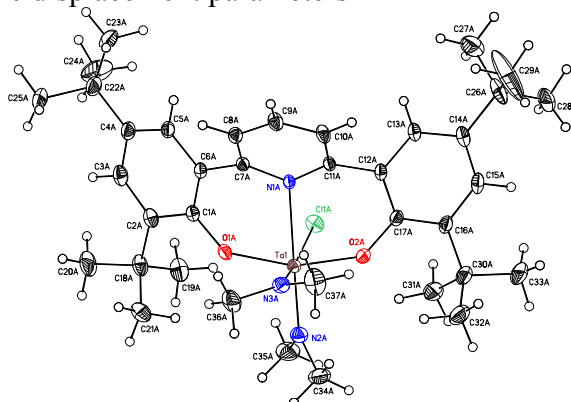
	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Ta(1)	139(1)	170(1)	152(1)	9(1)	38(1)	-13(1)
O(1)	166(8)	193(10)	171(7)	29(7)	59(6)	10(7)
O(2)	152(8)	181(10)	187(7)	8(7)	36(6)	6(7)
N(1)	134(9)	166(12)	171(9)	35(8)	35(7)	-12(8)
N(2)	152(8)	188(12)	181(8)	14(9)	35(7)	-13(9)
N(3)	224(10)	215(13)	205(9)	5(8)	92(8)	-60(9)
N(4)	162(8)	209(12)	261(9)	25(10)	69(7)	19(10)
C(1)	125(10)	180(14)	182(10)	6(10)	18(8)	-21(10)
C(2)	162(11)	187(14)	176(10)	12(10)	28(9)	-35(10)
C(3)	199(12)	141(14)	247(12)	46(10)	30(10)	-5(10)
C(4)	201(11)	133(14)	236(11)	-27(10)	52(10)	-27(10)
C(5)	191(11)	159(14)	184(11)	-16(10)	55(9)	-34(10)
C(6)	154(10)	125(13)	162(10)	11(9)	13(9)	-28(9)
C(7)	117(10)	175(14)	187(10)	18(10)	44(9)	-36(9)
C(8)	191(11)	182(15)	200(11)	-5(10)	61(9)	-19(10)
C(9)	235(12)	264(17)	145(10)	-10(10)	30(9)	-30(11)
C(10)	186(10)	278(17)	178(10)	43(11)	-8(8)	5(11)
C(11)	146(10)	194(16)	190(10)	43(10)	21(8)	-18(10)
C(12)	154(11)	175(15)	204(11)	38(10)	30(9)	-3(10)
C(13)	176(11)	248(16)	225(11)	31(11)	18(9)	-27(11)
C(14)	160(11)	226(16)	301(12)	84(11)	56(10)	18(11)
C(15)	221(12)	166(15)	337(13)	40(11)	88(11)	28(11)
C(16)	178(11)	187(15)	243(11)	18(10)	66(10)	-4(10)
C(17)	151(11)	205(15)	226(11)	51(10)	74(9)	-8(10)
C(18)	209(12)	245(16)	169(10)	48(10)	36(9)	-19(11)
C(19)	295(13)	324(18)	194(11)	46(11)	52(10)	-22(12)
C(20)	280(14)	311(18)	272(13)	121(12)	71(11)	0(12)
C(21)	230(12)	287(17)	258(12)	72(11)	92(10)	-41(11)
C(22)	225(12)	194(15)	282(12)	18(11)	101(10)	59(11)
C(23A)	249(16)	370(20)	442(19)	-55(16)	126(14)	42(14)
C(24A)	390(19)	310(20)	368(18)	-88(16)	107(15)	87(17)
C(25A)	570(20)	340(30)	560(20)	204(18)	280(20)	281(19)
C(26)	174(12)	249(17)	403(15)	99(12)	41(11)	40(11)
C(27)	234(13)	430(20)	381(15)	132(14)	-15(12)	100(13)
C(28)	269(14)	299(19)	528(17)	122(15)	40(13)	103(13)
C(29)	184(12)	390(20)	482(16)	102(15)	47(12)	64(12)
C(30)	224(12)	187(15)	260(12)	9(11)	60(10)	-9(11)
C(31)	286(14)	213(16)	340(14)	-18(12)	97(12)	-70(12)
C(32)	391(16)	220(17)	335(14)	-21(12)	78(12)	7(13)
C(33)	289(14)	247(17)	284(13)	-31(12)	91(11)	-40(12)
C(34)	186(11)	279(17)	222(11)	37(10)	69(10)	2(10)
C(35)	164(11)	260(16)	241(12)	40(11)	10(10)	-31(11)
C(36)	376(16)	400(20)	243(13)	-71(13)	97(12)	-88(14)
C(37)	411(15)	460(20)	267(12)	-15(13)	185(12)	-69(14)
C(38)	226(13)	324(18)	398(14)	-7(13)	195(12)	-32(12)
C(39)	190(12)	282(18)	447(16)	-27(13)	87(12)	-43(12)



# Crystal Structure Analysis of: Complex 3 (IAT10)

## Contents

Table 1. Crystal data
Figures Minimum overlap
Table 2. Atomic Coordinates
Table 3. Selected bond distances and angles
Table 4. Full bond distances and angles
Table 5. Anisotropic displacement parameters



IAT10

**Note:** The crystallographic data have been deposited in the Cambridge Database (CCDC) and has been placed on hold pending further instructions from me. The deposition number is 693797. Ideally the CCDC would like the publication to contain a footnote of the type: "Crystallographic data have been deposited at the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 693797."

**Table 1. Crystal data and structure refinement for IAT10 (CCDC 693797).**

Empirical formula	C <sub>37</sub> H <sub>55</sub> N <sub>3</sub> O <sub>2</sub> ClTa • 0.75(C <sub>4</sub> H <sub>10</sub> O)
Formula weight	845.83
Crystallization Solvent	Diethylether
Crystal Habit	Block
Crystal size	0.29 x 0.28 x 0.23 mm <sup>3</sup>
Crystal color	Pale yellow

### Data Collection

Type of diffractometer	Bruker KAPPA APEX II	
Wavelength	0.71073 Å MoK $\alpha$	
Data Collection Temperature	100(2) K	
$\theta$ range for 9746 reflections used in lattice determination	2.63 to 46.81°	
Unit cell dimensions	a = 14.7164(6) Å b = 18.3370(7) Å c = 31.2188(13) Å	$\beta$ = 101.802(2)°
Volume	8246.4(6) Å <sup>3</sup>	
Z	8	
Crystal system	Monoclinic	
Space group	P2 <sub>1</sub> /n	
Density (calculated)	1.363 Mg/m <sup>3</sup>	
F(000)	3484	
Data collection program	Bruker APEX2 v2.1-0	
$\theta$ range for data collection	1.68 to 47.98°	
Completeness to $\theta$ = 47.98°	96.0 %	
Index ranges	-30 ≤ h ≤ 29, -38 ≤ k ≤ 37, -64 ≤ l ≤ 64	
Data collection scan type	$\omega$ scans; 25 settings	
Data reduction program	Bruker SAINT-Plus v7.34A	
Reflections collected	492193	
Independent reflections	75746 [R <sub>int</sub> = 0.0554]	
Absorption coefficient	2.767 mm <sup>-1</sup>	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.4493 and 0.3540	

**Table 1 (cont.)****Structure solution and Refinement**

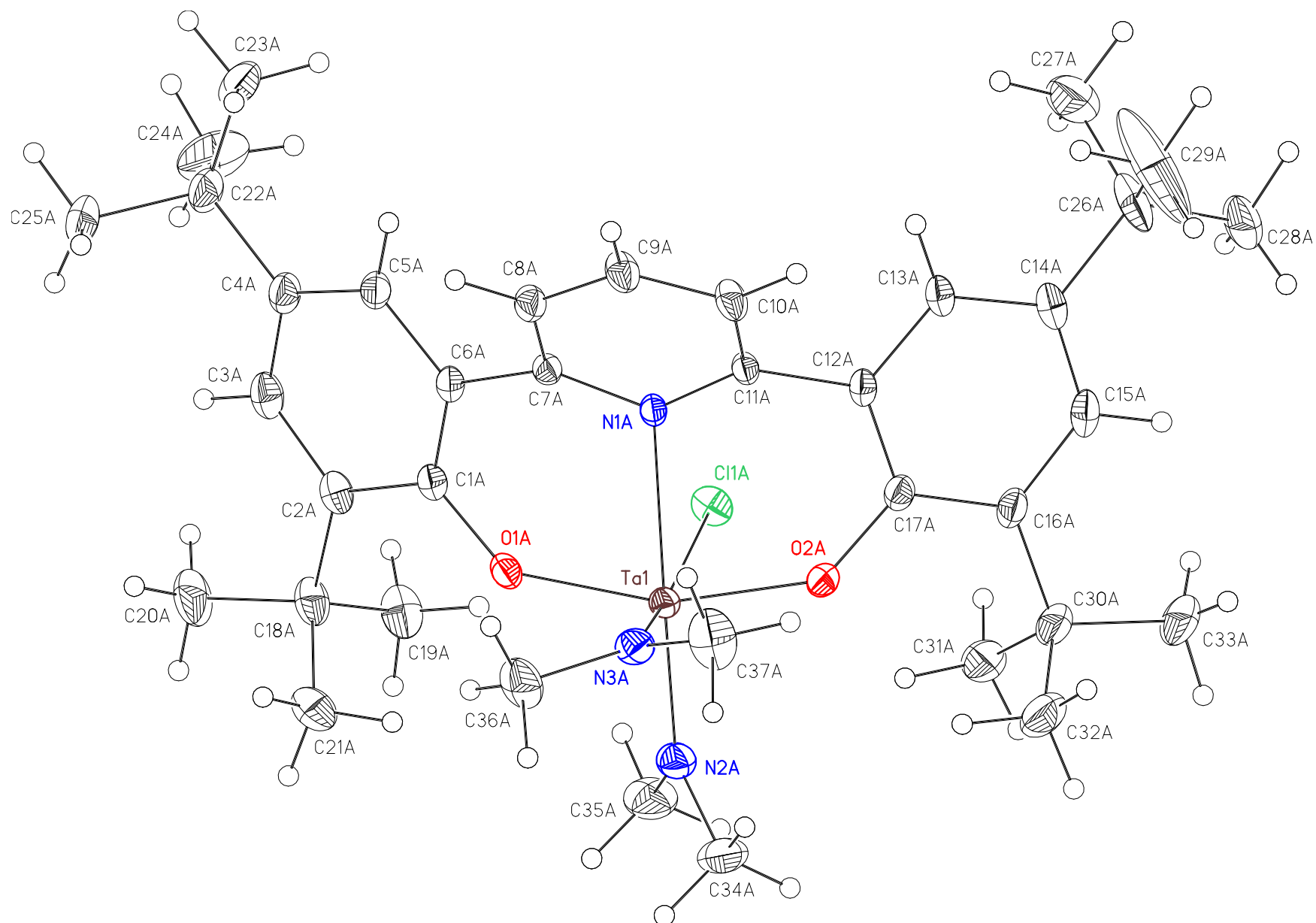
Structure solution program	SHELXS-97 (Sheldrick, 2008)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 2008)
Refinement method	Full matrix least-squares on $F^2$
Data / restraints / parameters	75746 / 0 / 905
Treatment of hydrogen atoms	Riding
Goodness-of-fit on $F^2$	2.760
Final R indices [ $I > 2\sigma(I)$ , 50883 reflections]	$R1 = 0.0731$ , $wR2 = 0.1059$
R indices (all data)	$R1 = 0.1097$ , $wR2 = 0.1074$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.004
Average shift/error	0.000
Largest diff. peak and hole	10.132 and -8.855 e.Å <sup>-3</sup>

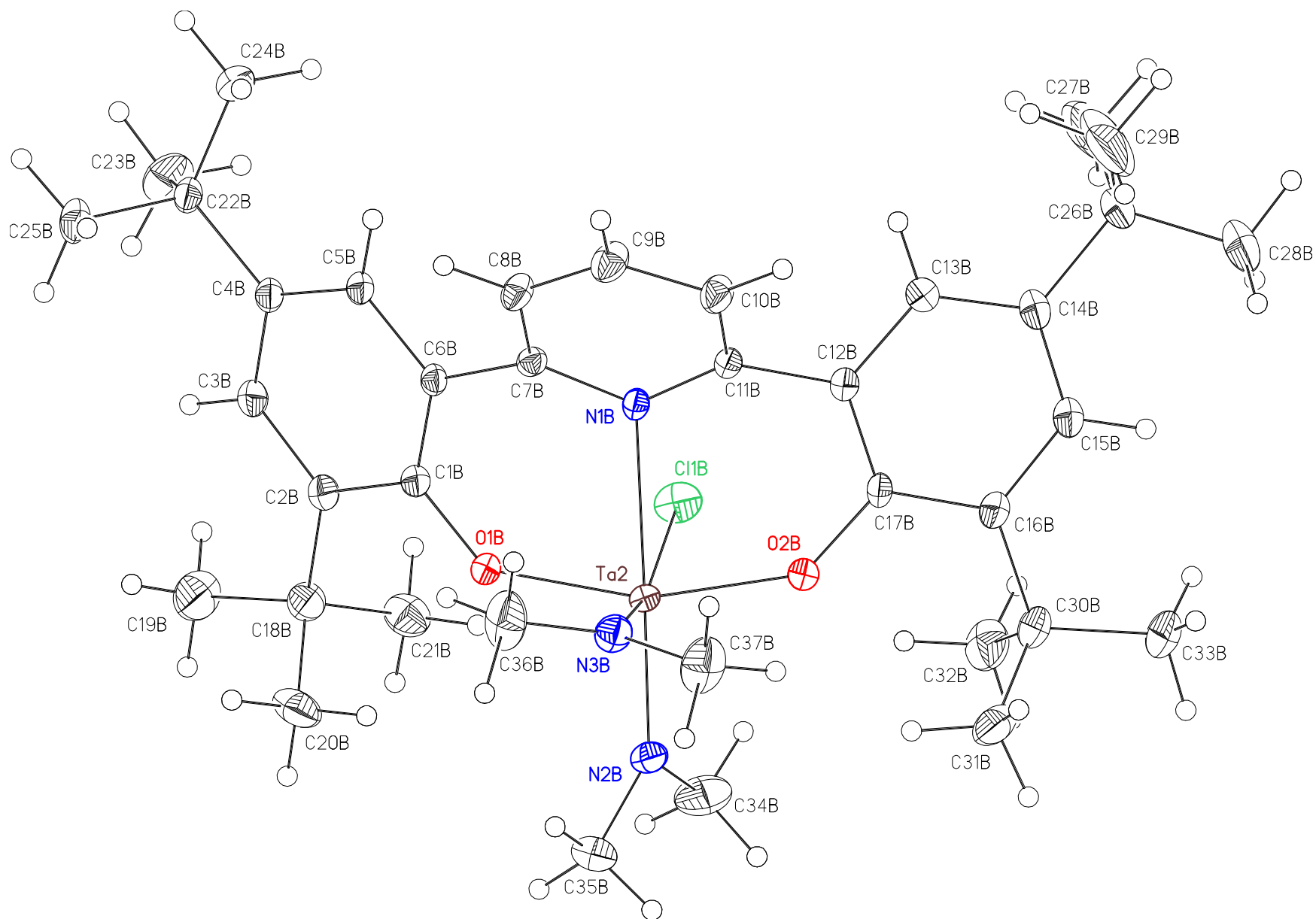
**Special Refinement Details**

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

Refinement of  $F^2$  against ALL reflections. The weighted R-factor ( $wR$ ) and goodness of fit ( $S$ ) are based on  $F^2$ , conventional R-factors ( $R$ ) are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.





**Table 2. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for IAT10 (CCDC 693797).  $U_{\text{eq}}$  is defined as the trace of the orthogonalized  $U^{\text{ij}}$  tensor.**

	x	y	z	$U_{\text{eq}}$
Ta(1)	6285(1)	7674(1)	2306(1)	15(1)
Cl(1A)	4780(1)	7080(1)	2305(1)	21(1)
O(1A)	6501(1)	7630(1)	2936(1)	17(1)
O(2A)	5755(1)	7931(1)	1711(1)	18(1)
N(1A)	5468(1)	8751(1)	2438(1)	13(1)
N(2A)	6943(2)	6783(1)	2188(1)	23(1)
N(3A)	7327(1)	8353(1)	2289(1)	20(1)
C(1A)	6178(2)	7907(1)	3277(1)	16(1)
C(2A)	6265(2)	7497(1)	3669(1)	20(1)
C(3A)	5884(2)	7814(1)	4000(1)	23(1)
C(4A)	5460(2)	8500(1)	3970(1)	21(1)
C(5A)	5438(2)	8886(1)	3590(1)	18(1)
C(6A)	5799(2)	8600(1)	3240(1)	15(1)
C(7A)	5705(2)	9057(1)	2843(1)	14(1)
C(8A)	5782(2)	9813(1)	2893(1)	18(1)
C(9A)	5553(2)	10264(1)	2533(1)	22(1)
C(10A)	5188(2)	9952(1)	2131(1)	20(1)
C(11A)	5139(2)	9195(1)	2090(1)	14(1)
C(12A)	4661(2)	8882(1)	1666(1)	16(1)
C(13A)	3860(2)	9232(1)	1442(1)	18(1)
C(14A)	3365(2)	8970(1)	1044(1)	21(1)
C(15A)	3710(2)	8348(1)	873(1)	23(1)
C(16A)	4506(2)	7977(1)	1081(1)	20(1)
C(17A)	4973(2)	8253(1)	1486(1)	17(1)
C(18A)	6717(2)	6745(1)	3726(1)	24(1)
C(19A)	6065(2)	6198(1)	3444(1)	30(1)
C(20A)	6875(2)	6483(2)	4204(1)	35(1)
C(21A)	7678(2)	6752(2)	3598(1)	30(1)
C(22A)	5034(2)	8780(2)	4345(1)	27(1)
C(23A)	4576(2)	9532(2)	4251(1)	31(1)
C(24A)	4275(3)	8251(2)	4418(1)	49(1)
C(25A)	5785(3)	8863(3)	4761(1)	49(1)
C(26A)	2477(2)	9362(2)	811(1)	34(1)
C(27A)	1799(3)	9415(2)	1146(2)	52(1)
C(28A)	1942(2)	8937(2)	431(1)	40(1)
C(29A)	2695(4)	10121(3)	688(2)	101(3)
C(30A)	4866(2)	7303(2)	872(1)	29(1)
C(31A)	4892(2)	6636(2)	1171(1)	33(1)
C(32A)	5848(2)	7471(2)	793(1)	40(1)
C(33A)	4250(3)	7115(2)	430(1)	42(1)
C(34A)	7636(2)	6782(2)	1909(1)	32(1)
C(35A)	6777(2)	6051(1)	2327(1)	34(1)
C(36A)	8104(2)	8435(2)	2661(1)	31(1)
C(37A)	7418(2)	8875(2)	1947(1)	38(1)

Ta(2)	3444(1)	2681(1)	2235(1)	14(1)
Cl(1B)	4965(1)	2064(1)	2376(1)	24(1)
O(1B)	3667(1)	2893(1)	1662(1)	16(1)
O(2B)	3565(1)	2672(1)	2858(1)	18(1)
N(1B)	4417(1)	3736(1)	2408(1)	13(1)
N(2B)	2655(2)	1808(1)	2082(1)	22(1)
N(3B)	2429(1)	3406(1)	2150(1)	20(1)
C(1B)	4299(2)	3226(1)	1474(1)	16(1)
C(2B)	4534(2)	2933(1)	1093(1)	18(1)
C(3B)	5177(2)	3322(1)	912(1)	19(1)
C(4B)	5602(2)	3963(1)	1091(1)	16(1)
C(5B)	5346(2)	4232(1)	1464(1)	15(1)
C(6B)	4697(2)	3874(1)	1660(1)	14(1)
C(7B)	4508(1)	4186(1)	2071(1)	13(1)
C(8B)	4510(2)	4943(1)	2120(1)	18(1)
C(9B)	4453(2)	5247(1)	2518(1)	22(1)
C(10B)	4476(2)	4785(1)	2871(1)	21(1)
C(11B)	4479(2)	4034(1)	2812(1)	15(1)
C(12B)	4611(2)	3566(1)	3207(1)	17(1)
C(13B)	5230(2)	3802(1)	3581(1)	22(1)
C(14B)	5348(2)	3421(1)	3972(1)	23(1)
C(15B)	4819(2)	2796(1)	3982(1)	22(1)
C(16B)	4201(2)	2522(1)	3618(1)	19(1)
C(17B)	4127(2)	2914(1)	3223(1)	16(1)
C(18B)	4136(2)	2199(1)	899(1)	24(1)
C(19B)	4450(3)	2026(2)	470(1)	41(1)
C(20B)	3066(2)	2202(1)	796(1)	30(1)
C(21B)	4494(2)	1588(1)	1228(1)	27(1)
C(22B)	6312(2)	4352(1)	875(1)	18(1)
C(23B)	7017(2)	3806(2)	766(1)	33(1)
C(24B)	6851(2)	4936(2)	1174(1)	30(1)
C(25B)	5801(2)	4714(2)	453(1)	27(1)
C(26B)	6058(2)	3681(2)	4374(1)	30(1)
C(27B)	7029(3)	3495(4)	4311(2)	80(2)
C(28B)	5921(3)	3315(2)	4794(1)	44(1)
C(29B)	5956(5)	4502(2)	4441(2)	81(2)
C(30B)	3646(2)	1820(1)	3641(1)	26(1)
C(31B)	2610(2)	1906(2)	3433(1)	34(1)
C(32B)	4077(2)	1214(1)	3402(1)	34(1)
C(33B)	3703(3)	1575(2)	4112(1)	36(1)
C(34B)	2932(2)	1060(1)	2205(1)	36(1)
C(35B)	1715(2)	1819(2)	1811(1)	32(1)
C(36B)	2173(2)	3909(2)	1785(1)	35(1)
C(37B)	1890(2)	3563(2)	2485(1)	34(1)
O(1C)	7650(2)	5619(1)	9974(1)	33(1)
C(1C)	8500(3)	6158(2)	10621(1)	47(1)
C(2C)	7669(2)	6212(2)	10264(1)	37(1)
C(3C)	6941(2)	5681(2)	9596(1)	35(1)
C(4C)	6891(3)	5000(2)	9329(2)	45(1)
O(1D)	5000	10000	0	78(2)
C(1D)	5801(5)	9798(4)	751(3)	105(3)
C(2E)	5803(7)	10128(5)	409(4)	70(3)
C(2D)	4961(9)	10361(6)	-312(4)	72(3)





**Table 3. Selected bond lengths [Å] and angles [°] for IAT10 (CCDC 693797).**

Ta(1)-O(2A)	1.9208(18)	Ta(2)-O(2B)	1.9161(16)
Ta(1)-O(1A)	1.9284(17)	Ta(2)-O(1B)	1.9232(15)
Ta(1)-N(2A)	1.9708(19)	Ta(2)-N(3B)	1.977(2)
Ta(1)-N(3A)	1.9847(19)	Ta(2)-N(2B)	1.9778(19)
Ta(1)-N(1A)	2.3921(17)	Ta(2)-N(1B)	2.4017(17)
Ta(1)-Cl(1A)	2.4686(6)	Ta(2)-Cl(1B)	2.4655(6)
O(2A)-Ta(1)-O(1A)	161.09(7)	O(2B)-Ta(2)-O(1B)	161.38(7)
O(2A)-Ta(1)-N(2A)	98.16(8)	O(2B)-Ta(2)-N(3B)	93.02(8)
O(1A)-Ta(1)-N(2A)	99.78(8)	O(1B)-Ta(2)-N(3B)	90.37(8)
O(2A)-Ta(1)-N(3A)	89.07(8)	O(2B)-Ta(2)-N(2B)	99.60(8)
O(1A)-Ta(1)-N(3A)	94.94(8)	O(1B)-Ta(2)-N(2B)	98.15(8)
N(2A)-Ta(1)-N(3A)	95.97(9)	N(3B)-Ta(2)-N(2B)	96.98(9)
O(2A)-Ta(1)-N(1A)	81.05(6)	O(2B)-Ta(2)-N(1B)	81.43(6)
O(1A)-Ta(1)-N(1A)	80.95(6)	O(1B)-Ta(2)-N(1B)	80.76(6)
N(2A)-Ta(1)-N(1A)	179.07(9)	N(3B)-Ta(2)-N(1B)	83.55(7)
N(3A)-Ta(1)-N(1A)	84.53(7)	N(2B)-Ta(2)-N(1B)	178.80(7)
O(2A)-Ta(1)-Cl(1A)	85.46(5)	O(2B)-Ta(2)-Cl(1B)	85.60(5)
O(1A)-Ta(1)-Cl(1A)	86.70(5)	O(1B)-Ta(2)-Cl(1B)	86.41(5)
N(2A)-Ta(1)-Cl(1A)	96.26(7)	N(3B)-Ta(2)-Cl(1B)	165.02(6)
N(3A)-Ta(1)-Cl(1A)	167.19(6)	N(2B)-Ta(2)-Cl(1B)	97.96(7)
N(1A)-Ta(1)-Cl(1A)	83.19(4)	N(1B)-Ta(2)-Cl(1B)	81.49(4)

**Table 4. Bond lengths [Å] and angles [°] for IAT10 (CCDC 693797).**

Ta(1)-O(2A)	1.9208(18)	Ta(2)-N(2B)	1.9778(19)
Ta(1)-O(1A)	1.9284(17)	Ta(2)-N(1B)	2.4017(17)
Ta(1)-N(2A)	1.9708(19)	Ta(2)-Cl(1B)	2.4655(6)
Ta(1)-N(3A)	1.9847(19)	O(1B)-C(1B)	1.343(2)
Ta(1)-N(1A)	2.3921(17)	O(2B)-C(17B)	1.339(3)
Ta(1)-Cl(1A)	2.4686(6)	N(1B)-C(11B)	1.359(3)
O(1A)-C(1A)	1.352(3)	N(1B)-C(7B)	1.364(3)
O(2A)-C(17A)	1.354(3)	N(2B)-C(34B)	1.461(4)
N(1A)-C(7A)	1.364(3)	N(2B)-C(35B)	1.467(4)
N(1A)-C(11A)	1.365(3)	N(3B)-C(36B)	1.453(3)
N(2A)-C(35A)	1.446(3)	N(3B)-C(37B)	1.467(3)
N(2A)-C(34A)	1.470(3)	C(1B)-C(6B)	1.397(3)
N(3A)-C(37A)	1.459(4)	C(1B)-C(2B)	1.413(3)
N(3A)-C(36A)	1.461(4)	C(2B)-C(3B)	1.393(3)
C(1A)-C(6A)	1.385(3)	C(2B)-C(18B)	1.541(3)
C(1A)-C(2A)	1.420(3)	C(3B)-C(4B)	1.395(3)
C(2A)-C(3A)	1.398(3)	C(4B)-C(5B)	1.386(3)
C(2A)-C(18A)	1.525(3)	C(4B)-C(22B)	1.532(3)
C(3A)-C(4A)	1.398(4)	C(5B)-C(6B)	1.398(3)
C(4A)-C(5A)	1.375(3)	C(6B)-C(7B)	1.483(3)
C(4A)-C(22A)	1.528(3)	C(7B)-C(8B)	1.397(3)
C(5A)-C(6A)	1.408(3)	C(8B)-C(9B)	1.379(3)
C(6A)-C(7A)	1.479(3)	C(9B)-C(10B)	1.385(3)
C(7A)-C(8A)	1.396(3)	C(10B)-C(11B)	1.390(3)
C(8A)-C(9A)	1.380(3)	C(11B)-C(12B)	1.482(3)
C(9A)-C(10A)	1.384(3)	C(12B)-C(17B)	1.397(3)
C(10A)-C(11A)	1.395(3)	C(12B)-C(13B)	1.396(4)
C(11A)-C(12A)	1.480(3)	C(13B)-C(14B)	1.387(3)
C(12A)-C(13A)	1.398(3)	C(14B)-C(15B)	1.390(3)
C(12A)-C(17A)	1.403(3)	C(14B)-C(26B)	1.536(4)
C(13A)-C(14A)	1.389(3)	C(15B)-C(16B)	1.396(4)
C(14A)-C(15A)	1.399(4)	C(16B)-C(17B)	1.414(3)
C(14A)-C(26A)	1.539(4)	C(16B)-C(30B)	1.534(3)
C(15A)-C(16A)	1.394(4)	C(18B)-C(19B)	1.536(4)
C(16A)-C(17A)	1.403(3)	C(18B)-C(21B)	1.539(4)
C(16A)-C(30A)	1.541(3)	C(18B)-C(20B)	1.543(4)
C(18A)-C(19A)	1.536(4)	C(22B)-C(23B)	1.529(3)
C(18A)-C(20A)	1.540(4)	C(22B)-C(25B)	1.528(4)
C(18A)-C(21A)	1.547(4)	C(22B)-C(24B)	1.532(4)
C(22A)-C(24A)	1.531(4)	C(26B)-C(27B)	1.520(5)
C(22A)-C(25A)	1.532(5)	C(26B)-C(28B)	1.523(4)
C(22A)-C(23A)	1.537(4)	C(26B)-C(29B)	1.531(6)
C(26A)-C(29A)	1.496(5)	C(30B)-C(33B)	1.524(4)
C(26A)-C(28A)	1.500(4)	C(30B)-C(31B)	1.538(4)
C(26A)-C(27A)	1.590(6)	C(30B)-C(32B)	1.546(4)
C(30A)-C(33A)	1.528(4)	O(1C)-C(2C)	1.410(4)
C(30A)-C(31A)	1.535(5)	O(1C)-C(3C)	1.410(4)
C(30A)-C(32A)	1.547(4)	C(1C)-C(2C)	1.481(5)
Ta(2)-O(2B)	1.9161(16)	C(3C)-C(4C)	1.496(5)
Ta(2)-O(1B)	1.9232(15)	O(1D)-C(2D)#1	1.171(10)
Ta(2)-N(3B)	1.977(2)	O(1D)-C(2D)	1.171(10)

O(1D)-C(2E)#1	1.570(13)	C(10A)-C(11A)-C(12A)	118.31(19)
O(1D)-C(2E)	1.570(13)	C(13A)-C(12A)-C(17A)	119.2(2)
C(1D)-C(2E)	1.226(13)	C(13A)-C(12A)-C(11A)	117.93(19)
C(1D)-C(2D)#1	1.609(14)	C(17A)-C(12A)-C(11A)	122.8(2)
C(2D)-C(1D)#1	1.609(14)	C(14A)-C(13A)-C(12A)	121.6(2)
		C(13A)-C(14A)-C(15A)	117.3(2)
O(2A)-Ta(1)-O(1A)	161.09(7)	C(13A)-C(14A)-C(26A)	120.1(2)
O(2A)-Ta(1)-N(2A)	98.16(8)	C(15A)-C(14A)-C(26A)	122.6(2)
O(1A)-Ta(1)-N(2A)	99.78(8)	C(16A)-C(15A)-C(14A)	123.7(2)
O(2A)-Ta(1)-N(3A)	89.07(8)	C(15A)-C(16A)-C(17A)	117.1(2)
O(1A)-Ta(1)-N(3A)	94.94(8)	C(15A)-C(16A)-C(30A)	121.5(2)
N(2A)-Ta(1)-N(3A)	95.97(9)	C(17A)-C(16A)-C(30A)	121.5(2)
O(2A)-Ta(1)-N(1A)	81.05(6)	O(2A)-C(17A)-C(12A)	117.7(2)
O(1A)-Ta(1)-N(1A)	80.95(6)	O(2A)-C(17A)-C(16A)	121.18(19)
N(2A)-Ta(1)-N(1A)	179.07(9)	C(12A)-C(17A)-C(16A)	121.1(2)
N(3A)-Ta(1)-N(1A)	84.53(7)	C(2A)-C(18A)-C(19A)	108.6(2)
O(2A)-Ta(1)-Cl(1A)	85.46(5)	C(2A)-C(18A)-C(20A)	111.8(2)
O(1A)-Ta(1)-Cl(1A)	86.70(5)	C(19A)-C(18A)-C(20A)	107.8(2)
N(2A)-Ta(1)-Cl(1A)	96.26(7)	C(2A)-C(18A)-C(21A)	111.2(2)
N(3A)-Ta(1)-Cl(1A)	167.19(6)	C(19A)-C(18A)-C(21A)	110.7(2)
N(1A)-Ta(1)-Cl(1A)	83.19(4)	C(20A)-C(18A)-C(21A)	106.7(2)
C(1A)-O(1A)-Ta(1)	141.00(15)	C(4A)-C(22A)-C(24A)	109.0(2)
C(17A)-O(2A)-Ta(1)	137.93(13)	C(4A)-C(22A)-C(25A)	110.3(2)
C(7A)-N(1A)-C(11A)	118.38(17)	C(24A)-C(22A)-C(25A)	111.0(3)
C(7A)-N(1A)-Ta(1)	117.22(14)	C(4A)-C(22A)-C(23A)	112.8(2)
C(11A)-N(1A)-Ta(1)	117.68(13)	C(24A)-C(22A)-C(23A)	107.0(3)
C(35A)-N(2A)-C(34A)	110.8(2)	C(25A)-C(22A)-C(23A)	106.6(3)
C(35A)-N(2A)-Ta(1)	126.50(18)	C(29A)-C(26A)-C(28A)	112.8(3)
C(34A)-N(2A)-Ta(1)	122.52(17)	C(29A)-C(26A)-C(14A)	110.5(3)
C(37A)-N(3A)-C(36A)	109.8(2)	C(28A)-C(26A)-C(14A)	112.9(3)
C(37A)-N(3A)-Ta(1)	128.1(2)	C(29A)-C(26A)-C(27A)	107.9(4)
C(36A)-N(3A)-Ta(1)	121.83(17)	C(28A)-C(26A)-C(27A)	104.7(3)
O(1A)-C(1A)-C(6A)	118.75(19)	C(14A)-C(26A)-C(27A)	107.6(3)
O(1A)-C(1A)-C(2A)	119.82(19)	C(33A)-C(30A)-C(31A)	107.8(3)
C(6A)-C(1A)-C(2A)	121.4(2)	C(33A)-C(30A)-C(16A)	111.8(2)
C(3A)-C(2A)-C(1A)	116.0(2)	C(31A)-C(30A)-C(16A)	110.5(2)
C(3A)-C(2A)-C(18A)	121.4(2)	C(33A)-C(30A)-C(32A)	107.5(2)
C(1A)-C(2A)-C(18A)	122.6(2)	C(31A)-C(30A)-C(32A)	110.4(3)
C(4A)-C(3A)-C(2A)	124.4(2)	C(16A)-C(30A)-C(32A)	108.8(2)
C(5A)-C(4A)-C(3A)	116.9(2)	O(2B)-Ta(2)-O(1B)	161.38(7)
C(5A)-C(4A)-C(22A)	123.2(2)	O(2B)-Ta(2)-N(3B)	93.02(8)
C(3A)-C(4A)-C(22A)	119.9(2)	O(1B)-Ta(2)-N(3B)	90.37(8)
C(4A)-C(5A)-C(6A)	122.1(2)	O(2B)-Ta(2)-N(2B)	99.60(8)
C(1A)-C(6A)-C(5A)	119.1(2)	O(1B)-Ta(2)-N(2B)	98.15(8)
C(1A)-C(6A)-C(7A)	123.80(19)	N(3B)-Ta(2)-N(2B)	96.98(9)
C(5A)-C(6A)-C(7A)	117.03(19)	O(2B)-Ta(2)-N(1B)	81.43(6)
N(1A)-C(7A)-C(8A)	120.76(19)	O(1B)-Ta(2)-N(1B)	80.76(6)
N(1A)-C(7A)-C(6A)	120.49(18)	N(3B)-Ta(2)-N(1B)	83.55(7)
C(8A)-C(7A)-C(6A)	118.49(19)	N(2B)-Ta(2)-N(1B)	178.80(7)
C(9A)-C(8A)-C(7A)	120.2(2)	O(2B)-Ta(2)-Cl(1B)	85.60(5)
C(8A)-C(9A)-C(10A)	118.5(2)	O(1B)-Ta(2)-Cl(1B)	86.41(5)
C(9A)-C(10A)-C(11A)	119.8(2)	N(3B)-Ta(2)-Cl(1B)	165.02(6)
N(1A)-C(11A)-C(10A)	121.1(2)	N(2B)-Ta(2)-Cl(1B)	97.96(7)
N(1A)-C(11A)-C(12A)	120.41(18)	N(1B)-Ta(2)-Cl(1B)	81.49(4)

C(1B)-O(1B)-Ta(2)	139.65(15)	C(13B)-C(14B)-C(26B)	120.4(2)
C(17B)-O(2B)-Ta(2)	140.64(13)	C(15B)-C(14B)-C(26B)	121.9(2)
C(11B)-N(1B)-C(7B)	118.23(17)	C(14B)-C(15B)-C(16B)	123.5(2)
C(11B)-N(1B)-Ta(2)	117.13(12)	C(15B)-C(16B)-C(17B)	116.8(2)
C(7B)-N(1B)-Ta(2)	117.47(14)	C(15B)-C(16B)-C(30B)	121.9(2)
C(34B)-N(2B)-C(35B)	110.1(2)	C(17B)-C(16B)-C(30B)	121.3(2)
C(34B)-N(2B)-Ta(2)	125.3(2)	O(2B)-C(17B)-C(12B)	118.92(19)
C(35B)-N(2B)-Ta(2)	124.52(18)	O(2B)-C(17B)-C(16B)	119.93(19)
C(36B)-N(3B)-C(37B)	110.0(2)	C(12B)-C(17B)-C(16B)	121.1(2)
C(36B)-N(3B)-Ta(2)	126.93(16)	C(19B)-C(18B)-C(21B)	108.1(2)
C(37B)-N(3B)-Ta(2)	122.69(18)	C(19B)-C(18B)-C(2B)	111.4(2)
O(1B)-C(1B)-C(6B)	118.32(18)	C(21B)-C(18B)-C(2B)	108.9(2)
O(1B)-C(1B)-C(2B)	120.37(19)	C(19B)-C(18B)-C(20B)	107.2(3)
C(6B)-C(1B)-C(2B)	121.28(18)	C(21B)-C(18B)-C(20B)	109.7(2)
C(3B)-C(2B)-C(1B)	116.8(2)	C(2B)-C(18B)-C(20B)	111.6(2)
C(3B)-C(2B)-C(18B)	121.54(19)	C(23B)-C(22B)-C(25B)	109.3(2)
C(1B)-C(2B)-C(18B)	121.58(19)	C(23B)-C(22B)-C(4B)	110.41(19)
C(2B)-C(3B)-C(4B)	123.6(2)	C(25B)-C(22B)-C(4B)	108.8(2)
C(5B)-C(4B)-C(3B)	117.50(18)	C(23B)-C(22B)-C(24B)	107.7(2)
C(5B)-C(4B)-C(22B)	122.2(2)	C(25B)-C(22B)-C(24B)	108.8(2)
C(3B)-C(4B)-C(22B)	120.29(19)	C(4B)-C(22B)-C(24B)	111.82(18)
C(4B)-C(5B)-C(6B)	121.8(2)	C(27B)-C(26B)-C(28B)	107.8(3)
C(1B)-C(6B)-C(5B)	118.92(18)	C(27B)-C(26B)-C(29B)	111.1(4)
C(1B)-C(6B)-C(7B)	123.38(17)	C(28B)-C(26B)-C(29B)	106.4(3)
C(5B)-C(6B)-C(7B)	117.60(18)	C(27B)-C(26B)-C(14B)	109.0(3)
N(1B)-C(7B)-C(8B)	121.11(19)	C(28B)-C(26B)-C(14B)	112.3(2)
N(1B)-C(7B)-C(6B)	119.97(16)	C(29B)-C(26B)-C(14B)	110.3(3)
C(8B)-C(7B)-C(6B)	118.67(17)	C(33B)-C(30B)-C(16B)	111.6(2)
C(9B)-C(8B)-C(7B)	119.94(19)	C(33B)-C(30B)-C(31B)	106.9(2)
C(8B)-C(9B)-C(10B)	118.3(2)	C(16B)-C(30B)-C(31B)	112.7(2)
C(9B)-C(10B)-C(11B)	120.0(2)	C(33B)-C(30B)-C(32B)	107.9(2)
N(1B)-C(11B)-C(10B)	121.34(19)	C(16B)-C(30B)-C(32B)	107.9(2)
N(1B)-C(11B)-C(12B)	120.74(17)	C(31B)-C(30B)-C(32B)	109.8(3)
C(10B)-C(11B)-C(12B)	117.81(19)	C(2C)-O(1C)-C(3C)	112.9(2)
C(17B)-C(12B)-C(13B)	119.1(2)	O(1C)-C(2C)-C(1C)	109.6(3)
C(17B)-C(12B)-C(11B)	123.0(2)	O(1C)-C(3C)-C(4C)	110.1(3)
C(13B)-C(12B)-C(11B)	117.95(19)	C(2D)-O(1D)-C(2E)	119.6(7)
C(14B)-C(13B)-C(12B)	121.6(2)	C(1D)-C(2E)-O(1D)	119.9(7)
C(13B)-C(14B)-C(15B)	117.8(2)	O(1D)-C(2D)-C(1D)#1	120.8(9)

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Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y+2,-z

**Table 5. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^4$ ) for IAT10 (CCDC 693797). The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$**

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Ta(1)	128(1)	153(1)	150(1)	-23(1)	10(1)	9(1)
Cl(1A)	179(2)	175(2)	263(3)	10(2)	29(2)	-28(2)
O(1A)	175(7)	161(6)	163(7)	8(5)	-17(5)	24(5)
O(2A)	166(7)	242(7)	148(7)	-29(6)	39(6)	16(6)
N(1A)	121(7)	163(6)	102(7)	6(5)	10(6)	-1(5)
N(2A)	207(10)	213(8)	246(11)	-47(7)	14(8)	31(7)
N(3A)	144(9)	237(8)	232(10)	-44(7)	44(7)	-27(6)
C(1A)	153(9)	179(7)	132(9)	19(6)	10(7)	-8(6)
C(2A)	204(11)	198(8)	179(10)	38(7)	-2(8)	-26(7)
C(3A)	222(11)	283(11)	164(10)	54(8)	11(8)	-44(8)
C(4A)	188(11)	310(10)	137(10)	10(8)	40(8)	-20(8)
C(5A)	170(10)	240(9)	132(9)	-5(7)	18(7)	-7(7)
C(6A)	156(9)	183(8)	116(8)	5(6)	9(7)	-12(6)
C(7A)	122(8)	173(7)	127(8)	1(6)	21(7)	2(6)
C(8A)	213(11)	166(8)	159(10)	-21(7)	-4(8)	15(7)
C(9A)	262(12)	151(8)	201(11)	-1(7)	-31(9)	16(7)
C(10A)	238(11)	166(8)	168(10)	42(7)	-20(8)	12(7)
C(11A)	137(9)	165(7)	111(8)	15(6)	5(7)	8(6)
C(12A)	158(9)	197(8)	107(8)	14(6)	12(7)	-26(7)
C(13A)	185(10)	216(8)	128(9)	32(7)	-1(7)	-19(7)
C(14A)	202(11)	245(9)	159(10)	51(7)	-22(8)	-43(8)
C(15A)	242(12)	305(11)	123(10)	-11(8)	-4(8)	-60(9)
C(16A)	191(10)	273(10)	134(9)	-47(8)	24(8)	-31(8)
C(17A)	152(9)	225(8)	128(9)	-13(7)	37(7)	-20(7)
C(18A)	304(13)	210(9)	190(11)	56(8)	-14(9)	-17(8)
C(19A)	365(16)	223(10)	279(14)	27(9)	-21(11)	-45(10)
C(20A)	472(19)	314(12)	230(14)	106(10)	-32(12)	37(12)
C(21A)	288(14)	262(11)	323(15)	66(10)	13(11)	64(9)
C(22A)	254(13)	413(13)	148(11)	10(9)	73(9)	6(10)
C(23A)	358(16)	374(13)	219(13)	-81(10)	115(12)	-11(11)
C(24A)	570(20)	421(16)	600(30)	67(15)	440(20)	-14(15)
C(25A)	420(20)	870(30)	146(13)	-71(15)	26(13)	199(18)
C(26A)	319(15)	255(11)	323(15)	51(10)	-182(11)	-34(10)
C(27A)	327(19)	650(20)	510(20)	-155(19)	-80(16)	167(16)
C(28A)	291(16)	512(18)	319(17)	-86(14)	-136(12)	34(13)
C(29A)	840(40)	630(30)	1190(50)	680(30)	-660(30)	-370(30)
C(30A)	257(12)	395(12)	202(11)	-122(11)	24(9)	14(11)
C(31A)	344(16)	303(12)	317(16)	-118(11)	10(12)	64(11)
C(32A)	288(15)	630(20)	289(15)	-152(13)	113(13)	29(13)
C(33A)	406(19)	556(19)	255(15)	-221(13)	-6(13)	39(14)
C(34A)	241(13)	347(13)	382(17)	-122(11)	86(12)	50(10)
C(35A)	371(16)	207(10)	441(19)	-61(10)	63(14)	60(10)
C(36A)	175(12)	407(14)	313(15)	10(11)	-11(10)	-68(10)
C(37A)	290(15)	485(17)	345(17)	89(13)	49(13)	-146(13)

Ta(2)	125(1)	143(1)	140(1)	4(1)	31(1)	-28(1)
Cl(1B)	204(3)	179(2)	348(3)	32(2)	46(2)	37(2)
O(1B)	156(7)	186(6)	134(7)	17(5)	30(6)	-26(5)
O(2B)	181(7)	204(6)	142(7)	15(6)	20(5)	-47(6)
N(1B)	151(8)	145(6)	107(7)	10(5)	38(6)	-6(5)
N(2B)	232(10)	246(8)	198(10)	-28(7)	67(8)	-104(7)
N(3B)	145(9)	245(8)	217(10)	32(7)	45(7)	22(6)
C(1B)	146(9)	191(8)	130(9)	18(6)	21(7)	-15(6)
C(2B)	213(11)	212(8)	135(9)	-15(7)	45(8)	-28(7)
C(3B)	205(11)	232(9)	132(9)	-11(7)	51(8)	-15(7)
C(4B)	148(9)	211(8)	123(9)	30(6)	26(7)	-1(7)
C(5B)	152(9)	176(7)	116(8)	28(6)	19(7)	-6(6)
C(6B)	134(8)	161(7)	108(8)	28(6)	17(7)	6(6)
C(7B)	117(8)	144(7)	142(9)	14(6)	29(7)	-3(6)
C(8B)	228(11)	144(7)	180(10)	32(7)	73(8)	-1(7)
C(9B)	318(13)	148(8)	230(12)	-4(7)	113(10)	-24(8)
C(10B)	304(13)	173(8)	164(10)	-30(7)	95(9)	-49(8)
C(11B)	174(9)	157(7)	119(8)	2(6)	41(7)	-36(6)
C(12B)	195(10)	199(8)	125(9)	11(7)	35(8)	-37(7)
C(13B)	228(11)	257(10)	163(10)	13(8)	14(9)	-72(8)
C(14B)	237(12)	303(11)	137(10)	21(8)	9(9)	-23(9)
C(15B)	235(11)	271(10)	144(9)	44(7)	23(8)	-14(8)
C(16B)	211(11)	213(8)	157(10)	48(7)	36(8)	-20(7)
C(17B)	203(10)	180(8)	101(8)	16(6)	38(7)	-22(7)
C(18B)	296(13)	213(10)	203(11)	-43(7)	69(10)	-49(8)
C(19B)	630(20)	367(14)	297(16)	-166(12)	226(16)	-193(15)
C(20B)	301(14)	236(11)	304(14)	-35(9)	-68(11)	-42(9)
C(21B)	295(14)	219(10)	291(14)	-38(9)	20(11)	9(9)
C(22B)	162(10)	251(9)	130(9)	36(7)	40(7)	-11(7)
C(23B)	225(13)	375(13)	418(18)	51(12)	151(13)	54(10)
C(24B)	266(14)	429(14)	226(13)	-2(10)	81(11)	-151(11)
C(25B)	277(13)	349(12)	171(11)	99(9)	33(10)	-13(10)
C(26B)	281(14)	448(15)	155(11)	13(10)	-21(10)	-105(11)
C(27B)	300(20)	1700(60)	370(20)	-220(30)	-11(17)	-200(30)
C(28B)	420(20)	690(20)	178(14)	35(14)	-43(12)	-106(16)
C(29B)	1390(50)	500(20)	320(20)	8(17)	-320(30)	-390(30)
C(30B)	310(14)	247(10)	206(12)	87(8)	48(10)	-44(9)
C(31B)	284(14)	400(14)	350(16)	136(12)	66(12)	-103(11)
C(32B)	490(19)	208(10)	307(15)	57(9)	74(13)	-9(11)
C(33B)	466(19)	377(14)	242(14)	121(11)	59(13)	-127(13)
C(34B)	393(17)	212(10)	500(20)	-37(11)	104(15)	-107(11)
C(35B)	280(14)	378(14)	280(14)	-41(11)	6(11)	-137(11)
C(36B)	271(14)	480(16)	305(16)	162(12)	61(12)	177(12)
C(37B)	271(14)	444(15)	341(16)	73(12)	153(13)	129(12)
O(1C)	242(10)	313(9)	434(14)	68(8)	82(9)	37(7)
C(1C)	364(19)	630(20)	430(20)	120(17)	104(16)	-101(16)
C(2C)	331(17)	372(14)	425(19)	67(13)	103(14)	1(12)
C(3C)	271(15)	345(13)	428(19)	100(12)	79(13)	38(11)
C(4C)	390(20)	400(16)	570(30)	-63(15)	88(18)	14(14)
O(1D)	530(30)	550(30)	1270(60)	0(30)	210(30)	-190(20)

C(1D)	640(40)	1050(50)	1450(80)	-310(50)	210(50)	-70(40)
C(2E)	660(70)	450(40)	1120(100)	20(50)	460(70)	-170(40)
C(2D)	1100(100)	630(50)	540(60)	90(50)	380(70)	-80(60)

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**Crystal Structure Analysis of:**  
**Complex 4 (IAT16)**

Contents

Table 1. Crystal data

Figures Minimum overlap

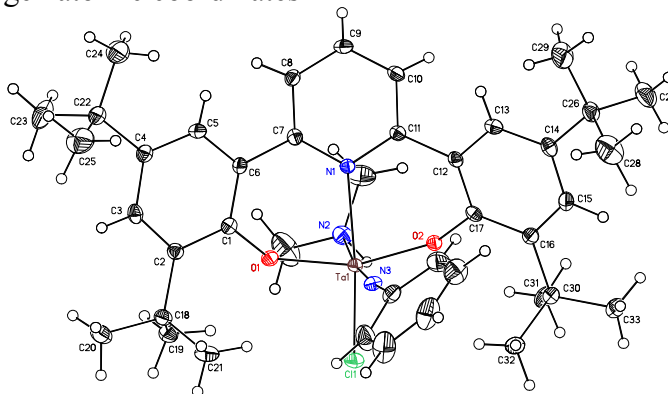
Table 2. Atomic Coordinates

Table 3. Selected bond distances and angles

Table 4. Full bond distances and angles

Table 5. Anisotropic displacement parameters

Table 6. Hydrogen atomic coordinates



IAT16

**Note:** The crystallographic data have been deposited in the Cambridge Database (CCDC) and has been placed on hold pending further instructions from me. The deposition number is 694413. Ideally the CCDC would like the publication to contain a footnote of the type: "Crystallographic data have been deposited at the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 694413."



**Table 1. Crystal data and structure refinement for IAT16 (CCDC 694413).**

Empirical formula	C <sub>41</sub> H <sub>55</sub> N <sub>3</sub> O <sub>2</sub> ClTa
Formula weight	838.28
Crystallization Solvent	Benzene
Crystal Habit	Block
Crystal size	0.09 x 0.09 x 0.06 mm <sup>3</sup>
Crystal color	Yellow

### Data Collection

Type of diffractometer	Bruker KAPPA APEX II	
Wavelength	0.71073 Å MoK $\alpha$	
Data Collection Temperature	100(2) K	
$\theta$ range for 9137 reflections used in lattice determination	2.18 to 34.73°	
Unit cell dimensions	a = 9.9628(5) Å b = 10.6118(5) Å c = 18.9240(9) Å	$\alpha$ = 80.804(3)° $\beta$ = 89.355(3)° $\gamma$ = 81.578(3)°
Volume	1953.54(16) Å <sup>3</sup>	
Z	2	
Crystal system	Triclinic	
Space group	P-1	
Density (calculated)	1.425 Mg/m <sup>3</sup>	
F(000)	856	
Data collection program	Bruker APEX2 v2.1-0	
$\theta$ range for data collection	1.97 to 35.14°	
Completeness to $\theta = 35.14^\circ$	90.5 %	
Index ranges	$-15 \leq h \leq 14$ , $-17 \leq k \leq 17$ , $-29 \leq l \leq 30$	
Data collection scan type	$\omega$ scans; 15 settings	
Data reduction program	Bruker SAINT-Plus v7.34A	
Reflections collected	58512	
Independent reflections	15730 [ $R_{\text{int}} = 0.0505$ ]	
Absorption coefficient	2.919 mm <sup>-1</sup>	
Absorption correction	None	
Max. and min. transmission	0.8444 and 0.7791	

**Table 1 (cont.)****Structure solution and Refinement**

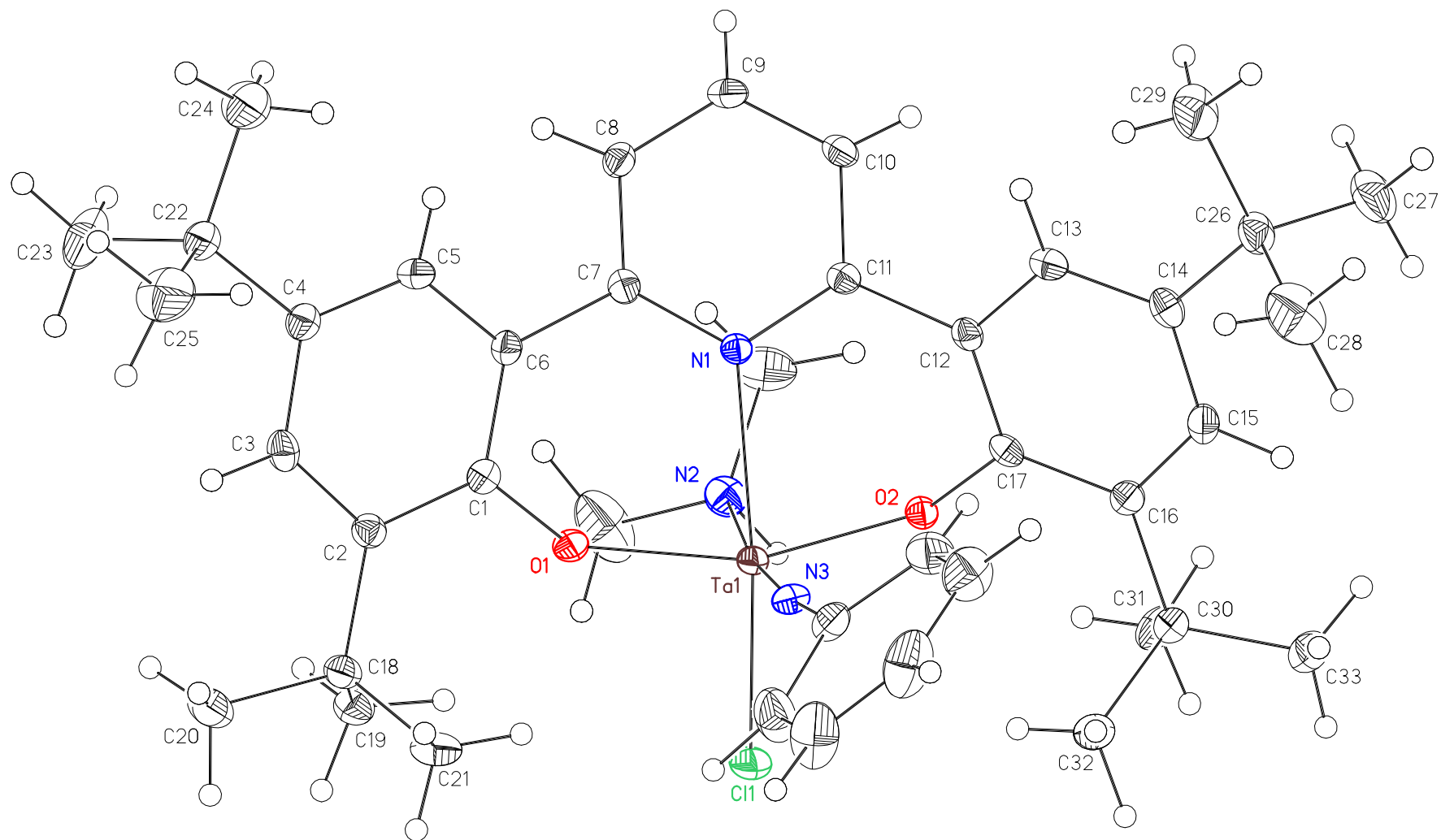
Structure solution program	SHELXS-97 (Sheldrick, 2008)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Difference Fourier map
Structure refinement program	SHELXL-97 (Sheldrick, 2008)
Refinement method	Full matrix least-squares on $F^2$
Data / restraints / parameters	15730 / 0 / 653
Treatment of hydrogen atoms	Unrestrained
Goodness-of-fit on $F^2$	1.281
Final R indices [ $I > 2\sigma(I)$ , 12866 reflections]	$R1 = 0.0326$ , $wR2 = 0.0452$
R indices (all data)	$R1 = 0.0473$ , $wR2 = 0.0462$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.005
Average shift/error	0.000
Largest diff. peak and hole	3.936 and -2.313 e.Å <sup>-3</sup>

**Special Refinement Details**

Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

Refinement of  $F^2$  against ALL reflections. The weighted R-factor ( $wR$ ) and goodness of fit ( $S$ ) are based on  $F^2$ , conventional R-factors ( $R$ ) are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.



**Table 2. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for IAT16 (CCDC 694413).  $U(\text{eq})$  is defined as the trace of the orthogonalized  $U^{\text{ij}}$  tensor.**

	x	y	z	$U_{\text{eq}}$
Ta(1)	7043(1)	5812(1)	7129(1)	12(1)
Cl(1)	4834(1)	6355(1)	6585(1)	22(1)
O(1)	6600(1)	6836(1)	7887(1)	15(1)
O(2)	8002(1)	5393(1)	6265(1)	13(1)
N(1)	9195(2)	5883(1)	7519(1)	11(1)
N(2)	7282(2)	7912(2)	6415(1)	19(1)
N(3)	7006(2)	4203(2)	7574(1)	15(1)
C(1)	7100(2)	6937(2)	8525(1)	13(1)
C(2)	6231(2)	7166(2)	9105(1)	14(1)
C(3)	6844(2)	7293(2)	9738(1)	16(1)
C(4)	8248(2)	7200(2)	9841(1)	15(1)
C(5)	9048(2)	6980(2)	9258(1)	16(1)
C(6)	8508(2)	6859(2)	8600(1)	13(1)
C(7)	9455(2)	6638(2)	8008(1)	13(1)
C(8)	10672(2)	7145(2)	7991(1)	14(1)
C(9)	11680(2)	6782(2)	7532(1)	16(1)
C(10)	11495(2)	5878(2)	7118(1)	15(1)
C(11)	10257(2)	5424(2)	7119(1)	12(1)
C(12)	10140(2)	4385(2)	6699(1)	12(1)
C(13)	11231(2)	3378(2)	6732(1)	14(1)
C(14)	11219(2)	2391(2)	6342(1)	15(1)
C(15)	10086(2)	2449(2)	5898(1)	16(1)
C(16)	8972(2)	3421(2)	5838(1)	14(1)
C(17)	9020(2)	4388(2)	6263(1)	12(1)
C(18)	4685(2)	7272(2)	9021(1)	17(1)
C(19)	4162(2)	8330(2)	8395(1)	20(1)
C(20)	3948(2)	7620(3)	9690(1)	28(1)
C(21)	4322(2)	5966(2)	8896(1)	22(1)
C(22)	8829(2)	7380(2)	10555(1)	21(1)
C(23)	8341(3)	8760(3)	10686(2)	39(1)
C(24)	10381(3)	7146(4)	10570(2)	42(1)
C(25)	8346(3)	6433(3)	11162(2)	36(1)
C(26)	12398(2)	1272(2)	6381(1)	19(1)
C(27)	13047(3)	1284(3)	5647(2)	31(1)
C(28)	11870(3)	-9(2)	6603(2)	38(1)
C(29)	13492(3)	1397(3)	6913(2)	35(1)
C(30)	7748(2)	3415(2)	5358(1)	16(1)
C(31)	7468(3)	4699(2)	4843(1)	23(1)
C(32)	6496(2)	3211(2)	5812(1)	23(1)
C(33)	7976(3)	2343(2)	4903(1)	25(1)
C(34)	6497(3)	9027(3)	6655(2)	44(1)
C(35)	8658(3)	8176(3)	6214(2)	30(1)
C(36)	7087(2)	2919(2)	7876(1)	18(1)
C(37)	6059(3)	2497(2)	8313(1)	31(1)
C(38)	6162(3)	1211(2)	8623(2)	40(1)
C(39)	7270(3)	349(2)	8510(1)	35(1)
C(40)	8275(3)	736(2)	8063(1)	30(1)
C(41)	8186(2)	2009(2)	7744(1)	24(1)

**Table 3. Selected bond lengths [Å] and angles [°] for IAT16 (CCDC 694413).**

Ta(1)-N(3)	1.7840(15)	N(3)-Ta(1)-O(1)	102.94(7)
Ta(1)-O(1)	1.9438(14)	N(3)-Ta(1)-O(2)	98.12(7)
Ta(1)-O(2)	1.9643(13)	O(1)-Ta(1)-O(2)	154.01(5)
Ta(1)-N(1)	2.2918(16)	N(3)-Ta(1)-N(1)	94.12(6)
Ta(1)-Cl(1)	2.3920(5)	O(1)-Ta(1)-N(1)	80.82(6)
Ta(1)-N(2)	2.4569(17)	O(2)-Ta(1)-N(1)	82.66(6)
		N(3)-Ta(1)-Cl(1)	102.00(5)
		O(1)-Ta(1)-Cl(1)	94.23(4)
		O(2)-Ta(1)-Cl(1)	96.12(4)
		N(1)-Ta(1)-Cl(1)	163.84(4)
		N(3)-Ta(1)-N(2)	173.15(7)
		O(1)-Ta(1)-N(2)	83.24(6)
		O(2)-Ta(1)-N(2)	75.15(6)
		N(1)-Ta(1)-N(2)	83.86(6)
		Cl(1)-Ta(1)-N(2)	80.26(5)

**Table 4. Bond lengths [Å] and angles [°] for IAT16 (CCDC 694413).**

Ta(1)-N(3)	1.7840(15)	C(21)-H(21A)	0.85(2)
Ta(1)-O(1)	1.9438(14)	C(21)-H(21B)	0.96(2)
Ta(1)-O(2)	1.9643(13)	C(21)-H(21C)	0.88(2)
Ta(1)-N(1)	2.2918(16)	C(22)-C(25)	1.524(3)
Ta(1)-Cl(1)	2.3920(5)	C(22)-C(24)	1.530(3)
Ta(1)-N(2)	2.4569(17)	C(22)-C(23)	1.532(3)
O(1)-C(1)	1.336(2)	C(23)-H(23A)	0.97(3)
O(2)-C(17)	1.360(2)	C(23)-H(23B)	0.98(3)
N(1)-C(7)	1.367(2)	C(23)-H(23C)	0.98(3)
N(1)-C(11)	1.371(2)	C(24)-H(24A)	1.01(3)
N(2)-C(34)	1.453(3)	C(24)-H(24B)	0.87(3)
N(2)-C(35)	1.472(3)	C(24)-H(24C)	1.07(3)
N(2)-H(2)	0.91(2)	C(25)-H(25A)	0.97(3)
N(3)-C(36)	1.382(2)	C(25)-H(25B)	1.00(3)
C(1)-C(6)	1.400(3)	C(25)-H(25C)	0.97(3)
C(1)-C(2)	1.420(3)	C(26)-C(29)	1.525(3)
C(2)-C(3)	1.386(3)	C(26)-C(27)	1.526(3)
C(2)-C(18)	1.536(3)	C(26)-C(28)	1.524(3)
C(3)-C(4)	1.401(3)	C(27)-H(27A)	1.04(3)
C(3)-H(3)	0.88(2)	C(27)-H(27B)	0.92(3)
C(4)-C(5)	1.383(3)	C(27)-H(27C)	0.86(2)
C(4)-C(22)	1.527(3)	C(28)-H(28A)	1.08(3)
C(5)-C(6)	1.395(3)	C(28)-H(28B)	0.91(3)
C(5)-H(5)	0.83(2)	C(28)-H(28C)	0.94(3)
C(6)-C(7)	1.481(3)	C(29)-H(29A)	0.89(2)
C(7)-C(8)	1.395(3)	C(29)-H(29B)	0.92(3)
C(8)-C(9)	1.373(3)	C(29)-H(29C)	1.03(2)
C(8)-H(8)	0.88(2)	C(30)-C(33)	1.523(3)
C(9)-C(10)	1.364(3)	C(30)-C(32)	1.529(3)
C(9)-H(9)	0.80(2)	C(30)-C(31)	1.535(3)
C(10)-C(11)	1.388(3)	C(31)-H(31A)	0.98(3)
C(10)-H(10)	0.83(2)	C(31)-H(31B)	0.85(2)
C(11)-C(12)	1.476(3)	C(31)-H(31C)	0.92(2)
C(12)-C(17)	1.395(3)	C(32)-H(32A)	1.02(2)
C(12)-C(13)	1.403(3)	C(32)-H(32B)	0.83(2)
C(13)-C(14)	1.377(3)	C(32)-H(32C)	0.95(2)
C(13)-H(13)	0.91(2)	C(33)-H(33A)	0.95(2)
C(14)-C(15)	1.403(3)	C(33)-H(33B)	0.90(2)
C(14)-C(26)	1.536(3)	C(33)-H(33C)	0.87(2)
C(15)-C(16)	1.393(3)	C(34)-H(34A)	1.09(2)
C(15)-H(15)	0.923(19)	C(34)-H(34B)	0.99(3)
C(16)-C(17)	1.408(3)	C(34)-H(34C)	0.88(3)
C(16)-C(30)	1.530(3)	C(35)-H(35A)	0.97(2)
C(18)-C(20)	1.524(3)	C(35)-H(35B)	1.01(3)
C(18)-C(19)	1.535(3)	C(35)-H(35C)	0.94(3)
C(18)-C(21)	1.536(3)	C(36)-C(37)	1.389(3)
C(19)-H(19A)	0.95(2)	C(36)-C(41)	1.398(3)
C(19)-H(19B)	1.00(2)	C(37)-C(38)	1.387(3)
C(19)-H(19C)	0.82(3)	C(37)-H(37)	0.99(2)
C(20)-H(20A)	0.92(2)	C(38)-C(39)	1.364(4)
C(20)-H(20B)	0.95(2)	C(38)-H(38)	0.87(3)
C(20)-H(20C)	0.91(2)	C(39)-C(40)	1.370(4)

C(39)-H(39)	0.83(2)	C(9)-C(8)-H(8)	125.4(13)
C(40)-C(41)	1.380(3)	C(7)-C(8)-H(8)	114.2(13)
C(40)-H(40)	0.87(2)	C(10)-C(9)-C(8)	119.0(2)
C(41)-H(41)	0.83(2)	C(10)-C(9)-H(9)	120.9(16)
		C(8)-C(9)-H(9)	120.1(16)
N(3)-Ta(1)-O(1)	102.94(7)	C(9)-C(10)-C(11)	119.9(2)
N(3)-Ta(1)-O(2)	98.12(7)	C(9)-C(10)-H(10)	124.7(15)
O(1)-Ta(1)-O(2)	154.01(5)	C(11)-C(10)-H(10)	115.2(15)
N(3)-Ta(1)-N(1)	94.12(6)	N(1)-C(11)-C(10)	121.29(19)
O(1)-Ta(1)-N(1)	80.82(6)	N(1)-C(11)-C(12)	120.91(17)
O(2)-Ta(1)-N(1)	82.66(6)	C(10)-C(11)-C(12)	117.74(18)
N(3)-Ta(1)-Cl(1)	102.00(5)	C(17)-C(12)-C(13)	119.46(19)
O(1)-Ta(1)-Cl(1)	94.23(4)	C(17)-C(12)-C(11)	122.92(17)
O(2)-Ta(1)-Cl(1)	96.12(4)	C(13)-C(12)-C(11)	117.60(18)
N(1)-Ta(1)-Cl(1)	163.84(4)	C(14)-C(13)-C(12)	121.26(19)
N(3)-Ta(1)-N(2)	173.15(7)	C(14)-C(13)-H(13)	120.5(13)
O(1)-Ta(1)-N(2)	83.24(6)	C(12)-C(13)-H(13)	118.2(13)
O(2)-Ta(1)-N(2)	75.15(6)	C(13)-C(14)-C(15)	117.32(18)
N(1)-Ta(1)-N(2)	83.86(6)	C(13)-C(14)-C(26)	121.75(19)
Cl(1)-Ta(1)-N(2)	80.26(5)	C(15)-C(14)-C(26)	120.93(19)
C(1)-O(1)-Ta(1)	138.20(12)	C(16)-C(15)-C(14)	124.4(2)
C(17)-O(2)-Ta(1)	124.00(11)	C(16)-C(15)-H(15)	114.1(12)
C(7)-N(1)-C(11)	117.76(17)	C(14)-C(15)-H(15)	121.5(12)
C(7)-N(1)-Ta(1)	122.51(12)	C(15)-C(16)-C(17)	115.99(19)
C(11)-N(1)-Ta(1)	117.58(13)	C(15)-C(16)-C(30)	122.00(18)
C(34)-N(2)-C(35)	110.5(2)	C(17)-C(16)-C(30)	121.98(17)
C(34)-N(2)-Ta(1)	115.65(16)	O(2)-C(17)-C(12)	115.83(18)
C(35)-N(2)-Ta(1)	118.09(14)	O(2)-C(17)-C(16)	122.59(18)
C(34)-N(2)-H(2)	102.3(16)	C(12)-C(17)-C(16)	121.57(17)
C(35)-N(2)-H(2)	108.8(15)	C(20)-C(18)-C(19)	107.24(18)
Ta(1)-N(2)-H(2)	99.4(14)	C(20)-C(18)-C(2)	111.79(18)
C(36)-N(3)-Ta(1)	174.17(15)	C(19)-C(18)-C(2)	110.49(17)
O(1)-C(1)-C(6)	118.15(17)	C(20)-C(18)-C(21)	107.74(19)
O(1)-C(1)-C(2)	121.24(18)	C(19)-C(18)-C(21)	110.21(19)
C(6)-C(1)-C(2)	120.56(19)	C(2)-C(18)-C(21)	109.32(17)
C(3)-C(2)-C(1)	116.91(19)	C(18)-C(19)-H(19A)	105.9(13)
C(3)-C(2)-C(18)	122.35(18)	C(18)-C(19)-H(19B)	109.2(11)
C(1)-C(2)-C(18)	120.74(18)	H(19A)-C(19)-H(19B)	113.1(19)
C(2)-C(3)-C(4)	124.4(2)	C(18)-C(19)-H(19C)	111.1(18)
C(2)-C(3)-H(3)	120.1(15)	H(19A)-C(19)-H(19C)	101(2)
C(4)-C(3)-H(3)	115.5(15)	H(19B)-C(19)-H(19C)	116(2)
C(5)-C(4)-C(3)	116.33(19)	C(18)-C(20)-H(20A)	111.8(13)
C(5)-C(4)-C(22)	122.98(19)	C(18)-C(20)-H(20B)	100.4(15)
C(3)-C(4)-C(22)	120.66(18)	H(20A)-C(20)-H(20B)	112.4(19)
C(4)-C(5)-C(6)	122.71(19)	C(18)-C(20)-H(20C)	112.3(14)
C(4)-C(5)-H(5)	120.0(15)	H(20A)-C(20)-H(20C)	110(2)
C(6)-C(5)-H(5)	117.2(15)	H(20B)-C(20)-H(20C)	109.2(19)
C(5)-C(6)-C(1)	119.04(18)	C(18)-C(21)-H(21A)	109.1(16)
C(5)-C(6)-C(7)	118.36(18)	C(18)-C(21)-H(21B)	111.5(12)
C(1)-C(6)-C(7)	122.60(18)	H(21A)-C(21)-H(21B)	112.5(19)
N(1)-C(7)-C(8)	120.40(18)	C(18)-C(21)-H(21C)	109.4(15)
N(1)-C(7)-C(6)	121.23(17)	H(21A)-C(21)-H(21C)	102.2(19)
C(8)-C(7)-C(6)	118.21(18)	H(21B)-C(21)-H(21C)	111.8(19)
C(9)-C(8)-C(7)	120.2(2)	C(25)-C(22)-C(24)	107.5(2)

C(25)-C(22)-C(4)	110.18(19)	C(16)-C(30)-C(31)	109.77(17)
C(24)-C(22)-C(4)	112.11(19)	C(32)-C(30)-C(31)	109.77(19)
C(25)-C(22)-C(23)	109.0(2)	C(30)-C(31)-H(31A)	110.2(14)
C(24)-C(22)-C(23)	108.9(2)	C(30)-C(31)-H(31B)	113.0(15)
C(4)-C(22)-C(23)	109.08(19)	H(31A)-C(31)-H(31B)	104(2)
C(22)-C(23)-H(23A)	109.1(16)	C(30)-C(31)-H(31C)	106.7(14)
C(22)-C(23)-H(23B)	108.7(17)	H(31A)-C(31)-H(31C)	111(2)
H(23A)-C(23)-H(23B)	104(2)	H(31B)-C(31)-H(31C)	112(2)
C(22)-C(23)-H(23C)	106.4(16)	C(30)-C(32)-H(32A)	109.8(13)
H(23A)-C(23)-H(23C)	118(2)	C(30)-C(32)-H(32B)	115.3(15)
H(23B)-C(23)-H(23C)	111(2)	H(32A)-C(32)-H(32B)	107.2(19)
C(22)-C(24)-H(24A)	104.9(16)	C(30)-C(32)-H(32C)	112.8(14)
C(22)-C(24)-H(24B)	103(2)	H(32A)-C(32)-H(32C)	110.5(19)
H(24A)-C(24)-H(24B)	108(3)	H(32B)-C(32)-H(32C)	101(2)
C(22)-C(24)-H(24C)	102.8(16)	C(30)-C(33)-H(33A)	110.8(14)
H(24A)-C(24)-H(24C)	116(2)	C(30)-C(33)-H(33B)	106.0(14)
H(24B)-C(24)-H(24C)	120(3)	H(33A)-C(33)-H(33B)	111.9(19)
C(22)-C(25)-H(25A)	108.2(16)	C(30)-C(33)-H(33C)	108.3(15)
C(22)-C(25)-H(25B)	108.6(16)	H(33A)-C(33)-H(33C)	104.8(18)
H(25A)-C(25)-H(25B)	104(2)	H(33B)-C(33)-H(33C)	115(2)
C(22)-C(25)-H(25C)	107.2(17)	N(2)-C(34)-H(34A)	103.6(13)
H(25A)-C(25)-H(25C)	117(2)	N(2)-C(34)-H(34B)	106.5(18)
H(25B)-C(25)-H(25C)	112(2)	H(34A)-C(34)-H(34B)	114(2)
C(29)-C(26)-C(27)	107.9(2)	N(2)-C(34)-H(34C)	107.6(18)
C(29)-C(26)-C(28)	109.3(2)	H(34A)-C(34)-H(34C)	112(2)
C(27)-C(26)-C(28)	109.0(2)	H(34B)-C(34)-H(34C)	113(2)
C(29)-C(26)-C(14)	111.48(19)	N(2)-C(35)-H(35A)	112.0(13)
C(27)-C(26)-C(14)	109.34(17)	N(2)-C(35)-H(35B)	110.2(15)
C(28)-C(26)-C(14)	109.81(19)	H(35A)-C(35)-H(35B)	107.2(19)
C(26)-C(27)-H(27A)	106.2(15)	N(2)-C(35)-H(35C)	106.8(15)
C(26)-C(27)-H(27B)	111.2(15)	H(35A)-C(35)-H(35C)	107(2)
H(27A)-C(27)-H(27B)	111(2)	H(35B)-C(35)-H(35C)	113(2)
C(26)-C(27)-H(27C)	105.5(17)	N(3)-C(36)-C(37)	120.4(2)
H(27A)-C(27)-H(27C)	112(2)	N(3)-C(36)-C(41)	121.30(19)
H(27B)-C(27)-H(27C)	111(2)	C(37)-C(36)-C(41)	118.3(2)
C(26)-C(28)-H(28A)	113.5(14)	C(36)-C(37)-C(38)	119.9(2)
C(26)-C(28)-H(28B)	106(2)	C(36)-C(37)-H(37)	117.2(14)
H(28A)-C(28)-H(28B)	115(2)	C(38)-C(37)-H(37)	122.6(14)
C(26)-C(28)-H(28C)	109.3(16)	C(39)-C(38)-C(37)	120.8(3)
H(28A)-C(28)-H(28C)	106(2)	C(39)-C(38)-H(38)	119.3(18)
H(28B)-C(28)-H(28C)	107(2)	C(37)-C(38)-H(38)	119.7(18)
C(26)-C(29)-H(29A)	107.2(17)	C(38)-C(39)-C(40)	120.2(2)
C(26)-C(29)-H(29B)	109.3(17)	C(38)-C(39)-H(39)	122.1(17)
H(29A)-C(29)-H(29B)	111(2)	C(40)-C(39)-H(39)	117.7(17)
C(26)-C(29)-H(29C)	105.6(12)	C(39)-C(40)-C(41)	119.9(2)
H(29A)-C(29)-H(29C)	106.8(19)	C(39)-C(40)-H(40)	122.6(14)
H(29B)-C(29)-H(29C)	117(2)	C(41)-C(40)-H(40)	117.5(14)
C(33)-C(30)-C(16)	112.48(17)	C(40)-C(41)-C(36)	120.8(2)
C(33)-C(30)-C(32)	107.20(18)	C(40)-C(41)-H(41)	123.1(15)
C(16)-C(30)-C(32)	110.35(18)	C(36)-C(41)-H(41)	115.9(15)
C(33)-C(30)-C(31)	107.18(19)		

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**Table 5. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^4$ ) for IAT16 (CCDC 694413). The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$**

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Ta(1)	92(1)	160(1)	121(1)	-38(1)	-6(1)	-13(1)
Cl(1)	117(2)	321(3)	217(3)	-84(2)	-34(2)	-10(2)
O(1)	121(7)	202(7)	118(7)	-31(6)	-36(6)	-3(5)
O(2)	110(7)	148(6)	116(7)	-24(5)	-18(5)	10(5)
N(1)	102(8)	116(7)	107(8)	-13(6)	-12(6)	0(6)
N(2)	177(9)	185(9)	193(10)	-10(7)	-10(8)	1(7)
N(3)	124(8)	164(8)	170(9)	-30(7)	-20(7)	-43(6)
C(1)	148(10)	110(8)	125(10)	-8(7)	-5(8)	-20(7)
C(2)	136(10)	135(9)	150(11)	-21(8)	4(8)	-16(7)
C(3)	194(11)	163(9)	121(11)	-24(8)	45(9)	-26(8)
C(4)	166(10)	156(9)	139(10)	-27(8)	-8(8)	-22(8)
C(5)	114(10)	179(10)	182(11)	-37(8)	-10(9)	-30(8)
C(6)	143(10)	115(8)	126(10)	-28(7)	8(8)	-18(7)
C(7)	129(10)	137(9)	99(10)	-6(7)	-15(8)	10(7)
C(8)	153(10)	150(9)	130(11)	-36(8)	-11(8)	-23(8)
C(9)	115(10)	197(10)	163(11)	-14(8)	-15(8)	-45(8)
C(10)	105(9)	195(10)	141(11)	-19(8)	13(8)	-11(8)
C(11)	112(9)	142(9)	104(10)	-2(7)	-9(8)	-10(7)
C(12)	125(9)	136(9)	106(10)	-10(7)	32(8)	-30(7)
C(13)	116(10)	177(9)	123(10)	11(8)	-2(8)	-23(8)
C(14)	166(10)	140(9)	140(10)	2(8)	38(8)	-2(8)
C(15)	192(11)	145(9)	151(11)	-46(8)	35(9)	-35(8)
C(16)	153(10)	153(9)	120(10)	-14(8)	23(8)	-54(8)
C(17)	114(9)	131(9)	114(10)	12(7)	32(8)	-19(7)
C(18)	124(10)	217(10)	167(11)	-37(8)	3(8)	-20(8)
C(19)	139(11)	205(11)	267(14)	-68(10)	-19(10)	11(9)
C(20)	162(12)	439(16)	226(14)	-85(12)	15(10)	-14(11)
C(21)	152(12)	258(12)	259(14)	9(10)	-13(10)	-69(9)
C(22)	190(11)	283(11)	166(11)	-82(9)	-2(9)	-48(9)
C(23)	493(19)	378(15)	337(17)	-190(13)	-54(15)	-70(14)
C(24)	269(15)	800(20)	246(16)	-239(16)	-38(12)	-63(15)
C(25)	372(17)	495(18)	216(15)	0(13)	-67(12)	-77(14)
C(26)	182(11)	180(10)	208(12)	-55(8)	-6(9)	36(8)
C(27)	240(14)	351(14)	306(16)	-77(12)	16(12)	93(11)
C(28)	378(17)	194(12)	520(20)	17(12)	41(15)	54(11)
C(29)	290(15)	333(14)	385(18)	-135(13)	-96(13)	162(12)
C(30)	156(10)	175(9)	165(11)	-52(8)	-4(8)	-29(8)
C(31)	270(13)	259(12)	170(12)	-46(10)	-75(10)	-43(10)
C(32)	176(12)	304(13)	247(13)	-105(11)	24(10)	-91(10)
C(33)	215(13)	278(13)	266(14)	-120(11)	-51(11)	-14(10)
C(34)	490(20)	185(12)	590(20)	26(13)	198(17)	49(12)
C(35)	239(13)	300(13)	342(16)	71(12)	-20(12)	-75(11)
C(36)	215(11)	194(10)	154(11)	-40(8)	-4(9)	-76(9)
C(37)	350(15)	266(12)	339(15)	-107(11)	160(12)	-106(11)
C(38)	600(20)	290(13)	361(17)	-91(12)	240(15)	-254(13)
C(39)	610(20)	164(11)	299(15)	-16(10)	38(13)	-114(12)
C(40)	352(15)	178(11)	350(15)	-27(10)	-14(12)	-20(10)

C(41)	226(12)	244(11)	248(13)	-14(10)	16(10)	-70(10)
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**Table 6. Hydrogen coordinates (  $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for IAT16 (CCDC 694413).**

	x	y	z	$U_{\text{iso}}$
H(2)	6820(20)	7850(20)	6013(13)	34(7)
H(3)	6350(20)	7425(19)	10112(12)	21(6)
H(5)	9890(20)	6895(18)	9295(11)	11(5)
H(8)	10735(19)	7650(17)	8310(10)	5(5)
H(9)	12370(20)	7090(19)	7514(11)	17(6)
H(10)	12060(20)	5601(18)	6832(11)	10(5)
H(13)	11940(20)	3381(18)	7029(11)	12(5)
H(15)	10027(18)	1829(17)	5613(10)	4(5)
H(19A)	3200(20)	8450(20)	8429(12)	33(7)
H(19B)	4490(20)	8051(19)	7937(12)	17(6)
H(19C)	4320(30)	9030(20)	8462(14)	40(8)
H(20A)	4180(20)	6997(19)	10083(11)	14(6)
H(20B)	3030(30)	7670(20)	9541(13)	35(7)
H(20C)	4090(20)	8410(20)	9783(12)	21(6)
H(21A)	4560(20)	5400(20)	9265(13)	27(7)
H(21B)	4720(20)	5719(18)	8465(11)	11(5)
H(21C)	3430(20)	6000(20)	8892(12)	26(7)
H(23A)	8720(30)	9360(20)	10324(15)	46(8)
H(23B)	8750(30)	8880(30)	11133(17)	65(10)
H(23C)	7350(30)	8840(20)	10721(14)	50(9)
H(24A)	10640(30)	6230(30)	10481(15)	52(9)
H(24B)	10570(30)	7160(30)	11013(17)	70(11)
H(24C)	10630(30)	7890(30)	10160(16)	68(10)
H(25A)	7370(30)	6660(30)	11204(15)	54(9)
H(25B)	8720(30)	6590(30)	11623(15)	57(9)
H(25C)	8680(30)	5570(30)	11067(16)	64(10)
H(27A)	13460(30)	2140(20)	5537(13)	46(8)
H(27B)	12410(20)	1250(20)	5301(13)	32(7)
H(27C)	13660(30)	620(20)	5692(13)	41(8)
H(28A)	11110(30)	-170(20)	6243(15)	57(9)
H(28B)	11600(30)	-10(30)	7063(17)	64(11)
H(28C)	12590(30)	-690(20)	6602(14)	48(8)
H(29A)	14130(30)	720(20)	6916(13)	38(7)
H(29B)	13120(30)	1400(20)	7357(14)	36(8)
H(29C)	13910(20)	2200(20)	6692(11)	20(6)
H(31A)	8280(30)	4860(20)	4570(13)	39(8)
H(31B)	7290(20)	5350(20)	5056(12)	21(6)
H(31C)	6760(20)	4630(20)	4545(13)	27(7)
H(32A)	5690(20)	3190(20)	5490(12)	30(7)
H(32B)	6260(20)	3760(20)	6069(12)	19(6)
H(32C)	6640(20)	2460(20)	6160(12)	28(7)
H(33A)	8760(20)	2413(19)	4614(12)	22(6)
H(33B)	7220(20)	2424(19)	4639(12)	19(6)
H(33C)	8170(20)	1610(20)	5189(12)	18(6)
H(34A)	7110(20)	9200(20)	7094(13)	30(7)
H(34B)	5610(30)	8780(30)	6803(17)	72(11)
H(34C)	6430(30)	9670(30)	6293(15)	49(9)
H(35A)	8640(20)	8960(20)	5866(12)	26(6)

H(35B)	9160(30)	7440(20)	5991(14)	46(8)
H(35C)	9080(30)	8310(20)	6630(14)	37(8)
H(37)	5240(20)	3130(20)	8349(13)	37(7)
H(38)	5550(30)	960(20)	8924(15)	51(9)
H(39)	7370(20)	-410(20)	8711(12)	30(7)
H(40)	8970(20)	200(20)	7961(11)	17(6)
H(41)	8730(20)	2270(20)	7444(12)	20(6)

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**Crystal Structure Analysis of:**  
**Complex 6 (IAT21)**

Contents

Table 1. Crystal data

Figures Minimum overlap, orientation A, B and A&B

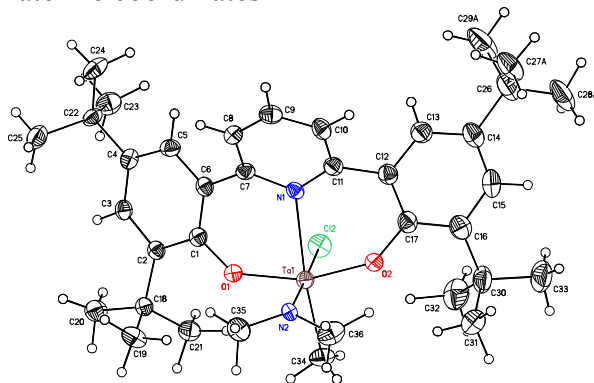
Table 2. Atomic Coordinates

Table 3. Selected bond distances and angles

Table 4. Full bond distances and angles

Table 5. Anisotropic displacement parameters

Table 6. Hydrogen atomic coordinates

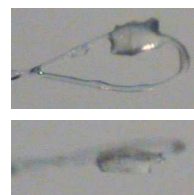


IAT21

**Note:** The crystallographic data have been deposited in the Cambridge Database (CCDC) and has been placed on hold pending further instructions from me. The deposition number is 698486. Ideally the CCDC would like the publication to contain a footnote of the type: "Crystallographic data have been deposited at the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 698486."

**Table 1. Crystal data and structure refinement for IAT21 (CCDC 698486).**

Empirical formula	$C_{36}H_{52}N_2O_2ClTa \cdot C_4H_{10}O$
Formula weight	835.32
Crystallization Solvent	Diethylether
Crystal Habit	Flake
Crystal size	0.22 x 0.13 x 0.04 mm <sup>3</sup>
Crystal color	Colorless



### Data Collection

Type of diffractometer	Bruker KAPPA APEX II	
Wavelength	0.71073 Å MoK $\alpha$	
Data Collection Temperature	100(2) K	
$\theta$ range for 9924 reflections used in lattice determination	2.32 to 27.32°	
Unit cell dimensions	a = 47.432(2) Å b = 13.7447(7) Å c = 12.4876(6) Å	$\beta = 100.510(3)^\circ$
Volume	8004.5(7) Å <sup>3</sup>	
Z	8	
Crystal system	Monoclinic	
Space group	C2/c	
Density (calculated)	1.386 Mg/m <sup>3</sup>	
F(000)	3440	
$\theta$ range for data collection	1.54 to 27.85°	
Completeness to $\theta = 27.85^\circ$	90.1 %	
Index ranges	$-60 \leq h \leq 60, -17 \leq k \leq 17, -16 \leq l \leq 16$	
Data collection scan type	$\omega$ scans; 11 settings	
Reflections collected	63460	
Independent reflections	8588 [ $R_{int} = 0.0730$ ]	
Absorption coefficient	2.850 mm <sup>-1</sup>	
Absorption correction	None	
Max. and min. transmission	0.8945 and 0.5729	

**Table 1 (cont.)****Structure solution and Refinement**

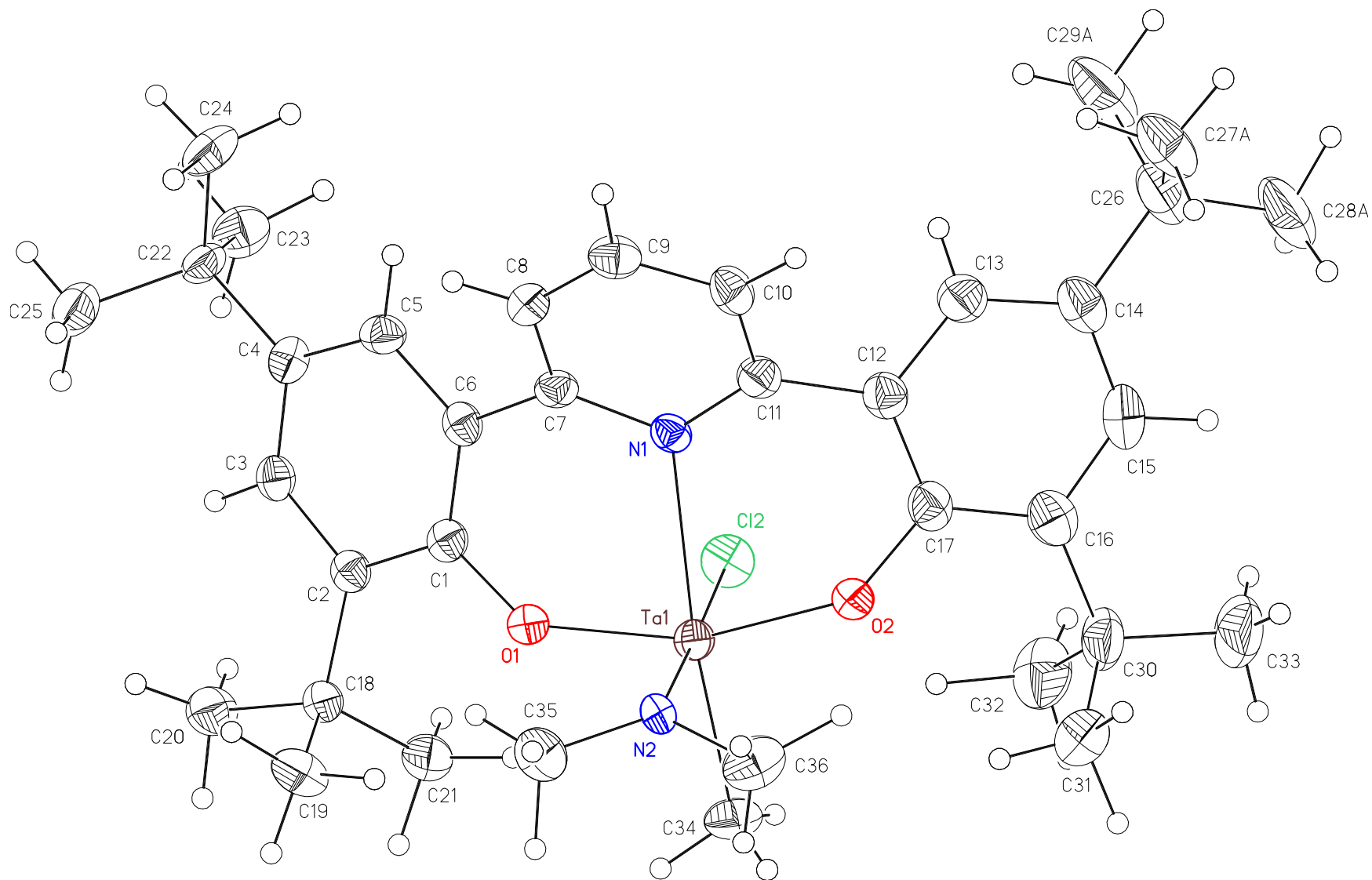
Structure solution program	SHELXS-97 (Sheldrick, 2008)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 2008)
Refinement method	Full matrix least-squares on $F^2$
Data / restraints / parameters	8588 / 0 / 632
Treatment of hydrogen atoms	Mixed
Goodness-of-fit on $F^2$	1.678
Final R indices [ $I > 2\sigma(I)$ , 6077 reflections]	$R1 = 0.0358$ , $wR2 = 0.0563$
R indices (all data)	$R1 = 0.0623$ , $wR2 = 0.0577$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.002
Average shift/error	0.000
Largest diff. peak and hole	2.305 and -1.993 e.Å <sup>-3</sup>

**Special Refinement Details**

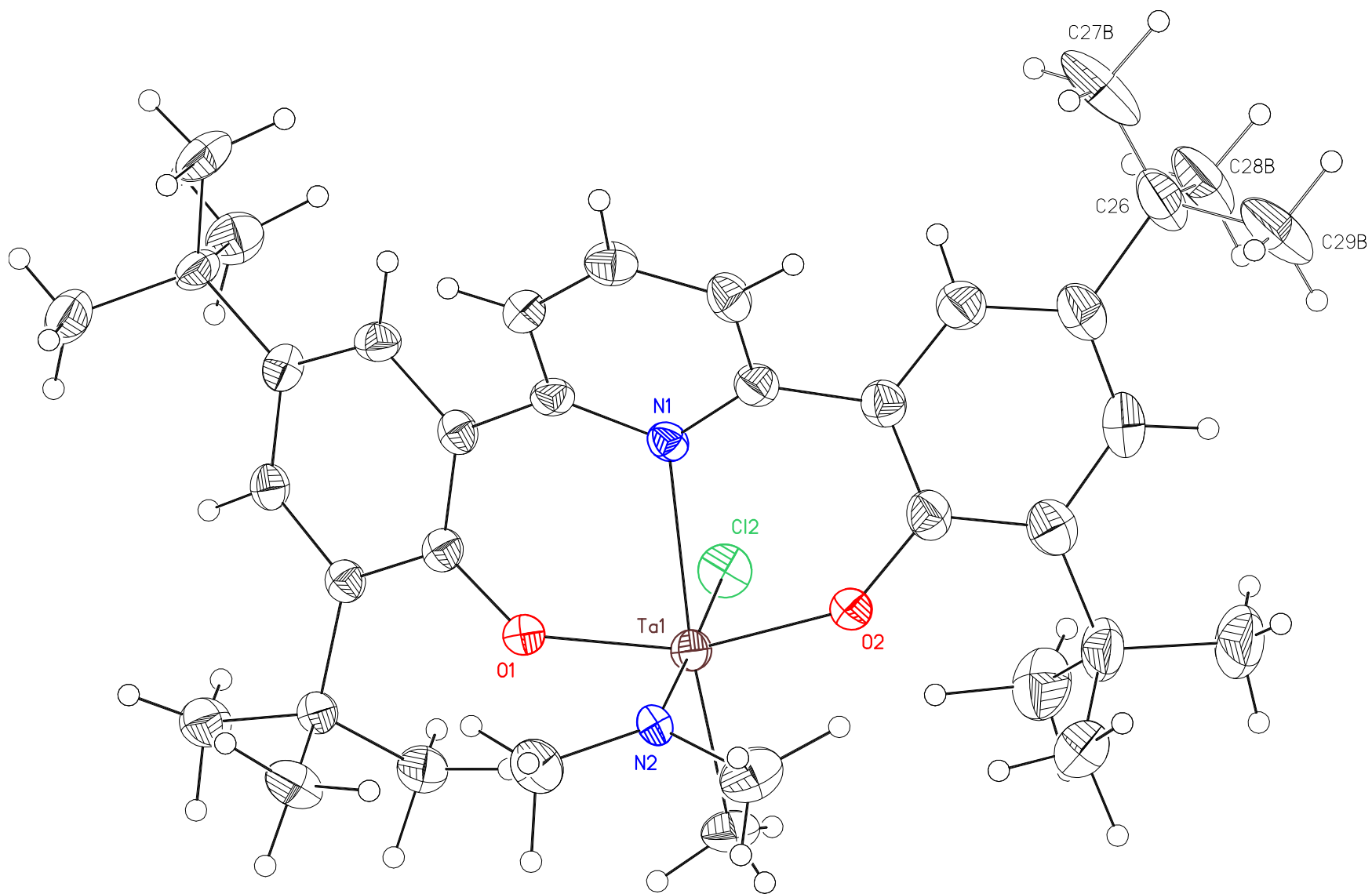
Crystals were mounted on a loop using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

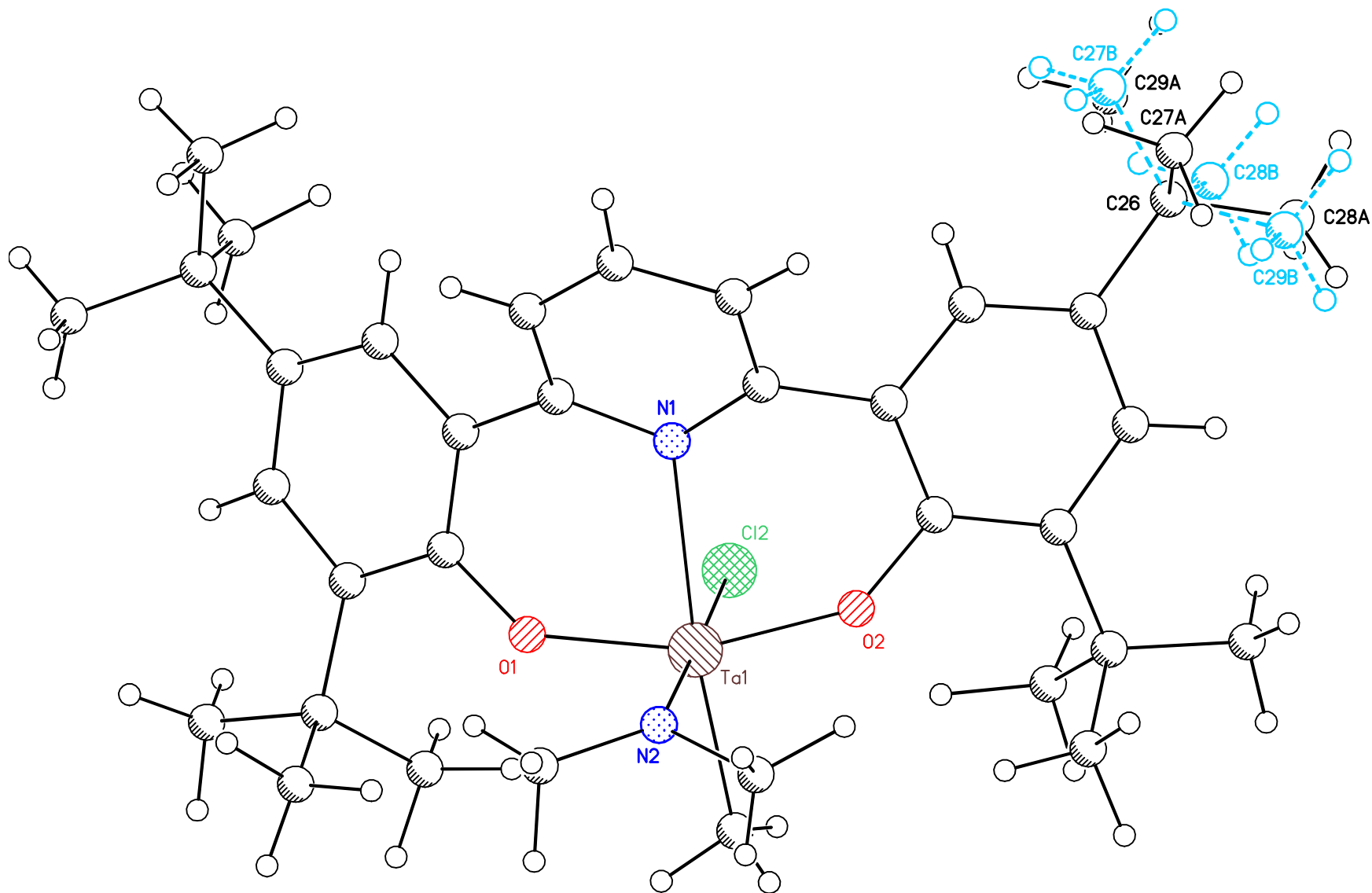
Refinement of  $F^2$  against ALL reflections. The weighted R-factor ( $wR$ ) and goodness of fit ( $S$ ) are based on  $F^2$ , conventional R-factors ( $R$ ) are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.









**Table 2. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for IAT21 (CCDC 698486).  $U(\text{eq})$  is defined as the trace of the orthogonalized  $U^{\text{ij}}$  tensor.**

	x	y	z	$U_{\text{eq}}$	Occ
Ta(1)	3421(1)	7288(1)	2319(1)	23(1)	1
Cl(2)	3683(1)	8069(1)	1016(1)	34(1)	1
O(1)	3157(1)	8332(2)	2051(2)	25(1)	1
O(2)	3754(1)	6578(2)	2877(2)	26(1)	1
N(1)	3674(1)	8445(2)	3567(3)	20(1)	1
N(2)	3180(1)	6665(2)	3258(3)	25(1)	1
C(1)	3154(1)	9327(3)	2052(3)	22(1)	1
C(2)	2963(1)	9859(3)	1263(3)	22(1)	1
C(3)	2985(1)	10873(3)	1316(3)	23(1)	1
C(4)	3174(1)	11374(3)	2109(3)	25(1)	1
C(5)	3352(1)	10823(3)	2894(3)	25(1)	1
C(6)	3342(1)	9796(3)	2883(3)	22(1)	1
C(7)	3530(1)	9275(3)	3773(3)	22(1)	1
C(8)	3574(1)	9681(3)	4807(4)	26(1)	1
C(9)	3773(1)	9306(3)	5632(4)	29(1)	1
C(10)	3942(1)	8546(3)	5397(4)	29(1)	1
C(11)	3894(1)	8134(3)	4352(3)	25(1)	1
C(12)	4106(1)	7442(3)	4084(3)	28(1)	1
C(13)	4391(1)	7569(3)	4566(4)	32(1)	1
C(14)	4611(1)	6986(3)	4322(4)	36(1)	1
C(15)	4531(1)	6227(4)	3579(4)	40(1)	1
C(16)	4250(1)	6054(3)	3063(4)	35(1)	1
C(17)	4034(1)	6681(3)	3334(3)	29(1)	1
C(18)	2729(1)	9364(3)	419(3)	25(1)	1
C(19)	2533(1)	8798(4)	1039(4)	37(1)	1
C(20)	2545(1)	10098(4)	-309(4)	34(1)	1
C(21)	2859(1)	8671(4)	-311(4)	35(1)	1
C(22)	3186(1)	12489(3)	2091(3)	26(1)	1
C(23)	3353(1)	12792(4)	1205(4)	38(1)	1
C(24)	3339(1)	12892(4)	3184(4)	38(1)	1
C(25)	2885(1)	12902(4)	1850(5)	37(1)	1
C(26)	4923(1)	7184(4)	4839(4)	46(1)	1
C(27A)	4953(1)	7089(6)	6108(6)	49(3)	0.672(9)
C(28A)	5138(2)	6522(7)	4477(9)	70(4)	0.672(9)
C(29A)	5003(2)	8258(7)	4628(8)	68(3)	0.672(9)
C(27B)	4966(3)	8020(14)	5608(16)	72(7)	0.328(9)
C(28B)	5078(3)	7410(13)	3845(14)	59(6)	0.328(9)
C(29B)	5053(3)	6221(13)	5355(16)	65(7)	0.328(9)
C(30)	4174(1)	5220(3)	2233(4)	42(1)	1
C(31)	3949(1)	4541(4)	2562(5)	49(2)	1
C(32)	4061(1)	5651(5)	1093(5)	52(2)	1
C(33)	4440(1)	4600(5)	2144(6)	58(2)	1
C(34)	3246(1)	6252(4)	1034(4)	34(1)	1
C(35)	2889(1)	6953(4)	3364(4)	34(1)	1
C(36)	3247(1)	5684(4)	3725(5)	40(1)	1

O(40)	4328(1)	9291(3)	7784(3)	62(1)	1
C(41)	4515(2)	10506(7)	6860(7)	252(8)	1
C(42)	4370(2)	10313(5)	7781(7)	125(3)	1
C(43)	4208(1)	9036(5)	8700(5)	77(2)	1
C(44)	4145(1)	7982(5)	8659(5)	90(2)	1

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**Table 3. Selected bond lengths [Å] and angles [°] for IAT21 (CCDC 698486).**

Ta(1)-O(2)	1.881(3)	O(2)-Ta(1)-O(1)	160.50(11)
Ta(1)-O(1)	1.894(3)	O(2)-Ta(1)-N(2)	95.39(12)
Ta(1)-N(2)	1.974(3)	O(1)-Ta(1)-N(2)	89.89(12)
Ta(1)-C(34)	2.191(5)	O(2)-Ta(1)-C(34)	96.53(17)
Ta(1)-N(1)	2.392(3)	O(1)-Ta(1)-C(34)	102.38(17)
Ta(1)-Cl(2)	2.4681(11)	N(2)-Ta(1)-C(34)	88.38(18)
		O(2)-Ta(1)-N(1)	79.32(11)
		O(1)-Ta(1)-N(1)	81.28(11)
		N(2)-Ta(1)-N(1)	100.21(12)
		C(34)-Ta(1)-N(1)	170.73(18)
		O(2)-Ta(1)-Cl(2)	88.75(8)
		O(1)-Ta(1)-Cl(2)	87.17(8)
		N(2)-Ta(1)-Cl(2)	174.89(10)
		C(34)-Ta(1)-Cl(2)	88.17(16)
		N(1)-Ta(1)-Cl(2)	83.48(8)

**Table 4. Bond lengths [Å] and angles [°] for IAT21 (CCDC 698486).**

Ta(1)-O(2)	1.881(3)	C(21)-H(21B)	1.04(4)
Ta(1)-O(1)	1.894(3)	C(21)-H(21C)	0.92(4)
Ta(1)-N(2)	1.974(3)	C(22)-C(25)	1.518(6)
Ta(1)-C(34)	2.191(5)	C(22)-C(24)	1.527(6)
Ta(1)-N(1)	2.392(3)	C(22)-C(23)	1.529(6)
Ta(1)-Cl(2)	2.4681(11)	C(23)-H(23A)	1.01(4)
O(1)-C(1)	1.368(4)	C(23)-H(23B)	1.03(4)
O(2)-C(17)	1.354(5)	C(23)-H(23C)	1.01(4)
N(1)-C(11)	1.363(5)	C(24)-H(24A)	0.92(4)
N(1)-C(7)	1.377(5)	C(24)-H(24B)	0.98(4)
N(2)-C(35)	1.463(5)	C(24)-H(24C)	1.11(4)
N(2)-C(36)	1.480(6)	C(25)-H(25A)	0.89(4)
C(1)-C(6)	1.397(5)	C(25)-H(25B)	0.98(4)
C(1)-C(2)	1.414(5)	C(25)-H(25C)	1.00(4)
C(2)-C(3)	1.398(5)	C(26)-C(27B)	1.487(19)
C(2)-C(18)	1.543(5)	C(26)-C(28A)	1.496(9)
C(3)-C(4)	1.391(5)	C(26)-C(29B)	1.549(17)
C(3)-H(3)	0.95(4)	C(26)-C(29A)	1.558(9)
C(4)-C(5)	1.394(5)	C(26)-C(27A)	1.570(9)
C(4)-C(22)	1.534(5)	C(26)-C(28B)	1.586(16)
C(5)-C(6)	1.412(5)	C(27A)-H(27A)	0.9800
C(5)-H(5)	0.94(3)	C(27A)-H(27B)	0.9800
C(6)-C(7)	1.478(5)	C(27A)-H(27C)	0.9800
C(7)-C(8)	1.387(5)	C(28A)-H(28A)	0.9800
C(8)-C(9)	1.366(6)	C(28A)-H(28B)	0.9800
C(8)-H(8)	0.89(3)	C(28A)-H(28C)	0.9800
C(9)-C(10)	1.379(6)	C(29A)-H(29A)	0.9800
C(9)-H(9)	1.02(4)	C(29A)-H(29B)	0.9800
C(10)-C(11)	1.402(6)	C(29A)-H(29C)	0.9800
C(10)-H(10)	0.90(3)	C(27B)-H(27D)	0.9800
C(11)-C(12)	1.467(5)	C(27B)-H(27E)	0.9800
C(12)-C(13)	1.389(5)	C(27B)-H(27F)	0.9800
C(12)-C(17)	1.403(5)	C(28B)-H(28D)	0.9800
C(13)-C(14)	1.394(6)	C(28B)-H(28E)	0.9800
C(13)-H(13)	0.84(3)	C(28B)-H(28F)	0.9800
C(14)-C(15)	1.402(6)	C(29B)-H(29D)	0.9800
C(14)-C(26)	1.526(6)	C(29B)-H(29E)	0.9800
C(15)-C(16)	1.389(6)	C(29B)-H(29F)	0.9800
C(15)-H(15)	0.97(4)	C(30)-C(31)	1.530(7)
C(16)-C(17)	1.425(6)	C(30)-C(32)	1.545(7)
C(16)-C(30)	1.543(6)	C(30)-C(33)	1.544(7)
C(18)-C(20)	1.522(6)	C(31)-H(31A)	1.01(5)
C(18)-C(21)	1.526(6)	C(31)-H(31B)	1.09(5)
C(18)-C(19)	1.529(6)	C(31)-H(31C)	0.97(5)
C(19)-H(19A)	1.03(4)	C(32)-H(32A)	1.19(5)
C(19)-H(19B)	1.05(4)	C(32)-H(32B)	1.05(5)
C(19)-H(19C)	0.98(4)	C(32)-H(32C)	0.96(5)
C(20)-H(20A)	1.01(4)	C(33)-H(33A)	1.02(5)
C(20)-H(20B)	0.90(4)	C(33)-H(33B)	0.98(5)
C(20)-H(20C)	1.03(4)	C(33)-H(33C)	1.02(5)
C(21)-H(21A)	1.06(4)	C(34)-H(34A)	0.87(5)

C(34)-H(34B)	1.03(5)	C(2)-C(3)-H(3)	114(2)
C(34)-H(34C)	0.95(6)	C(3)-C(4)-C(5)	117.4(4)
C(35)-H(35A)	1.07(5)	C(3)-C(4)-C(22)	120.4(4)
C(35)-H(35B)	0.98(4)	C(5)-C(4)-C(22)	122.2(4)
C(35)-H(35C)	1.02(4)	C(4)-C(5)-C(6)	121.5(4)
C(36)-H(36A)	0.95(4)	C(4)-C(5)-H(5)	119(2)
C(36)-H(36B)	1.12(4)	C(6)-C(5)-H(5)	120(2)
C(36)-H(36C)	0.90(4)	C(1)-C(6)-C(5)	118.9(4)
O(40)-C(43)	1.413(7)	C(1)-C(6)-C(7)	123.5(4)
O(40)-C(42)	1.418(6)	C(5)-C(6)-C(7)	117.6(4)
C(41)-C(42)	1.467(11)	N(1)-C(7)-C(8)	120.8(4)
C(41)-H(41A)	0.9800	N(1)-C(7)-C(6)	120.8(4)
C(41)-H(41B)	0.9800	C(8)-C(7)-C(6)	118.3(4)
C(41)-H(41C)	0.9800	C(9)-C(8)-C(7)	121.1(4)
C(42)-H(42A)	0.9900	C(9)-C(8)-H(8)	122(2)
C(42)-H(42B)	0.9900	C(7)-C(8)-H(8)	117(2)
C(43)-C(44)	1.478(7)	C(8)-C(9)-C(10)	118.2(4)
C(43)-H(43A)	0.9900	C(8)-C(9)-H(9)	127(2)
C(43)-H(43B)	0.9900	C(10)-C(9)-H(9)	115(2)
C(44)-H(44A)	0.9800	C(9)-C(10)-C(11)	120.2(4)
C(44)-H(44B)	0.9800	C(9)-C(10)-H(10)	124(2)
C(44)-H(44C)	0.9800	C(11)-C(10)-H(10)	116(2)
		N(1)-C(11)-C(10)	121.1(4)
O(2)-Ta(1)-O(1)	160.50(11)	N(1)-C(11)-C(12)	120.4(4)
O(2)-Ta(1)-N(2)	95.39(12)	C(10)-C(11)-C(12)	118.2(4)
O(1)-Ta(1)-N(2)	89.89(12)	C(13)-C(12)-C(17)	118.7(4)
O(2)-Ta(1)-C(34)	96.53(17)	C(13)-C(12)-C(11)	118.3(4)
O(1)-Ta(1)-C(34)	102.38(17)	C(17)-C(12)-C(11)	122.9(4)
N(2)-Ta(1)-C(34)	88.38(18)	C(12)-C(13)-C(14)	122.9(4)
O(2)-Ta(1)-N(1)	79.32(11)	C(12)-C(13)-H(13)	123(3)
O(1)-Ta(1)-N(1)	81.28(11)	C(14)-C(13)-H(13)	114(3)
N(2)-Ta(1)-N(1)	100.21(12)	C(13)-C(14)-C(15)	116.5(4)
C(34)-Ta(1)-N(1)	170.73(18)	C(13)-C(14)-C(26)	120.8(4)
O(2)-Ta(1)-Cl(2)	88.75(8)	C(15)-C(14)-C(26)	122.7(4)
O(1)-Ta(1)-Cl(2)	87.17(8)	C(16)-C(15)-C(14)	123.9(4)
N(2)-Ta(1)-Cl(2)	174.89(10)	C(16)-C(15)-H(15)	120(3)
C(34)-Ta(1)-Cl(2)	88.17(16)	C(14)-C(15)-H(15)	116(3)
N(1)-Ta(1)-Cl(2)	83.48(8)	C(15)-C(16)-C(17)	117.0(4)
C(1)-O(1)-Ta(1)	139.8(2)	C(15)-C(16)-C(30)	121.7(4)
C(17)-O(2)-Ta(1)	142.8(3)	C(17)-C(16)-C(30)	121.3(4)
C(11)-N(1)-C(7)	117.7(3)	O(2)-C(17)-C(12)	117.2(4)
C(11)-N(1)-Ta(1)	119.2(3)	O(2)-C(17)-C(16)	121.9(4)
C(7)-N(1)-Ta(1)	117.8(2)	C(12)-C(17)-C(16)	120.8(4)
C(35)-N(2)-C(36)	110.4(4)	C(20)-C(18)-C(21)	107.9(4)
C(35)-N(2)-Ta(1)	126.3(3)	C(20)-C(18)-C(19)	107.7(4)
C(36)-N(2)-Ta(1)	121.5(3)	C(21)-C(18)-C(19)	109.6(4)
O(1)-C(1)-C(6)	117.3(3)	C(20)-C(18)-C(2)	112.3(3)
O(1)-C(1)-C(2)	121.4(3)	C(21)-C(18)-C(2)	111.4(4)
C(6)-C(1)-C(2)	121.3(4)	C(19)-C(18)-C(2)	107.9(4)
C(3)-C(2)-C(1)	116.7(4)	C(18)-C(19)-H(19A)	109(2)
C(3)-C(2)-C(18)	120.6(3)	C(18)-C(19)-H(19B)	114(3)
C(1)-C(2)-C(18)	122.5(4)	H(19A)-C(19)-H(19B)	108(3)
C(4)-C(3)-C(2)	124.1(4)	C(18)-C(19)-H(19C)	111(3)
C(4)-C(3)-H(3)	121(2)	H(19A)-C(19)-H(19C)	106(3)

H(19B)-C(19)-H(19C)	109(3)	C(14)-C(26)-C(28B)	104.8(6)
C(18)-C(20)-H(20A)	108(2)	C(29B)-C(26)-C(28B)	106.7(10)
C(18)-C(20)-H(20B)	106(2)	C(29A)-C(26)-C(28B)	61.3(7)
H(20A)-C(20)-H(20B)	108(3)	C(27A)-C(26)-C(28B)	146.8(7)
C(18)-C(20)-H(20C)	112(2)	C(26)-C(27A)-H(27A)	109.5
H(20A)-C(20)-H(20C)	109(3)	C(26)-C(27A)-H(27B)	109.5
H(20B)-C(20)-H(20C)	114(4)	C(26)-C(27A)-H(27C)	109.5
C(18)-C(21)-H(21A)	106(2)	C(26)-C(28A)-H(28A)	109.5
C(18)-C(21)-H(21B)	121(2)	C(26)-C(28A)-H(28B)	109.5
H(21A)-C(21)-H(21B)	112(3)	C(26)-C(28A)-H(28C)	109.5
C(18)-C(21)-H(21C)	111(3)	C(26)-C(29A)-H(29A)	109.5
H(21A)-C(21)-H(21C)	106(4)	C(26)-C(29A)-H(29B)	109.5
H(21B)-C(21)-H(21C)	100(4)	C(26)-C(29A)-H(29C)	109.5
C(25)-C(22)-C(24)	108.6(4)	C(26)-C(27B)-H(27D)	109.5
C(25)-C(22)-C(23)	110.4(4)	C(26)-C(27B)-H(27E)	109.5
C(24)-C(22)-C(23)	108.9(4)	H(27D)-C(27B)-H(27E)	109.5
C(25)-C(22)-C(4)	109.8(4)	C(26)-C(27B)-H(27F)	109.5
C(24)-C(22)-C(4)	111.2(3)	H(27D)-C(27B)-H(27F)	109.5
C(23)-C(22)-C(4)	108.0(3)	H(27E)-C(27B)-H(27F)	109.5
C(22)-C(23)-H(23A)	103(2)	C(26)-C(28B)-H(28D)	109.5
C(22)-C(23)-H(23B)	110(2)	C(26)-C(28B)-H(28E)	109.5
H(23A)-C(23)-H(23B)	126(3)	H(28D)-C(28B)-H(28E)	109.5
C(22)-C(23)-H(23C)	112(2)	C(26)-C(28B)-H(28F)	109.5
H(23A)-C(23)-H(23C)	105(3)	H(28D)-C(28B)-H(28F)	109.5
H(23B)-C(23)-H(23C)	102(3)	H(28E)-C(28B)-H(28F)	109.5
C(22)-C(24)-H(24A)	109(2)	C(26)-C(29B)-H(29D)	109.5
C(22)-C(24)-H(24B)	112(2)	C(26)-C(29B)-H(29E)	109.5
H(24A)-C(24)-H(24B)	108(3)	H(29D)-C(29B)-H(29E)	109.5
C(22)-C(24)-H(24C)	104(2)	C(26)-C(29B)-H(29F)	109.5
H(24A)-C(24)-H(24C)	106(3)	H(29D)-C(29B)-H(29F)	109.5
H(24B)-C(24)-H(24C)	118(3)	H(29E)-C(29B)-H(29F)	109.5
C(22)-C(25)-H(25A)	109(3)	C(31)-C(30)-C(16)	111.2(4)
C(22)-C(25)-H(25B)	108(2)	C(31)-C(30)-C(32)	109.9(5)
H(25A)-C(25)-H(25B)	111(4)	C(16)-C(30)-C(32)	109.5(4)
C(22)-C(25)-H(25C)	115(2)	C(31)-C(30)-C(33)	107.6(5)
H(25A)-C(25)-H(25C)	107(4)	C(16)-C(30)-C(33)	111.4(4)
H(25B)-C(25)-H(25C)	107(3)	C(32)-C(30)-C(33)	107.2(5)
C(27B)-C(26)-C(28A)	130.2(8)	C(30)-C(31)-H(31A)	108(3)
C(27B)-C(26)-C(14)	114.6(7)	C(30)-C(31)-H(31B)	112(3)
C(28A)-C(26)-C(14)	115.1(5)	H(31A)-C(31)-H(31B)	110(4)
C(27B)-C(26)-C(29B)	113.3(10)	C(30)-C(31)-H(31C)	117(3)
C(28A)-C(26)-C(29B)	50.6(7)	H(31A)-C(31)-H(31C)	110(4)
C(14)-C(26)-C(29B)	107.8(7)	H(31B)-C(31)-H(31C)	101(4)
C(27B)-C(26)-C(29A)	50.8(8)	C(30)-C(32)-H(32A)	112(2)
C(28A)-C(26)-C(29A)	109.0(6)	C(30)-C(32)-H(32B)	112(3)
C(14)-C(26)-C(29A)	110.0(4)	H(32A)-C(32)-H(32B)	102(4)
C(29B)-C(26)-C(29A)	142.1(7)	C(30)-C(32)-H(32C)	109(3)
C(27B)-C(26)-C(27A)	55.7(8)	H(32A)-C(32)-H(32C)	115(4)
C(28A)-C(26)-C(27A)	108.3(6)	H(32B)-C(32)-H(32C)	106(4)
C(14)-C(26)-C(27A)	108.4(4)	C(30)-C(33)-H(33A)	106(3)
C(29B)-C(26)-C(27A)	63.5(8)	C(30)-C(33)-H(33B)	103(3)
C(29A)-C(26)-C(27A)	105.6(6)	H(33A)-C(33)-H(33B)	104(5)
C(27B)-C(26)-C(28B)	109.0(10)	C(30)-C(33)-H(33C)	108(3)
C(28A)-C(26)-C(28B)	56.2(7)	H(33A)-C(33)-H(33C)	122(4)



H(33B)-C(33)-H(33C)	113(4)	H(41A)-C(41)-H(41B)	109.5
Ta(1)-C(34)-H(34A)	111(3)	C(42)-C(41)-H(41C)	109.5
Ta(1)-C(34)-H(34B)	116(3)	H(41A)-C(41)-H(41C)	109.5
H(34A)-C(34)-H(34B)	103(4)	H(41B)-C(41)-H(41C)	109.5
Ta(1)-C(34)-H(34C)	113(4)	O(40)-C(42)-C(41)	105.5(7)
H(34A)-C(34)-H(34C)	114(5)	O(40)-C(42)-H(42A)	110.6
H(34B)-C(34)-H(34C)	100(4)	C(41)-C(42)-H(42A)	110.6
N(2)-C(35)-H(35A)	108(2)	O(40)-C(42)-H(42B)	110.6
N(2)-C(35)-H(35B)	112(2)	C(41)-C(42)-H(42B)	110.6
H(35A)-C(35)-H(35B)	108(3)	H(42A)-C(42)-H(42B)	108.8
N(2)-C(35)-H(35C)	113(2)	O(40)-C(43)-C(44)	108.9(5)
H(35A)-C(35)-H(35C)	102(3)	O(40)-C(43)-H(43A)	109.9
H(35B)-C(35)-H(35C)	114(3)	C(44)-C(43)-H(43A)	109.9
N(2)-C(36)-H(36A)	112(2)	O(40)-C(43)-H(43B)	109.9
N(2)-C(36)-H(36B)	112.6(18)	C(44)-C(43)-H(43B)	109.9
H(36A)-C(36)-H(36B)	99(3)	H(43A)-C(43)-H(43B)	108.3
N(2)-C(36)-H(36C)	105(3)	C(43)-C(44)-H(44A)	109.5
H(36A)-C(36)-H(36C)	102(4)	C(43)-C(44)-H(44B)	109.5
H(36B)-C(36)-H(36C)	126(4)	H(44A)-C(44)-H(44B)	109.5
C(43)-O(40)-C(42)	109.0(5)	C(43)-C(44)-H(44C)	109.5
C(42)-C(41)-H(41A)	109.5	H(44A)-C(44)-H(44C)	109.5
C(42)-C(41)-H(41B)	109.5	H(44B)-C(44)-H(44C)	109.5

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**Table 5. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^4$ ) for IAT21 (CCDC 698486). The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$**

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Ta(1)	229(1)	205(1)	235(1)	10(1)	7(1)	9(1)
Cl(2)	361(7)	364(7)	303(6)	27(5)	79(5)	-9(5)
O(1)	227(17)	218(18)	296(17)	-24(13)	16(13)	-3(13)
O(2)	234(17)	231(17)	286(17)	-14(13)	4(14)	7(13)
N(1)	178(19)	210(20)	220(19)	26(15)	33(16)	-22(15)
N(2)	153(19)	240(20)	360(20)	-41(17)	21(16)	18(15)
C(1)	220(20)	240(30)	220(20)	-9(19)	86(19)	1(19)
C(2)	210(20)	250(30)	200(20)	-8(19)	59(19)	9(19)
C(3)	230(30)	230(30)	230(20)	40(20)	40(20)	30(20)
C(4)	290(30)	230(30)	230(20)	5(19)	60(20)	40(20)
C(5)	250(30)	270(30)	240(20)	-50(20)	50(20)	-40(20)
C(6)	200(20)	270(30)	210(20)	-1(19)	68(19)	15(19)
C(7)	220(20)	190(30)	250(20)	15(19)	50(20)	-46(19)
C(8)	260(30)	220(30)	290(30)	-20(20)	40(20)	10(20)
C(9)	320(30)	300(30)	240(30)	10(20)	20(20)	-60(20)
C(10)	250(30)	320(30)	270(30)	60(20)	0(20)	30(20)
C(11)	250(20)	240(30)	270(20)	30(20)	30(20)	-10(20)
C(12)	250(20)	280(30)	300(20)	30(20)	62(19)	0(20)
C(13)	310(30)	300(40)	350(30)	30(20)	40(20)	-10(20)
C(14)	270(30)	410(30)	390(30)	110(20)	30(20)	30(20)
C(15)	340(30)	410(30)	440(30)	70(30)	100(30)	150(30)
C(16)	330(30)	390(30)	310(30)	60(20)	50(20)	30(20)
C(17)	280(30)	270(30)	320(30)	80(20)	40(20)	20(20)
C(18)	230(20)	230(30)	280(20)	0(20)	-10(20)	20(20)
C(19)	280(30)	440(40)	380(30)	20(30)	0(30)	-70(30)
C(20)	360(30)	320(30)	320(30)	30(30)	-30(30)	0(30)
C(21)	320(30)	370(30)	330(30)	-60(30)	-10(30)	10(30)
C(22)	330(20)	150(30)	270(20)	-5(17)	3(19)	0(19)
C(23)	450(30)	310(30)	380(30)	40(30)	80(30)	-40(30)
C(24)	560(40)	180(40)	380(30)	-20(20)	80(30)	-10(30)
C(25)	430(30)	250(40)	440(30)	-10(30)	90(30)	50(30)
C(26)	230(30)	540(40)	590(30)	110(30)	0(20)	40(30)
C(27A)	290(40)	660(80)	480(50)	210(40)	-30(40)	0(40)
C(28A)	210(50)	950(90)	940(90)	-100(70)	60(50)	80(50)
C(29A)	250(50)	840(80)	890(90)	260(60)	-10(50)	-90(50)
C(27B)	200(90)	1200(200)	700(150)	200(140)	-10(90)	-130(100)
C(28B)	210(80)	720(160)	800(130)	330(110)	10(80)	-40(90)
C(29B)	280(100)	910(160)	670(150)	180(120)	-130(100)	-90(100)
C(30)	380(30)	450(30)	440(30)	-70(30)	90(30)	160(30)
C(31)	490(40)	380(40)	580(40)	-90(30)	50(30)	40(30)
C(32)	620(40)	610(40)	360(30)	-60(30)	130(30)	150(40)
C(33)	520(40)	600(50)	650(50)	-140(40)	160(40)	210(40)
C(34)	400(30)	320(30)	290(30)	-70(20)	10(30)	-90(30)
C(35)	310(30)	400(30)	330(30)	50(20)	90(20)	-20(20)
C(36)	560(40)	310(30)	360(30)	70(30)	110(30)	-60(30)

O(40)	790(30)	540(30)	440(20)	-16(19)	-100(20)	-140(20)
C(41)	3280(150)	3250(140)	780(70)	490(80)	-270(80)	-2870(130)
C(42)	1420(80)	620(50)	1390(80)	-40(50)	-600(60)	-310(50)
C(43)	620(40)	1130(60)	510(40)	-290(40)	-70(30)	250(40)
C(44)	660(40)	1410(70)	610(40)	220(40)	10(30)	-220(40)

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**Table 6. Hydrogen coordinates (  $\times 10^4$ ) and isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for IAT21 (CCDC 698486).**

	x	y	z	$U_{\text{iso}}$
H(3)	2862(8)	11210(20)	740(30)	23(11)
H(5)	3480(7)	11150(20)	3440(30)	10(10)
H(8)	3471(8)	10200(20)	4900(30)	19(12)
H(9)	3822(9)	9560(30)	6410(40)	56(15)
H(10)	4091(7)	8300(20)	5870(30)	14(10)
H(13)	4450(8)	8050(30)	4960(30)	24(13)
H(15)	4686(10)	5820(30)	3420(40)	53(15)
H(19A)	2453(9)	9270(30)	1560(40)	47(14)
H(19B)	2635(10)	8220(30)	1510(40)	64(16)
H(19C)	2365(10)	8540(30)	550(30)	48(14)
H(20A)	2405(8)	9720(20)	-870(30)	22(11)
H(20B)	2664(8)	10420(30)	-660(30)	22(13)
H(20C)	2431(9)	10540(30)	120(40)	50(15)
H(21A)	2684(9)	8390(30)	-880(30)	41(13)
H(21B)	3000(9)	8130(30)	20(30)	50(15)
H(21C)	2973(9)	9000(30)	-710(30)	41(15)
H(23A)	3220(9)	12590(30)	520(30)	44(13)
H(23B)	3565(9)	12590(30)	1430(30)	45(13)
H(23C)	3370(8)	13520(30)	1140(30)	33(12)
H(24A)	3338(8)	13560(30)	3150(30)	29(13)
H(24B)	3540(8)	12680(30)	3360(30)	31(12)
H(24C)	3194(9)	12700(30)	3760(30)	46(12)
H(25A)	2895(9)	13550(30)	1860(40)	49(16)
H(25B)	2781(8)	12670(30)	2410(30)	42(13)
H(25C)	2769(9)	12710(30)	1130(40)	48(13)
H(27A)	5155	7157	6450	73
H(27B)	4883	6450	6288	73
H(27C)	4841	7601	6379	73
H(28A)	5332	6711	4836	106
H(28B)	5123	6572	3686	106
H(28C)	5101	5850	4671	106
H(29A)	4874	8697	4926	101
H(29B)	4983	8367	3842	101
H(29C)	5201	8384	4983	101
H(27D)	4858	7908	6194	109
H(27E)	4897	8618	5217	109
H(27F)	5170	8085	5918	109
H(28D)	4998	8006	3479	88
H(28E)	5049	6867	3329	88
H(28F)	5284	7498	4112	88
H(29D)	5256	6320	5664	97
H(29E)	5035	5714	4794	97
H(29F)	4951	6020	5932	97
H(31A)	4030(10)	4280(30)	3310(40)	64(17)
H(31B)	3895(11)	3940(40)	1990(40)	88(19)
H(31C)	3761(11)	4820(30)	2560(40)	68(18)
H(32A)	3850(10)	6130(30)	1080(40)	69(16)
H(32B)	3989(11)	5110(40)	520(40)	83(19)

H(32C)	4215(10)	5980(30)	840(40)	66(18)
H(33A)	4571(12)	5040(40)	1790(50)	90(20)
H(33B)	4366(11)	4130(40)	1570(40)	80(20)
H(33C)	4510(11)	4280(30)	2880(40)	67(18)
H(34A)	3354(10)	6220(30)	540(40)	60(19)
H(34B)	3049(11)	6420(30)	590(40)	73(18)
H(34C)	3207(14)	5640(40)	1310(50)	130(30)
H(35A)	2742(10)	6500(30)	2830(40)	64(16)
H(35B)	2850(8)	7630(30)	3150(30)	45(14)
H(35C)	2842(9)	6780(30)	4100(40)	50(14)
H(36A)	3215(8)	5640(20)	4450(30)	25(12)
H(36B)	3482(8)	5520(20)	3870(30)	29(12)
H(36C)	3108(10)	5300(30)	3370(40)	61(18)
H(41A)	4677	10058	6890	378
H(41B)	4585	11178	6901	378
H(41C)	4379	10411	6176	378
H(42A)	4490	10528	8473	150
H(42B)	4184	10658	7685	150
H(43A)	4029	9408	8697	93
H(43B)	4344	9194	9377	93
H(44A)	4012	7829	7982	136
H(44B)	4058	7800	9284	136
H(44C)	4324	7617	8684	136

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## Crystal Structure Analysis of:

### Complex 7 (IAT06)

#### Contents

Table 1. Crystal data

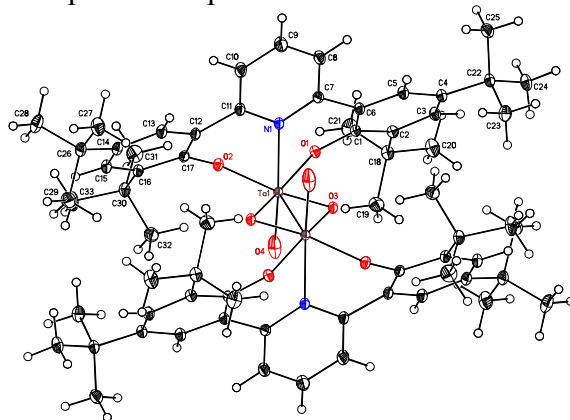
Figures Minimum overlap (see Special Refinement Details)

Table 2. Atomic Coordinates

Table 3. Selected bond distances and angles

Table 4. Full bond distances and angles

Table 5. Anisotropic displacement parameters



IAT06

**Note:** The crystallographic data have been deposited in the Cambridge Database (CCDC) and has been placed on hold pending further instructions from me. The deposition number is 683409. Ideally the CCDC would like the publication to contain a footnote of the type: "Crystallographic data have been deposited at the CCDC, 12 Union Road, Cambridge CB2 1EZ, UK and copies can be obtained on request, free of charge, by quoting the publication citation and the deposition number 683409."

**Table 1. Crystal data and structure refinement for IAT06 (CCDC 683409).**

Empirical formula	$C_{66}H_{86}N_2O_8Ta_2 \cdot 4(CH_2Cl_2)$
Formula weight	1736.97
Crystallization Solvent	Petroleum ether
Crystal Habit	Block
Crystal size	0.25 x 0.23 x 0.14 mm <sup>3</sup>
Crystal color	Colorless

### Data Collection

Type of diffractometer	Bruker KAPPA APEX II	
Wavelength	0.71073 Å MoK $\alpha$	
Data Collection Temperature	100(2) K	
$\theta$ range for 9853 reflections used in lattice determination	2.87 to 51.25°	
Unit cell dimensions	a = 16.0570(7) Å b = 12.8407(6) Å c = 18.4561(9) Å	$\beta$ = 103.542(3)°
Volume	3699.5(3) Å <sup>3</sup>	
Z	2	
Crystal system	Monoclinic	
Space group	P2 <sub>1</sub> /c	
Density (calculated)	1.559 Mg/m <sup>3</sup>	
F(000)	1748	
Data collection program	Bruker APEX2 v2.1-0	
$\theta$ range for data collection	1.95 to 51.56°	
Completeness to $\theta$ = 51.56°	98.9 %	
Index ranges	-35 ≤ h ≤ 28, -27 ≤ k ≤ 28, -40 ≤ l ≤ 40	
Data collection scan type	$\omega$ scans; 31 settings	
Data reduction program	Bruker SAINT-Plus v7.34A	
Reflections collected	331499	
Independent reflections	41036 [ $R_{int}$ = 0.0418]	
Absorption coefficient	3.297 mm <sup>-1</sup>	
Absorption correction	Semi-empirical from equivalents	
Max. and min. transmission	0.7501 and 0.6005	

**Table 1 (cont.)****Structure solution and Refinement**

Structure solution program	SHELXS-97 (Sheldrick, 2008)
Primary solution method	Direct methods
Secondary solution method	Difference Fourier map
Hydrogen placement	Geometric positions
Structure refinement program	SHELXL-97 (Sheldrick, 2008)
Refinement method	Full matrix least-squares on $F^2$
Data / restraints / parameters	41036 / 0 / 418
Treatment of hydrogen atoms	Riding
Goodness-of-fit on $F^2$	1.629
Final R indices [ $I > 2\sigma(I)$ , 31962 reflections]	$R_1 = 0.0250$ , $wR_2 = 0.0388$
R indices (all data)	$R_1 = 0.0440$ , $wR_2 = 0.0400$
Type of weighting scheme used	Sigma
Weighting scheme used	$w = 1/\sigma^2(F_o^2)$
Max shift/error	0.009
Average shift/error	0.000
Largest diff. peak and hole	3.987 and -2.005 e.Å <sup>-3</sup>

**Special Refinement Details**

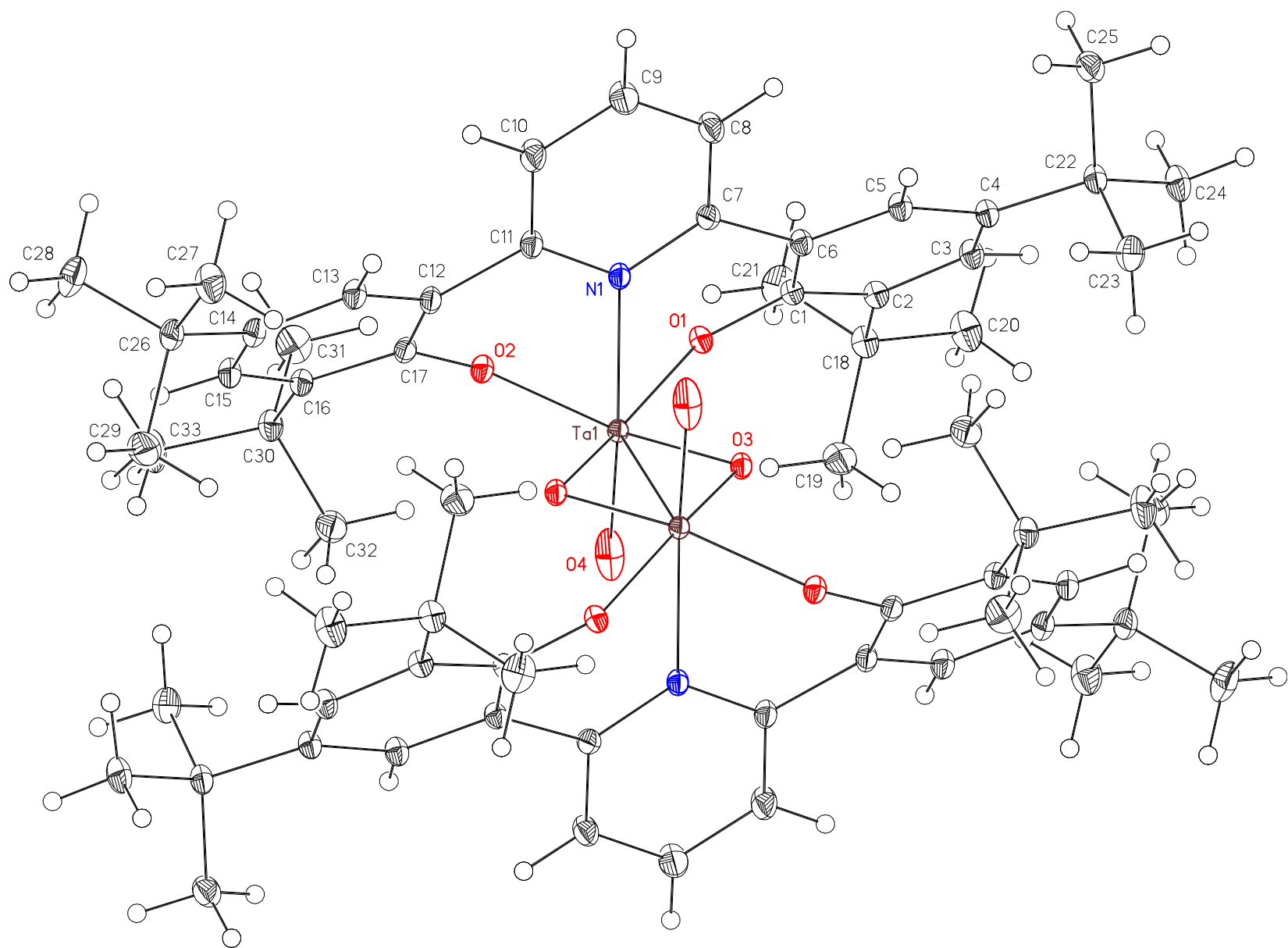
Crystals were mounted on a glass fiber using Paratone oil then placed on the diffractometer under a nitrogen stream at 100K.

The identity of the terminal ligand (O4) on Ta is ambiguous. Based on bond distance it appears to be oxygen and therefore an OH group. The location of an hydrogen is notoriously difficult in third row transition metals and it is not included in the model as its position would be arbitrary. One large peak in the difference map is  $\sim 1.4\text{\AA}$  from O4 and is believed to arise from incomplete absorption correction.

Refinement of  $F^2$  against ALL reflections. The weighted R-factor ( $wR$ ) and goodness of fit ( $S$ ) are based on  $F^2$ , conventional R-factors ( $R$ ) are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculating R-factors(gt) etc. and is not relevant to the choice of reflections for refinement. R-factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and R-factors based on ALL data will be even larger.

All esds (except the esd in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell esds are taken into account individually in the estimation of esds in distances, angles and torsion angles; correlations between esds in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell esds is used for estimating esds involving l.s. planes.





**Table 2. Atomic coordinates ( $\times 10^4$ ) and equivalent isotropic displacement parameters ( $\text{\AA}^2 \times 10^3$ ) for IAT06 (CCDC 683409).  $U(\text{eq})$  is defined as the trace of the orthogonalized  $U^{\text{ij}}$  tensor.**

	x	y	z	$U_{\text{eq}}$
Ta(1)	5209(1)	9163(1)	4507(1)	9(1)
O(1)	4520(1)	7954(1)	4103(1)	11(1)
O(2)	6325(1)	8491(1)	4592(1)	11(1)
O(3)	4236(1)	9768(1)	4801(1)	11(1)
O(4)	5149(1)	9889(1)	3492(1)	26(1)
N(1)	5332(1)	8109(1)	5542(1)	10(1)
C(1)	3814(1)	7596(1)	4307(1)	10(1)
C(2)	3083(1)	7318(1)	3756(1)	11(1)
C(3)	2347(1)	7060(1)	3998(1)	12(1)
C(4)	2304(1)	7075(1)	4742(1)	11(1)
C(5)	3057(1)	7294(1)	5273(1)	12(1)
C(6)	3826(1)	7524(1)	5069(1)	10(1)
C(7)	4640(1)	7565(1)	5653(1)	11(1)
C(8)	4698(1)	6990(1)	6300(1)	16(1)
C(9)	5462(1)	6958(1)	6839(1)	20(1)
C(10)	6168(1)	7472(1)	6706(1)	18(1)
C(11)	6098(1)	8041(1)	6052(1)	12(1)
C(12)	6883(1)	8521(1)	5902(1)	11(1)
C(13)	7558(1)	8799(1)	6510(1)	13(1)
C(14)	8347(1)	9116(1)	6405(1)	13(1)
C(15)	8464(1)	9104(1)	5677(1)	13(1)
C(16)	7822(1)	8854(1)	5052(1)	12(1)
C(17)	7006(1)	8615(1)	5179(1)	11(1)
C(18)	3082(1)	7324(1)	2928(1)	13(1)
C(19)	3236(1)	8435(1)	2677(1)	18(1)
C(20)	2221(1)	6952(1)	2444(1)	19(1)
C(21)	3776(1)	6587(1)	2776(1)	20(1)
C(22)	1462(1)	6883(1)	4978(1)	12(1)
C(23)	1196(1)	7899(1)	5299(1)	17(1)
C(24)	732(1)	6560(1)	4320(1)	17(1)
C(25)	1585(1)	6023(1)	5570(1)	18(1)
C(26)	9074(1)	9498(1)	7046(1)	14(1)
C(27)	8846(1)	9413(1)	7801(1)	21(1)
C(28)	9890(1)	8852(1)	7080(1)	22(1)
C(29)	9256(1)	10642(1)	6902(1)	21(1)
C(30)	7983(1)	8835(1)	4266(1)	14(1)
C(31)	7781(1)	7749(1)	3923(1)	21(1)
C(32)	7426(1)	9650(1)	3763(1)	19(1)
C(33)	8924(1)	9073(1)	4272(1)	22(1)
C(41)	1709(1)	3688(1)	3999(1)	24(1)
Cl(1)	1726(1)	4014(1)	3077(1)	42(1)
Cl(2)	659(1)	3602(1)	4124(1)	29(1)
C(42)	4285(1)	4102(1)	5696(1)	44(1)
Cl(3)	3799(1)	4440(1)	6420(1)	53(1)
Cl(4)	3795(1)	4688(1)	4839(1)	48(1)

**Table 3. Selected bond lengths [Å] and angles [°] for IAT06 (CCDC 683409).**

Ta(1)-O(3)	1.9328(5)	O(3)-Ta(1)-O(3)#1	78.95(2)
Ta(1)-O(3)#1	1.9431(5)	O(3)-Ta(1)-O(1)	90.25(2)
Ta(1)-O(1)	1.9502(5)	O(3)#1-Ta(1)-O(1)	160.96(2)
Ta(1)-O(2)	1.9623(5)	O(3)-Ta(1)-O(2)	159.60(2)
Ta(1)-O(4)	2.0733(8)	O(3)#1-Ta(1)-O(2)	89.29(2)
Ta(1)-N(1)	2.3119(6)	O(1)-Ta(1)-O(2)	96.07(2)
Ta(1)-Ta(1)#1	2.99175(11)	O(3)-Ta(1)-O(4)	101.76(2)
O(3)-Ta(1)#1	1.9431(5)	O(3)#1-Ta(1)-O(4)	101.13(2)
		O(1)-Ta(1)-O(4)	96.37(2)
		O(2)-Ta(1)-O(4)	96.80(2)
		O(3)-Ta(1)-N(1)	85.50(2)
		O(3)#1-Ta(1)-N(1)	86.16(2)
		O(1)-Ta(1)-N(1)	77.32(2)
		O(2)-Ta(1)-N(1)	77.05(2)
		O(4)-Ta(1)-N(1)	170.51(2)
		O(3)-Ta(1)-Ta(1)#1	39.603(14)
		O(3)#1-Ta(1)-Ta(1)#1	39.348(14)
		O(1)-Ta(1)-Ta(1)#1	128.080(15)
		O(2)-Ta(1)-Ta(1)#1	126.687(15)
		O(4)-Ta(1)-Ta(1)#1	104.896(19)
		N(1)-Ta(1)-Ta(1)#1	84.590(15)

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y+2,-z+1

**Table 4. Bond lengths [Å] and angles [°] for IAT06 (CCDC 683409).**

Ta(1)-O(3)	1.9328(5)	O(3)-Ta(1)-O(3)#1	78.95(2)
Ta(1)-O(3)#1	1.9431(5)	O(3)-Ta(1)-O(1)	90.25(2)
Ta(1)-O(1)	1.9502(5)	O(3)#1-Ta(1)-O(1)	160.96(2)
Ta(1)-O(2)	1.9623(5)	O(3)-Ta(1)-O(2)	159.60(2)
Ta(1)-O(4)	2.0733(8)	O(3)#1-Ta(1)-O(2)	89.29(2)
Ta(1)-N(1)	2.3119(6)	O(1)-Ta(1)-O(2)	96.07(2)
Ta(1)-Ta(1)#1	2.99175(11)	O(3)-Ta(1)-O(4)	101.76(2)
O(1)-C(1)	1.3552(8)	O(3)#1-Ta(1)-O(4)	101.13(2)
O(2)-C(17)	1.3569(8)	O(1)-Ta(1)-O(4)	96.37(2)
O(3)-Ta(1)#1	1.9431(5)	O(2)-Ta(1)-O(4)	96.80(2)
N(1)-C(11)	1.3652(8)	O(3)-Ta(1)-N(1)	85.50(2)
N(1)-C(7)	1.3677(9)	O(3)#1-Ta(1)-N(1)	86.16(2)
C(1)-C(6)	1.4050(10)	O(1)-Ta(1)-N(1)	77.32(2)
C(1)-C(2)	1.4074(9)	O(2)-Ta(1)-N(1)	77.05(2)
C(2)-C(3)	1.3970(10)	O(4)-Ta(1)-N(1)	170.51(2)
C(2)-C(18)	1.5283(10)	O(3)-Ta(1)-Ta(1)#1	39.603(14)
C(3)-C(4)	1.3926(10)	O(3)#1-Ta(1)-Ta(1)#1	39.348(14)
C(4)-C(5)	1.3948(9)	O(1)-Ta(1)-Ta(1)#1	128.080(15)
C(4)-C(22)	1.5336(10)	O(2)-Ta(1)-Ta(1)#1	126.687(15)
C(5)-C(6)	1.4032(9)	O(4)-Ta(1)-Ta(1)#1	104.896(19)
C(6)-C(7)	1.4882(8)	N(1)-Ta(1)-Ta(1)#1	84.590(15)
C(7)-C(8)	1.3890(10)	C(1)-O(1)-Ta(1)	126.89(4)
C(8)-C(9)	1.3867(10)	C(17)-O(2)-Ta(1)	124.29(4)
C(9)-C(10)	1.3829(11)	Ta(1)-O(3)-Ta(1)#1	101.05(2)
C(10)-C(11)	1.3926(10)	C(11)-N(1)-C(7)	119.74(6)
C(11)-C(12)	1.4873(9)	C(11)-N(1)-Ta(1)	119.75(4)
C(12)-C(17)	1.3978(10)	C(7)-N(1)-Ta(1)	120.50(4)
C(12)-C(13)	1.4108(9)	O(1)-C(1)-C(6)	118.95(5)
C(13)-C(14)	1.3864(10)	O(1)-C(1)-C(2)	119.73(6)
C(14)-C(15)	1.4004(11)	C(6)-C(1)-C(2)	121.31(6)
C(14)-C(26)	1.5346(9)	C(3)-C(2)-C(1)	117.03(6)
C(15)-C(16)	1.3936(10)	C(3)-C(2)-C(18)	121.20(6)
C(16)-C(17)	1.4174(9)	C(1)-C(2)-C(18)	121.74(6)
C(16)-C(30)	1.5307(11)	C(4)-C(3)-C(2)	123.56(6)
C(18)-C(20)	1.5366(10)	C(3)-C(4)-C(5)	117.41(6)
C(18)-C(19)	1.5364(11)	C(3)-C(4)-C(22)	121.91(6)
C(18)-C(21)	1.5375(11)	C(5)-C(4)-C(22)	120.67(6)
C(22)-C(25)	1.5331(11)	C(4)-C(5)-C(6)	121.78(7)
C(22)-C(24)	1.5346(10)	C(5)-C(6)-C(1)	118.43(6)
C(22)-C(23)	1.5352(11)	C(5)-C(6)-C(7)	119.43(6)
C(26)-C(27)	1.5262(12)	C(1)-C(6)-C(7)	121.76(6)
C(26)-C(29)	1.5339(11)	N(1)-C(7)-C(8)	120.48(6)
C(26)-C(28)	1.5405(10)	N(1)-C(7)-C(6)	121.06(6)
C(30)-C(31)	1.5341(11)	C(8)-C(7)-C(6)	118.37(6)
C(30)-C(33)	1.5377(10)	C(9)-C(8)-C(7)	120.12(7)
C(30)-C(32)	1.5402(11)	C(10)-C(9)-C(8)	118.94(7)
C(41)-Cl(1)	1.7584(11)	C(9)-C(10)-C(11)	119.99(6)
C(41)-Cl(2)	1.7586(9)	N(1)-C(11)-C(10)	120.59(6)
C(42)-Cl(3)	1.7518(16)	N(1)-C(11)-C(12)	120.74(6)
C(42)-Cl(4)	1.7622(15)	C(10)-C(11)-C(12)	118.58(6)
		C(17)-C(12)-C(13)	118.88(6)

C(17)-C(12)-C(11)	121.96(6)	C(25)-C(22)-C(24)	108.47(6)
C(13)-C(12)-C(11)	118.89(7)	C(4)-C(22)-C(24)	112.28(6)
C(14)-C(13)-C(12)	121.32(7)	C(25)-C(22)-C(23)	109.79(7)
C(13)-C(14)-C(15)	117.62(6)	C(4)-C(22)-C(23)	108.44(6)
C(13)-C(14)-C(26)	122.78(7)	C(24)-C(22)-C(23)	107.71(6)
C(15)-C(14)-C(26)	119.57(6)	C(27)-C(26)-C(29)	108.80(7)
C(16)-C(15)-C(14)	123.84(6)	C(27)-C(26)-C(14)	112.32(6)
C(15)-C(16)-C(17)	116.48(7)	C(29)-C(26)-C(14)	108.35(6)
C(15)-C(16)-C(30)	122.05(6)	C(27)-C(26)-C(28)	107.95(7)
C(17)-C(16)-C(30)	121.47(6)	C(29)-C(26)-C(28)	109.12(7)
O(2)-C(17)-C(12)	118.91(6)	C(14)-C(26)-C(28)	110.26(6)
O(2)-C(17)-C(16)	119.76(6)	C(16)-C(30)-C(31)	109.82(7)
C(12)-C(17)-C(16)	121.32(6)	C(16)-C(30)-C(33)	111.97(6)
C(2)-C(18)-C(20)	111.64(6)	C(31)-C(30)-C(33)	107.25(6)
C(2)-C(18)-C(19)	110.11(6)	C(16)-C(30)-C(32)	110.92(6)
C(20)-C(18)-C(19)	107.45(6)	C(31)-C(30)-C(32)	109.36(7)
C(2)-C(18)-C(21)	110.39(6)	C(33)-C(30)-C(32)	107.42(7)
C(20)-C(18)-C(21)	107.11(6)	Cl(1)-C(41)-Cl(2)	111.97(5)
C(19)-C(18)-C(21)	110.05(7)	Cl(3)-C(42)-Cl(4)	113.18(6)
C(25)-C(22)-C(4)	110.10(6)		

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Symmetry transformations used to generate equivalent atoms:

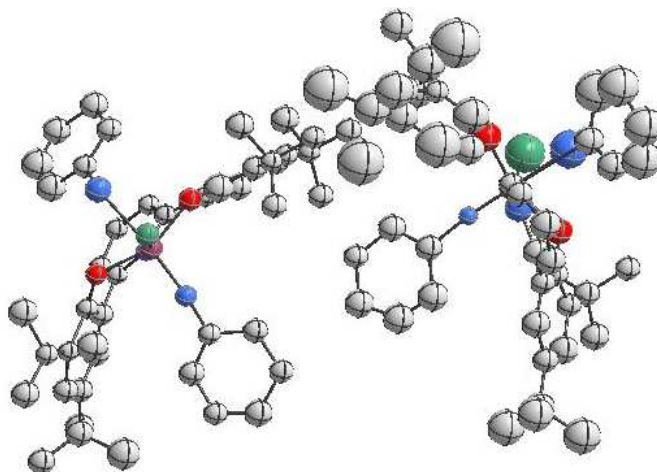
#1 -x+1,-y+2,-z+1

**Table 5. Anisotropic displacement parameters ( $\text{\AA}^2 \times 10^4$ ) for IAT06 (CCDC 683409). The anisotropic displacement factor exponent takes the form:  $-2\pi^2 [h^2 a^{*2} U^{11} + \dots + 2 h k a^* b^* U^{12}]$**

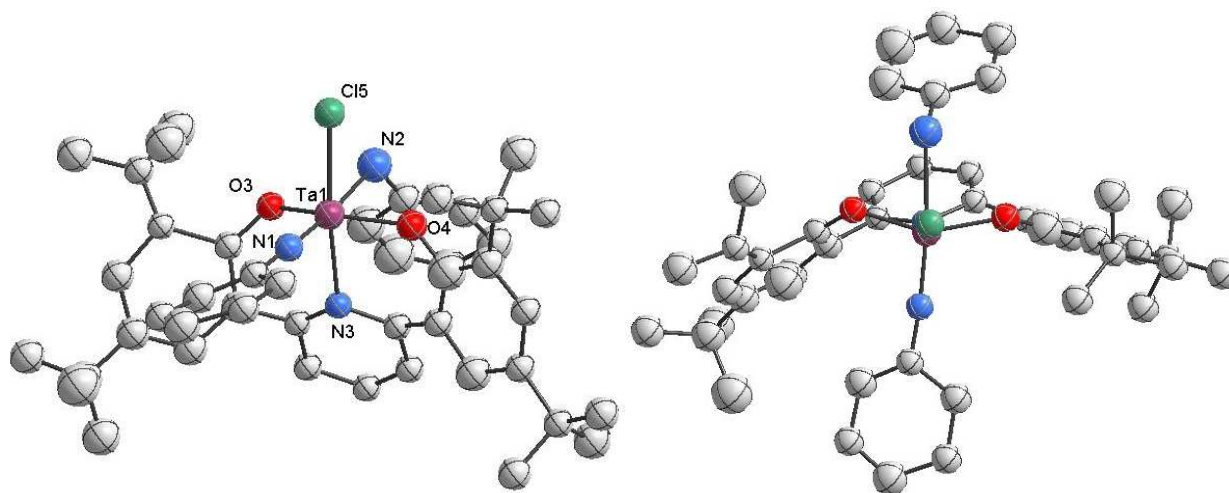
	$U^{11}$	$U^{22}$	$U^{33}$	$U^{23}$	$U^{13}$	$U^{12}$
Ta(1)	72(1)	98(1)	92(1)	11(1)	14(1)	-1(1)
O(1)	95(2)	129(2)	103(2)	-2(2)	30(2)	-16(1)
O(2)	86(2)	138(2)	101(2)	0(2)	12(2)	3(1)
O(3)	76(2)	112(2)	124(2)	-4(2)	16(2)	-4(1)
O(4)	145(2)	358(4)	291(4)	-165(3)	66(2)	-41(2)
N(1)	77(2)	122(2)	104(2)	3(2)	16(2)	-4(2)
C(1)	95(2)	110(2)	98(2)	-5(2)	27(2)	-8(2)
C(2)	109(2)	127(2)	94(2)	-10(2)	23(2)	-10(2)
C(3)	102(2)	153(3)	105(3)	-12(2)	14(2)	-20(2)
C(4)	94(2)	135(2)	112(3)	-4(2)	23(2)	-14(2)
C(5)	99(2)	154(3)	98(2)	-2(2)	22(2)	-15(2)
C(6)	90(2)	122(2)	96(2)	-3(2)	16(2)	-12(2)
C(7)	97(2)	122(2)	113(2)	14(2)	6(2)	-8(2)
C(8)	115(2)	218(3)	135(3)	51(2)	21(2)	-27(2)
C(9)	145(3)	293(4)	150(3)	95(3)	10(3)	-25(3)
C(10)	110(2)	264(3)	144(3)	70(3)	-4(2)	-14(2)
C(11)	85(2)	151(3)	109(3)	8(2)	14(2)	0(2)
C(12)	76(2)	137(2)	120(3)	7(2)	16(2)	4(2)
C(13)	97(2)	163(3)	117(3)	1(2)	14(2)	0(2)
C(14)	89(2)	146(2)	134(3)	-3(2)	8(2)	5(2)
C(15)	88(2)	159(3)	148(3)	-1(2)	23(2)	-3(2)
C(16)	92(2)	126(2)	134(3)	0(2)	34(2)	7(2)
C(17)	84(2)	113(2)	118(3)	0(2)	17(2)	9(2)
C(18)	129(2)	175(3)	96(3)	-23(2)	29(2)	-17(2)
C(19)	192(3)	218(3)	120(3)	29(2)	26(2)	-19(2)
C(20)	172(3)	286(4)	114(3)	-46(3)	15(2)	-58(3)
C(21)	202(3)	244(4)	165(3)	-58(3)	62(3)	27(3)
C(22)	91(2)	164(3)	114(3)	3(2)	22(2)	-16(2)
C(23)	140(3)	209(3)	163(3)	-26(2)	42(2)	3(2)
C(24)	112(2)	230(3)	151(3)	-20(2)	17(2)	-42(2)
C(25)	142(3)	220(3)	179(3)	56(2)	38(3)	-24(2)
C(26)	97(2)	170(3)	145(3)	-6(2)	-3(2)	-7(2)
C(27)	158(3)	303(4)	143(3)	-11(3)	-1(3)	-37(3)
C(28)	125(3)	261(4)	236(4)	-34(3)	-20(3)	43(2)
C(29)	190(3)	184(3)	225(4)	-16(3)	8(3)	-43(2)
C(30)	108(2)	174(3)	142(3)	-6(2)	48(2)	1(2)
C(31)	235(3)	206(3)	225(4)	-67(3)	99(3)	0(3)
C(32)	172(3)	228(3)	172(3)	57(3)	55(3)	2(2)
C(33)	122(3)	345(4)	198(4)	-13(3)	77(3)	-20(3)
C(41)	176(3)	275(4)	267(4)	26(3)	66(3)	52(3)
Cl(1)	401(1)	500(2)	413(2)	251(1)	216(1)	170(1)
Cl(2)	224(1)	244(1)	433(2)	12(1)	160(1)	8(1)
C(42)	340(5)	329(6)	641(9)	15(6)	79(6)	148(4)
Cl(3)	558(2)	248(1)	861(3)	-85(2)	338(2)	-111(1)
Cl(4)	275(1)	391(2)	690(2)	32(2)	-73(1)	1(1)



## Structure Drawings of Complex 5



The crystal is twinned; one of the two molecules in the asymmetric unit (right) prevents proper refinement of the system.



Despite the lack of refinement, the  $C_s$ -symmetry of the *bis*(phenolate)pyridine ligand is apparent.